## ALKALOIDS OF Dipsacus azureus

T. U. Rakhmatullaev and S. Yu. Yunusov

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We have investigated the alkaloids of the seeds of Dipsacus azureus Schrenk. (family Dipsacaceae) collected in the Chimgan mountains (Tashkent oblast, September $1 \overline{12,1968)}$ in the stage of the withering of the epigeal part. The air-dry comminuted seeds ( 1 kg ) were defatted with petroleum ether. The petroleum ether extractive substances amounted to 41.5 g . The seeds were dried in the air until the petroleum ether had volatilized completely. Then they were moistened with ammonia and extracted with chloroform. This gave $5.1 \mathrm{~g}(0.51 \%$ of the weight of the seeds) of total bases. This material was separated according to the basic strengths of its components with sodium phosphate buffer.

When the fraction with pH 2 was separated chromatographically, the ethereal-chloroformic eluate deposited crystals ( 0.3 g ) with the composition $\mathrm{C}_{11} \mathrm{H}_{13} \mathrm{O}_{3} \mathrm{~N}, \mathrm{mp} 130-131^{\circ} \mathrm{C}$ (ether), mol. wt. 207 (mass-spectrometrically), $\mathrm{R}_{f} 0.85$ [in the butanol-acetic acid-water ( $20: 1: 20$ ) system].

The UV spectrum exhibited one maximum at $273 \mathrm{~nm}(\log \varepsilon 3.25)$. The IR spectrum had characteristic absorption bands at $3255 \mathrm{~cm}^{-1}(-\mathrm{OH}), 1735 \mathrm{~cm}^{-1}$ (ester group), and $1595 \mathrm{~cm}^{-1}$ (pyridine).

Because of the strong absorption of the pyridine chromophore, the optical rotatory dispersion (ORD) could be measured as far as 300 nm . A flat negative ORD curve corresponded to this region.

The mass spectrum of the base had peaks of ions with $\mathrm{m} / \mathrm{e} 207,179,175,160,147,118,91,77$, and 65. The NMR spectrum of the base showed two one-proton singlets at $\tau 1.22$ and 1.69 ppm corresponding to two hydrogen atoms present in the $\alpha$ and $\alpha^{\prime}$ positions to the nitrogen atom of the pyridine ring. The nonequivalence of these protons can be explained by the influence on them of the carbonyl of a $\mathrm{COOCH}_{3}$ group present in the $\beta$ position to the nitrogen atom of the ring. A three-proton doublet at $8.68 \mathrm{ppm}(\mathrm{J}=7$ Hz ) and a one-proton multiplet at $\tau 6.85 \mathrm{ppm}$ correspond to a $>\mathrm{CH}-\mathrm{CH}_{3}$ grouping. The appearance of a one-proton multiplet in the weak-field region is due to the propinquity of the pyridine nucleus. A two-proton doublet with its center at $\tau 6.65 \mathrm{ppm}$ relates to the $\mathrm{H}_{\mathrm{c}}$ protons. The signal from the $\mathrm{H}_{\mathrm{b}}$ proton appears at $\tau 5.46 \mathrm{ppm}$ and that from the hydroxy group at 6.14 ppm . The three protons of the methyl of the ester group give a sharp singlet at $\tau 6.16 \mathrm{ppm}$. The spectrum was recorded on a $\mathrm{JNM}-4 \mathrm{H}-100 / 100 \mathrm{MHz}$ instrument in $\mathrm{CDCl}_{3}$ with HMDS as internal standard.

In its physical constants and elementary composition, the base that we have isolated is similar to the known alkaloid cantleyine [1, 2].

## LITERATURE CITED

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