

¹⁶O₃ Coriolis Coupling Constants for the (021) and (120) Vibrational States

Natural isotopic abundance: 0.992728.

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|--------------------|---|
| Reference | [95Bar1] |
| Method | Fourier transform spectroscopy. |
| Equations | Equations 6 and 18 in chapter “Introduction”. |
| Statistical errors | One standard deviation. |
| Remarks | All values are given in cm ⁻¹ . Calculated constants are purposely given with two supplementary digits in order to reproduce the energy levels to experimental accuracy. Molecular constants determined in the same fit are given in chapter “ ¹⁶ O ₃ Vibrational Energy, Rotational and Centrifugal Distortion Constants for the (021) and (120) States”. Constant without SE, fixed in the fit has been proposed in chapter “ ¹⁶ O ₃ Coriolis Coupling Constants for the (100) and (001) Vibrational States”. The isotopic composition of the elements used for the calculation of the natural isotopic abundance is taken from [2007Coh]. |
| Abbreviation | SE: Statistical error. |

| Coefficient | Coupling constant | SE | |
|------------------------|-------------------|---------|------------------|
| $h_{(021,120)}^C$ | -0.1007004 | 0.00061 | $\times 10^{-1}$ |
| $h_{(021,120)}^{'C}$ | -0.470 | | |
| $h_{(021,120)}^{''C}$ | 0.12226 | 0.039 | $\times 10^{-5}$ |
| $h_{(021,120)}^{'''C}$ | -0.8046 | 0.12 | $\times 10^{-7}$ |

Symbols and abbreviations

| Short form | Full form |
|--|---|
| $h_{(021,120)}^C, h_{(021,120)}^{'C}, h_{(021,120)}^{''C}, h_{(021,120)}^{'''C}$ | Coriolis coupling constant for the (021) and (120) states |
| SE | Statistical error |

References

- [95Bar1] Barbe, A., Sulakshina, O., Plateaux, J.J., Hamdouni, A., and Bouazza, S.: High-Resolution Infrared Spectra of Ozone in the 2300–2600 cm⁻¹ Region. *J. Mol. Spectrosc.* **170** (1995) 244–250.
- [2007Coh] Cohen, E.R., Cvitaš, T., Frey, J.G., Holmström, B., Kuchitsu, K., Marquardt, R., Mills, I., Pavese, F., Quack, M., Stohner, J., Strauss, H.L., Takami, M., Thor, A.J.: Quantities, Units and Symbols in Physical Chemistry. The IUPAC Green Book, 3rd Ed., Cambridge: RSC Publishing, 2007.