Research Article

Effect of lanthanum, neodymium on piezoelectric, dielectric and ferroelectric properties of PZT

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Abstract: The effect of La^{3+} and Nd^{3+} on piezoelectric, dielectric and ferroelectric properties of PZT (Pb(Zr_{0.53}Ti_{0.47})O₃) was studied. The powders were prepared by "wet-chemical" route followed by calcination at 800 °C for 4 h. XRD analysis of the calcined powders confirms the tetragonal phase in the undoped PZT, which gradually decreases with the dopant concentration and the appearance of rhombohedral phase simultaneously. SEM study of sintered pellets reveals the decrease in grain size with the increase in dopant concentration beyond 0.02 mol Nd³⁺ and 0.03 mol La³⁺ respectively. The piezoelectric constant (d_{33}) and dielectric properties were maximum for 0.02 mol Nd³⁺ and 0.03 mol La³⁺ coped sample, and the sample with combined dopants shows intermediate remnant polarization (P_r). Based on the above study, it is concluded that La³⁺ is more effective than Nd³⁺, including in a mixture of La³⁺ and Nd³⁺ dopants.

Keywords: lead zirconate titanate; wet-chemical; piezoelectric; dielectric; ferroelectric

1 Introduction

Lead zirconate titanate (PZT) is considered as one of the most widely studied piezoelectric materials for various applications. Currently, these materials are in the fore front for the application in aerospace vibration control, precision flow control as a replacement of solenoid-based flow control, fuel injection system in automobile industry, energy harvesting from different vibration sources, and so on [1–5]. These applications of PZT materials require the development of materials with very good properties. In the case of PZT, the properties are enhanced or tailor made by the addition

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of different dopants in "A" or "B" sites [6-11]. These dopants are mainly of two types: "donor" dopants, such as La³⁺, Nb⁵⁺, Ce³⁺ and Ta⁵⁺, which produce "soft" PZT; "acceptor" dopants, such as K⁺, Na⁺, Sc³⁺ and Fe^{3+} , which produce "hard" PZT. The soft dopants facilitate the domain wall motion; thereby, the electronic properties are much better compared to undoped PZT. La is one of the most widely studied dopants in PZT system [12-15]. Haertling [12] reported that La³⁺ increases the squareness of the hysteresis loop and decreases the coercive field (E_c) , improves the dielectric constant, electromechanical coupling coefficient, mechanical compliance (S) and enhances the optical transparency. Xiang *et al.* [13] reported that lanthanum substitution increases the level of diffuseness of the phase transition and 7.6 mol%

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lanthanum-modified PZT exhibits relaxor behavior. They also reported the tetragonality of PZT decreases with the increase in La^{3+} concentration. Mohidden *et al.* [16] studied the effect of Nd-doped PZT and reported that the grain size decreases with the increase of Nd concentration and the phase transition is of diffused nature without relaxor behavior. They also reported that the dielectric constant and dissipation factor have shown strong lower frequency dispersion. Thamjaree et al. [17] studied the development of phase formation of Nd on PZT. However, there is hardly any study about the effect of combination of both La and Nd on properties of PZT. Therefore, in this study, an attempt had been made to study the effect of both the dopants on microstructure, piezoelectric, dielectric and ferroelectric properties of PZT.

2 Experiment

2.1 Materials

Analytical grade lead nitrate (99.5%), zirconium oxychloride (99.9%), titanium tetrachloride (99%), neodymium nitrate (99.9%) and lanthanum nitrate (99.9%) were purchased from M/s. Loba Chemie. The above materials were used as starting materials without any further purification.

2.2 Preparation of powders

PZT powders corresponding to compositions $Pb_{1-x}(La_x or Nd_x)(Zr_{0.53}Ti_{0.47})O_3$, where x=0, 0.01, 0.02, 0.03,

0.04, 0.06 or 0.08 mol, were prepared by wet-chemical method using salt solution as precursor [18,19]. To compensate PbO loss during calcination and sintering, 3 wt% excess lead solution was added. The combined salt solution was converted to a mixed hydroxide in the pH range of 8.0-8.5 by adding dilute ammonia solution. The precipitate was filtered and repeatedly washed till it was free of nitrate ions. The precipitate was dried at 110 °C overnight and calcined at 800 °C for 4 h. A typical flow sheet for the preparation of PZT powder by wet-chemical route is presented in Fig. 1. After calcination, the powders were de-agglomerated, granulated, uniaxially pressed into pellets and sintered at 1250 °C for 2 h in a closed lead-rich atmosphere to minimize lead oxide loss. Sintered pellets were leveled, polished and electroded with silver paste. The samples were then poled in a DC field of 2 kV/mm in a silicone oil bath for 30 min, and the piezoelectric, dielectric and ferroelectric properties were measured.

3 Characterization

PZT phase formation was confirmed by X-ray diffraction (XRD) technique (M/s. Phillips, Holand). The morphology of the PZT sintered pellets was ascertained by scanning electron microscope (LEO 440i). The linear piezo-electric strain coefficient (d_{33}) and room temperature dielectric properties were measured using a piezometer (Model PM-35, M/s. Take Control, UK). The P–E hystersis loop (Model HLT-2, M/s. Digital System, Mumbai) was studied by

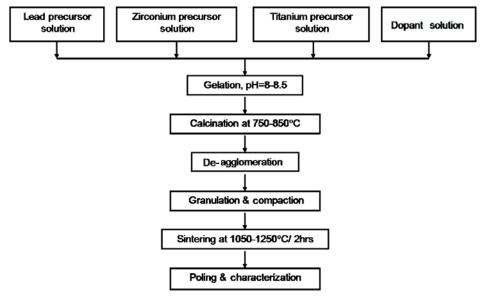


Fig. 1 Typical flow sheet of PZT preparation by wet-chemical route.

applying an AC voltage gradually up to saturation level normally observed around 35–40 kV/cm.

4 Results and discussion

4.1 X-ray diffraction studies

XRD patterns of undoped, La^{3+} and Nd^{3+} -doped PZT samples are presented in Fig. 2. Pure tetragonal phase (characterized by tetragonal splitting) was found in undoped PZT and pure rhombohedral phase in higher dopant concentration (0.06 mol). A mixture of tetragonal and rhombohedral phase was found in the intermediate compositions with tetragonal phase decreasing as the increasing in dopant concentration. The results are similar to that observed by Xiang *et al.* [13] and are attributed to the change in lattice parameters, i.e. the lattice parameters "*c*" decreases while "*a*" increases with lanthanum content; therefore, the tetragonality (*c/a*) of lanthanum-modified PZT

4.2 Microstructural analysis

Typical SEM pictures of chemically etched sintered pellets are presented in Fig. 3. The average grain sizes of undoped (1.18 μ m), 0.02 mol Nd³⁺ (2.13 μ m), 0.06 mol Nd³⁺ (1.16 μ m), 0.03 mol La³⁺ (2.45 μ m) and 0.06 mol La³⁺ (1.00 μ m) were measured. The grain size decreases with the increase in dopant concentration beyond 0.02 mol Nd³⁺ and 0.03 mol La³⁺ respectively.

4.3 Piezoelectric properties

The piezoelectric properties of the samples are

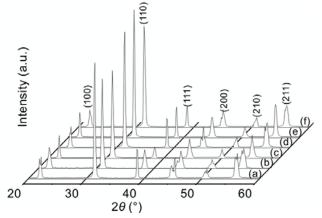


Fig. 2 XRD patterns of (a) undoped, (b) 0.03 mol La^{3+} doped, (c) 0.06 mol La^{3+} doped, (d) 0.02 mol Nd³⁺ doped, (e) 0.06 mol Nd³⁺ doped and (f) 0.01 mol La^{3+} + 0.01 mol Nd³⁺ doped PZT samples.

presented in Fig. 4. The d_{33} of doped samples are much higher than undoped PZT samples due to "soft" nature of the dopants which facilitates the easy reorientation of dipoles in the presence of a DC electric field. In general, La³⁺ and Nd³⁺ ions replace Pb²⁺ ions from "A" site and extra positive charges creates lattice vacancy to maintain electrical neutrality [6]. These Pb vacancies facilitate domain wall movement by application of an electric field [20,21]. It was also observed that d_{33} increases with the increase in Nd³⁺ concentration up to 0.02 mol and then decreases. Similarly, it increases up to 0.03 mol for La³⁺. Higher

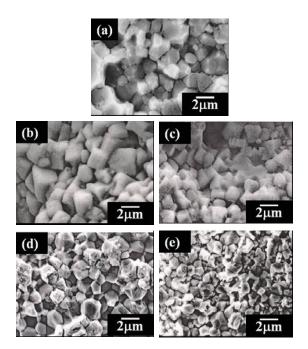


Fig. 3 SEM pictures of chemically etched (a) undoped, (b) 0.02 mol Nd³⁺ doped, (c) 0.06 mol Nd³⁺ doped, (d) 0.03 mol La³⁺ doped and (e) 0.06 mol La³⁺ doped PZT sintered pellets.

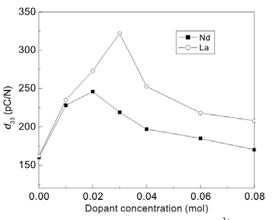


Fig. 4 Variation in d_{33} with increase in Nd³⁺ and La³⁺ concentration.

dopant concentration leads to precipitation of the dopant on grain boundaries, as observed in case of ceria doped PZTs [8]. Due to this, the grain size as well as domain size decreases (also observed by Hoffmann *et al.* [21]); therefore, hinders domain wall orientation and subsequently decreases d_{33} .

4.4 Dielectric properties

The dielectric constant and loss factor of PZT samples are presented in Fig. 5. The dielectric constant was maximum for samples with 0.02 mol of Nd³⁺ and also for samples with 0.03 mol of La³⁺. As per micro structural study, the grain size was maximum at these compositions. This could be based on a similar effect observed by Kong *et al.* [22], i.e., the decrease in grain size increases the volume fractions of grain boundaries. Therefore, decreasing the domain wall mobility thus, decrease the dielectric constant. As expected, the loss factor trend was opposite to that of dielectric constant.

4.5 Ferroelectric properties

P-E hysteresis loops of the samples are presented in Fig. 6. The undoped sample possesses very low remnant and saturation polarization compared to the doped samples. Sample with $La^{3+}(0.03 \text{ mol})$ measured maximum remnant polarization (P_r) of 18.03 μ C/cm². Nd³⁺-doped samples have less remnant and saturation polarization than La^{3+} counterpart. The combined dopants produce intermediate P_r value compared to individual dopants. Due to better uniform microstructure, achievable domain alignment occurs with higher value of saturation and remnant polarization in La³⁺-doped samples than the undoped and Nd^{3+} -doped samples.

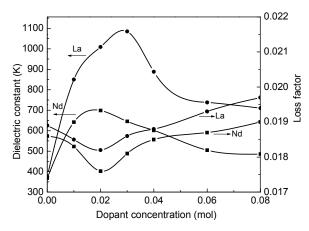


Fig. 5 Variation in dielectric constant and loss factor with increase in Nd^{3+} and La^{3+} concentration.

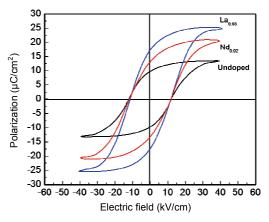


Fig. 6 Hysteresis loops of undoped, 0.02 mol Nd^{3+} doped and 0.03 mol La^{3+} doped PZT samples.

5 Conclusions

PZT samples were prepared by wet-chemical route doped with Nd³⁺, La³⁺ and their combinations. Based on the study of piezoelectric, dielectric and ferroelectric properties, it is concluded that La³⁺ is more effective than Nd³⁺, including in a mixture of La³⁺ and Nd³⁺ dopants.

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