# Four new labdane-type diterpenoid glycosides from *Diplopterygium laevissimum*

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**Abstract:** Four new labdane-type diterpenoid glycosides, laevissiosides A–D (1–4) were isolated from the 95% ethanol extract of *Diplopterygium laevissimum* (Christ) Nakai, along with two known analogues,  $18-\beta$ -D-glucopyranosyl ester-sclareol (5) and 18-hydroxy-sclareol (6). The structures of compounds 1–4 were elucidated by extensive 1D and 2D NMR spectroscopy as well as high-resolution MS analyses. All isolated compounds were evaluated for their cytotoxic effects.

Keywords: Diplopterygium laevissimum, labdane-type diterpenoid glycosides, laevissiosides

# Introduction

*Diplopterygium laevissimum* (Christ) Nakai, belonging to the Gleicheniaceae family, is widely distributed in south China. Its rhizome has been used for treating hemostasis, stomach, and epistaxis as Chinese herbal medicine.<sup>1</sup> Many clerodaneand labdane-type diterpenoid glycosides, which are commonly glycosidated at C-13 in ferns, have been isolated from this family.<sup>2–8</sup> Previous research showed that some clerodane-type diterpenoid glycosides isolated from *Dicranopteris* species could accelerate the growth of the stems of lettuce and inhibit the root growth.<sup>5</sup>

Our previous chemical studies have led to the isolation of two highly oxygenated phenolic derivatives and some clerodane-type diterpenoid glycosides from *Dicranopteris*.<sup>6,9-11</sup> As a systematic research work on the bioactive constituents from the ferns, the whole plant of *D. laevissimum* had been studied, which led to the isolation of four new labdane-type diterpenoid glycosides (1–4), along with two known analogues (5 and 6). All of these showed no *in vitro* cytotoxicity against five human cancer cell lines (HL-60, SMMC-7712, A-549, SK-BR-3 and PANC-1). Herein, the isolation and structure elucidation of compounds 1–4 were described.

### **Results and Discussion**

Compound **1** was obtained as a white amorphous powder. The molecular formula  $C_{35}H_{52}O_{10}$  was established by the HRESIMS (631.3471 [M - H]<sup>-</sup>; calcd. 631.3482),

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corresponding to ten degrees of unsaturation. The IR spectrum showed the presence of hydroxyl (3428 cm<sup>-1</sup>) and carbonyl (1704 cm<sup>-1</sup>) groups. The <sup>1</sup>H and <sup>13</sup>C NMR (Tables 1 and 2) indicated the existence of a set of signals for a hexose [anomeric signals at  $\delta_{\rm H}$  4.30 (d, J = 8.0 Hz);  $\delta_{\rm C}$  104.6] and other 20 carbon resonances, including two olefinic carbons ( $\delta_{\rm C}$ 147.5 and 110.8) and three oxygen-bearing carbons ( $\delta_{\rm C}$  73.4, 73.9 and 79.2). These data were very similar to those of  $18-\beta$ -D-glucopyranosyl ester-sclareol (5), a known compound also isolated from this plant. However, detailed comparison the MS and NMR data of 1 with those of 5 revealed that 1 had one more p-coumaroyl group, which was attached to C-4' of the sugar moiety as concluded from the HMBC (Figure 1) correlations of H-4' ( $\delta_{\rm H}$  4.84) with C-1" ( $\delta_{\rm C}$  167.4). The double bond of the p-coumaroyl group was suggested as trans- due to the coupling constant (J = 15.6 Hz). Acidic hydrolysis of 1 gave D-glucose as sugar residue. The coupling constants of the anomeric proton (J = 8.0 Hz) indicated the  $\beta$  configuration of glucosyl moiety. Assignment of glycosidic protons system was achieved by analysis of <sup>1</sup>H-<sup>1</sup>H COSY and HSQC. The location

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Table 1.	<sup>1</sup> H NMR data	of compounds 1–4	$(\delta \text{ in ppm},$	J in Hz
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no.	<b>1</b> <sup>a</sup>	no.	<b>2</b> <sup>a</sup>	<b>3</b> <sup>b</sup>	<b>4</b> <sup>a,c</sup>
1	1.66 (m); 0.95 (m)	1	1.59 (m); 1.09 (m)	1.59 (m); 1.04 (m)	1.79 (m); 1.13 (m)
2	1.63 (m); 1.42 (m)	2	1.59 (m); 1.41 (m)	1.60 (m); 1.45 (m)	1.69 (m); 1.51 (m)
3	1.56 (m); 1.27 (m)	3	1.59 (m); 1.18 (m)	1.50 (m); 1.30 (m)	1.48 (m); 1.40 (m)
5	1.48 (m)	5	1.36 (m)	1.38 (m)	1.32 (m)
6	1.55 (m); 1.25 (m)	6	1.52 (m); 1.27 (m)	1.55 (m); 1.36 (m)	1.62 (m); 1.32 (m)
7	1.70 (m); 1.26 (m)	7	1.74 (m); 1.52 (m)	1.75 (m); 1.47 (m)	1.80 (m); 1.50 (m)
9	1.16 (m)	9	1.35 (m)	1.24 (m)	1.00 (m)
11	1.33 (m)	11	1.47 (m); 1.29 (m)	1.50 (m); 1.36 (m)	1.43 (m)
12	1.73 (m); 1.55 (m)	12	1.78 (m); 1.56 (m)	1.62 (m); 1.51 (m)	1.76 (m); 1.54 (m)
14	5.91 (dd, 10.8, 17.6)	14	5.89 (dd,10.5, 17.0)	5.91 (dd, 10.8, 17.6)	5.88 (dd, 10.8, 18.0)
15	4.92 (d, 10.8); 5.19 (d, 17.6)	15	5.15 (d, 17.0); 4.91(d, 10.5)	5.17 (d, 17.6); 5.15 (d, 10.8)	5.18 (m); 5.16 (m)
16	1.20 (s)	16	1.05 (s)	1.09 (s)	1.09 (s)
17	1.10 (s)	17	1.19 (s)	1.32 (s)	1.32 (s)
18	3.45 (m); 3.25 (m)	18	3.38 (m); 3.20 (m)	3.48 (m); 3.21 (m)	3.44 (m); 3.35 (m)
19	0.75 (s)	19	0.70 (s)	0.76 (s)	0.81 (s)
20	0.82 (s)	20	0.79 (s)	0.82 (s)	0.85 (s)
1′	4.30 (d, 8.0)	1′	4.27 (d, 7.5)	4.25 (d, 7.2)	4.42 (d, 7.2)
2'	3.33 (m)	2'	3.24 (m)	3.45 (m)	3.53 (m)
3'	3.68 (m)	3'	3.52 (m)	3.46 (m)	3.68 (m)
4′	4.84 (t, 9.6)	4′	3.26 (m)	3.24 (m)	3.36 (m)
5'	3.48 (m)	5'	3.48 (m)	3.19 (m)	3.35 (m)
6'	3.54 (m); 3.58 (m)	6'	3.79 (m); 3.62 (m)	3.64 (m); 3.83 (m)	3.68 (m); 3.84 (m)
2''	6.36 (d, 15.6)	1''	5.51 (d, 1.5)	5.42 (d, 1.5)	5.62 (d, 0.8)
3''	7.64 (d, 15.6)	2''	3.87 (m)	3.91 (m)	5.32 (m)
5''/9''	7.55 (d, 8.8)	3''	3.77 (m)	3.74 (m)	5.15 (m)
6''/8''	6.90 (d, 8.8)	4''	3.43 (m)	3.40 (m)	5.12 (m)
		5''	4.08 (m)	4.06 (m)	4.46 (m)
		6''	1.25 (d, 6.5)	1.27 (d, 6.0)	1.18 (d, 6.4)
		1‴		4.28 (d, 8.0)	4.52 (d, 8.0)
		2‴′		3.41 (m)	5.37 (m)
		3‴		3.40 (m)	3.87 (m)
		4‴		3.55 (m)	4.92 (m)
		5‴		3.52 (m)	3.82 (m)
		6‴′		1.20 (d, 6.4)	1.08 (d, 6.4)

<sup>a</sup>Measured in acetone- $d_6$ . <sup>b</sup>Determined in CD<sub>3</sub>OD. <sup>c</sup>Acetyl groups  $\delta_{\rm H}$ : (1.98, 2.06, 2.13, 2.13, 2.16).

of the sugar unit was established by the HMBC correlations of H-1' ( $\delta_{\rm H}$  4.30) with C-18 ( $\delta_{\rm C}$  79.2).

The relative configuration of the aglycone was established on a ROESY experiment. The ROESY correlations (Figure 2) between H-5 and H-9 confirmed that these hydrogen atoms were  $\alpha$ -oriented, while correlations of H-11/Me-17, H-11/Me-20, Me-17/Me-19, Me-17/Me-20, and Me-19/Me-20 indicated they were  $\beta$ -orientation. The absolute configuration of C-13 was inferred as *S* according to the chinical shift of C-13 ( $\delta_C$ 73.4).<sup>12-14</sup> Therefore, the structure of **1** was determined as shown, named laevissioside A.

Compound **2**, a white amorphous powder, and its molecular formula,  $C_{32}H_{56}O_{12}$ , was determined on the basis of the HRESIMS (667.3454 [M + Cl]<sup>-</sup>; calcd. 667.3460). The <sup>1</sup>H and <sup>13</sup>C NMR spectroscopic data of **2** were very similar to those of **5**, except for one more sugar moiety signals ( $\delta_C$  100.8, 72.0, 71.7, 73.7, 69.2, 18.3) presented in **2** which was further confirmed by mass spectra. Acidic hydrolysis of **2** gave D-glucose and L-rhamnose as sugar residues. The coupling



Figure 1. Selected HMBC correlations of compound 1 and 2

constants ( $\delta_{\rm H}$  4.27, J = 7.5 Hz and  $\delta_{\rm H}$  5.51, J = 1.5 Hz) of anomeric protons of the two sugar moieties indicated the  $\beta$ configuration glucose and  $\alpha$  configuration of rhamnose. The HMBC correlations between H-1" ( $\delta_{\rm H}$  5.51) and C-2' ( $\delta_{\rm C}$  77.5) identified a rhamnosyl (1->2) glucopyranosyl linkage. Furthermore, the sugar chain was linked to C-18 of the aglycone as inferred from the HMBC (Figure 1) correlation of



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Table 2. <sup>13</sup>C NMR data of compounds 1-4 ( $\delta$  in ppm)

no.	1 <sup>a</sup>	no.	<b>2</b> <sup>a</sup>	<b>3</b> <sup>b</sup>	<b>4</b> <sup>a,c</sup>
1	39.8 CH <sub>2</sub>	1	39.7 CH <sub>2</sub>	40.4 CH <sub>2</sub>	39.5 CH <sub>2</sub>
2	18.5 CH <sub>2</sub>	2	18.4 CH <sub>2</sub>	18.8 CH <sub>2</sub>	18.4 CH <sub>2</sub>
3	36.6 CH <sub>2</sub>	3	36.5 CH <sub>2</sub>	37.0 CH <sub>2</sub>	36.4 CH <sub>2</sub>
4	37.9 C	4	37.8 C	38.3 C	38.2 C
5	49.6 CH	5	48.9 CH	50.7 CH	51.2 CH
6	21.0 CH2	6	20.9 CH <sub>2</sub>	21.4 CH <sub>2</sub>	21.5 CH <sub>2</sub>
7	44.6 CH <sub>2</sub>	7	44.7 CH <sub>2</sub>	44.8 CH2	45.4 CH <sub>2</sub>
8	73.9 C	8	74.6 C	75.4 C	73.8 C
9	62.1 CH	9	60.9 CH	62.0 CH	62.9 CH
10	40.0 C	10	40.0 C	40.4 C	40.0 C
11	20.1 CH <sub>2</sub>	11	20.4 CH <sub>2</sub>	20.5 CH <sub>2</sub>	19.9 CH <sub>2</sub>
12	46.5 CH <sub>2</sub>	12	45.7 CH <sub>2</sub>	44.8 CH <sub>2</sub>	46.7 CH <sub>2</sub>
13	73.4 C	13	73.7 C	81.8 C	80.7 C
14	147.5 CH	14	147.0 CH	145.0 CH	143.5 CH
15	110.8 CH <sub>2</sub>	15	111.1 CH <sub>2</sub>	115.2 CH <sub>2</sub>	115.5 CH <sub>2</sub>
16	28.1 CH <sub>3</sub>	16	23.6 CH <sub>3</sub>	23.0 CH3	23.8 CH3
17	24.6 CH <sub>3</sub>	17	$27.3 \ \mathrm{CH}_3$	23.7 CH <sub>3</sub>	24.6 CH <sub>3</sub>
18	79.2 CH <sub>2</sub>	18	$78.2 \ \mathrm{CH}_2$	79.4 CH <sub>2</sub>	79.0 CH <sub>2</sub>
19	17.9 CH <sub>3</sub>	19	18.2 CH <sub>3</sub>	18.2 CH <sub>3</sub>	17.9 CH <sub>3</sub>
20	16.2 CH <sub>3</sub>	20	16.2 CH <sub>3</sub>	16.5 CH <sub>3</sub>	16.3 CH <sub>3</sub>
1′	104.6 CH	1′	102.9 CH	103.5 CH	102.5 CH
2'	75.2 CH	2'	77.5 CH	78.0 CH	75.9 CH
3'	75.5 CH	3'	79.6 CH	79.8 CH	79.3 CH
4′	72.3 CH	4′	71.9 CH	72.0 CH	72.1 CH
5'	75.8 CH	5'	76.8 CH	77.9 CH	77.5 CH
6'	$62.4\ \mathrm{CH}_2$	6'	62.6 CH <sub>2</sub>	62.7 CH <sub>2</sub>	62.5 CH <sub>2</sub>
1''	167.4 C	1′′	100.8 CH	101.5 CH	97.8 CH
2''	115.2 CH	2''	72.0 CH	72.3 CH	70.5 CH
3′′	146.1 CH	3''	71.7 CH	72.1 CH	71.0 CH
4''	121.8 C	4''	73.7 CH	74.0 CH	74.0 CH
5''/9''	116.7 CH	5''	69.2 CH	70.0 CH	67.1 CH
6''/8''	131.0 CH	6''	18.3 CH <sub>3</sub>	18.8 CH <sub>3</sub>	17.6 CH <sub>3</sub>
7''	160.8 C	1‴		99.6 CH	97.1 CH
		2‴′		72.4 CH	70.1 CH
		3‴′		75.2 CH	71.1 CH
		4‴′		73.0 CH	73.4 CH
		5‴′		71.5 CH	69.5 CH
		6‴		17.0 CH <sub>3</sub>	16.9 CH <sub>3</sub>

<sup>a</sup>Measured in acetone-*d*<sub>6</sub>. <sup>b</sup>Determined in CD<sub>3</sub>OD. <sup>c</sup>Acetyl groups (δ<sub>C</sub>: C=O 170.0, 170.4, 170.6, 171.0, 171.5; Me: 20.7, 20.8, 21.2, 21.2, 21.2).

H-1' ( $\delta_{\rm H}$  4.27) with C-18 ( $\delta_{\rm C}$  78.2). Thus, the structure of **2** was elucidated and named as laevissioside B.

Compound **3**, obtained as a white amorphous powder, had the molecular formula  $C_{38}H_{66}O_{16}$  as determined by the analysis of <sup>1</sup>H, <sup>13</sup>C, and DEPT NMR data and verified by the HRESIMS (813.4042 [M + Cl]<sup>-</sup>; calcd. 813.4039). Compared with **2**, compound **3** had one more sugar moiety ( $\delta_C$  99.6, 72.4, 75.2, 73.0, 71.5, and 17.0) which was attached to C-13 as inferred from the HMBC correlations from H-1''' to C-13. Acidic hydrolysis of **3** gave D-glucose, L-rhamnose, and Dfucose as sugar residues. They were in  $\beta$ ,  $\alpha$  and  $\beta$ configurations, respectively, by the coupling constants ( $\delta_H$ 





4.25, J = 7.2 Hz,  $\delta_{\rm H} 5.42$ , J = 1.5 Hz, and  $\delta_{\rm H} 4.28$ , J = 8.0 Hz) of their anomeric protons. The sugar moiety linked to C-13 was supposed to be fucose by the HMBC correlations of the anomeric proton H-1"" ( $\delta_{\rm H} 4.28$ ) with C-13 ( $\delta_{\rm C} 81.8$ ). The linkage and location of the other two sugar moieties were suggested the same as compound **2** deduced from the HMBC correlations. Accordingly, the structure of **3** was established as shown, named laevissioside C.

The molecular formula of compound **4** was deduced as  $C_{48}H_{76}O_{21}$  by the HRESIMS (1023.4557 [M + Cl]<sup>-</sup>; calcd 1023.4567). The <sup>1</sup>H and <sup>13</sup>C NMR features of **4** were closely related to those of **3**. The only difference was that there were five more acetyl groups in **4**. The locations of acetyl groups were confirmed by the HMBC experiments. The correlations of H-2<sup>'''</sup> ( $\delta_{H}$  5.37) with  $\delta_{C}$  170.4, H-4<sup>'''</sup> ( $\delta_{H}$  4.92) with  $\delta_{C}$  170.0, H-2<sup>''</sup> ( $\delta_{H}$  5.32) with  $\delta_{C}$  171.5, H-3<sup>''</sup> ( $\delta_{H}$  5.15) with  $\delta_{C}$  171.0, and H-4<sup>'''</sup> ( $\delta_{H}$  5.12) with  $\delta_{C}$  170.6 indicated the five acetyl groups attached to C-2<sup>'''</sup>, C-4<sup>'''</sup>, C-2<sup>''</sup>, C-3<sup>''</sup>, and C-4<sup>''</sup>, respectively. Therefore, the structure of **4** was identified and named as laevissioside D.



Figure 2. Significant ROESY correlations of compounds 1-4

Since the known compounds  $18-\beta$ -D-glucopyranosyl estersclareol (5,  $[\alpha]_{D}^{24.3} - 27.7$ ) and 18-hydroxy-sclareol (6,  $[\alpha]_{D}^{24.3} - 4.7$ ), whose physical properites were quite difference with that reported,<sup>15,16</sup> were also isolated from this plant, compounds 1–4 should be labdane-type diterpenoid glycosides from the biogenic view.

All compounds isolated were evaluated for their cytotoxic activity against five human cancer cell lines, HL-60 myeloid leukemia, SMMC-7721 hepatocellular carcinoma, A-549 lung cancer, SK-BR-3 breast cancer, PANC-1 pancreatic cancer, applying the MTT method. However, all of the compounds were inactive, and they showed IC<sub>50</sub> values > 40  $\mu$ M.

# **Experimental Section**

**General Experimental Procedures.** Optical rotations were measured on a Horiba SEPA-300 polarimeter. IR spectra were obtained by Tensor 27 FT-IR spectrometer with KBr pellets. The <sup>1</sup>H and <sup>13</sup>C NMR spectra were recorded on Bruker AV-400 spectrometers in acetone- $d_6$  at room temperature ( $\delta$  in ppm, *J* in Hz). FABMS was carried out on a VG Autospec-3000 spectrometer. HRESIMS was recorded with an API QSTAR Pulsar i spectrometer. Silica gel (200–300 mesh), Silica gel H (Qingdao Marine Chemical Ltd., China), and LiChroprep RP-18 silica gel (40–63  $\mu$ m, Merck, Dramstadt, Germany) were used for column chromatography. Fractions were monitored by TLC and spots visualized by heating silica gel plates immersed with 15% H<sub>2</sub>SO<sub>4</sub> in ethanol. Solvents were distilled prior to use. Preparative HPLC was performed on a Shimadzu

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LC-8A preparative liquid chromatograph with Shimadzu PRC-ODS (K) column. Sephadex LH-20 (Amersham Pharmacia biotech, Sweden).

**Plant Material.** The aerial parts of *D. laevissimum* were collected from Pingbian, Yunnan Province, China in July 2007 and identified by Professor Xiao Cheng at Kunming Institute of Botany, Chinese Academy of Sciences. A voucher specimen (No. 200707A03) has been deposited in the State Key Laboratory of Phytochemistry and Plant Resources in West China, Kunming Institute of Botany, Chinese Academy of Sciences.

**Extraction and Isolation.** The dried and powdered plant materials (2.6 kg) were extracted with 95% ethanol (15.0 L, each 2 d) for three times. After evaporation of the solvent in vacuo, the concentrate was suspended into  $H_2O$  and partitioned successively with ethyl acetate. The ethyl acetate extract (120 g) was chromatographed on a silica gel column eluted with CHCl<sub>3</sub>-MeOH (1:0 to 5:5) to give five fractions 1–5. Fraction 2 was subjected to column chromatograph (CC) over silica gel (petroleum ether-acetone 8.5:1.5) and further purified by recrystallization to obtain **6** (20 mg). Fraction 3 was eluted with CHCl<sub>3</sub>-MeOH (9:1) over silica gel CC then further purified by RP-18 and Sephadex LH-20 to yield **1** (3 g), **4** (500 mg), and **5** (2 g). Fraction 4 was subjected to (CHCl<sub>3</sub>:MeOH = 8.5:1.5) and further purified by RP-18 and Sephadex LH-20 to afford **2** (8 mg) and **3** (20 mg).

**Laevissioside A:** amorphous powder;  $[\alpha]_{D}^{19.2} - 50.4$  (c = 0.14, MeOH). UV (MeOH)  $\lambda_{max}$  (log  $\varepsilon$ ): 314 (4.18), 227 (3.94), 211 (3.95), 200 (3.99), 192 (3.91) nm. IR (KBr): 3428, 2931, 1704, 1630, 1604, 1515, 1450, 1387, 1162, 1066, 1031 cm<sup>-1</sup>. <sup>1</sup>H and <sup>13</sup>C NMR: see Table 1. FABMS (neg.) *m/z*: 631 [M – H]<sup>-</sup>. HRESIMS (neg.) *m/z*: 631.3471 (C<sub>35</sub>H<sub>51</sub>O<sub>10</sub>; calcd. 631.3482).

**Laevissioside B:** amorphous powder;  $[\alpha]_{D}^{26.4} - 62.2$  (c = 0.18, MeOH). UV (MeOH)  $\lambda_{max}$  (log  $\varepsilon$ ): 203 (3.38), 194 (3.12) nm. IR (KBr): 3423, 2927, 1069, 1052 cm<sup>-1</sup>. <sup>1</sup>H and <sup>13</sup>C NMR: see Table 2. FABMS (neg.) m/z: 631 [M – H]<sup>-</sup>. HRESIMS (neg.) m/z: 667.3454 (C<sub>32</sub>H<sub>56</sub>O<sub>12</sub>Cl; calcd. 667.3460).

**Laevissioside C:** amorphous powder;  $[\alpha]_{D}^{18.9} - 11.2$  (c = 0.10, MeOH). UV (MeOH)  $\lambda_{max}$  (log  $\varepsilon$ ): 201 (3.31), 191 (3.10) nm. IR (KBr): 3431, 2926, 1704, 1638, 1384, 1169, 1128, 1054 cm<sup>-1</sup>. <sup>1</sup>H and <sup>13</sup>C NMR: see Table 2. FABMS (neg.) *m/z*: 777 [M - H]<sup>-</sup>, 631 [M - 146 - H]<sup>-</sup>. HRESIMS (neg.) *m/z*: 813.4042 ( $C_{38}H_{66}O_{16}Cl$ ; calcd. 813.4039).

**Laevissioside D:** amorphous powder;  $[\alpha]_{D}^{26.6} - 58.3$  (c = 0.25, MeOH). UV (MeOH)  $\lambda_{max}$  (log  $\varepsilon$ ): 199 (3.07), 192 (3.12) nm. IR (KBr): 3442, 2935, 1749, 1373, 1228, 1062, 1062 cm<sup>-1</sup>. <sup>1</sup>H and <sup>13</sup>C NMR: see Table 2. Acetyl groups ( $\delta_{C}$ : C=O 170.0, 170.4, 170.6, 171.0, 171.5; Me: 20.7, 20.8, 21.2, 21.2, 21.2)  $\delta_{H}$ : (1.98, 2.06, 2.13, 2.13, 2.16). FABMS (neg.) m/z: 987 [M –

H]<sup>-</sup>. HRESIMS (neg.) m/z: 1023.4557 (C<sub>48</sub>H<sub>76</sub>O<sub>21</sub>Cl; calcd. 1023.4567).

Acidic Hydrolysis of Compounds 1-3. Compounds 1-3 (6-8 mg) were hydrolyzed with 2 M HCl-dioxane (1:1, 4 mL) under reflux for 6 h. The reaction mixture was extracted with CHCl<sub>3</sub> five times (4 mL  $\times$  5). The aqueous layer was neutralized with 2 M NaHCO<sub>3</sub>, and was evaporated to dryness. The dry powders were dissolved in pyridine (2 mL). Then L-cysteine methyl ester hydrochloride (about 1.5 mg) was added and kept at 60 °C for 1 h. Next, trimethylsilylimidazole (about 1.5 mL) was added to the reaction mixture in ice water and kept at 60 °C for 30 min. The mixture was subjected to GC analysis, run on a Shimadzu GC-14C gas chromatograph equipped with a 30 m  $\times$  0.32 mm i.d. 30QC2/AC-5 quartz capillary column and an H<sub>2</sub> flame ionization detector with the following conditions: column temperature, 180-280 °C; programmed increase, 3 °C/min; carrier gas, N<sub>2</sub> (1 mL/min); injector and detector temperature, 250 °C; injection volume, 4  $\mu$ L; and split ratio, 1/50. The configuration of D-glucose, L-rhamnose, and D-fucose were determined by comparison of the retention time of the corresponding derivatives with those of standard D-glucose, L-rhamnose, and D-fucose, giving a peak at 18.576, 16.173, and 14.865 min, respectively.

**Cytotoxic Bioassay.** The cytotoxicity assay was performed according to the MTT [3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide] method,<sup>17</sup> by use of the following five human cancer cell lines: Human myeloid leukemia HL-60, hepatocellular carcinoma SMMC-7721, lung cancer A-549, breast cancer SK-BR-3, and pancreatic cancer PANC-1. The IC<sub>50</sub> values were calculated by the Reed and Muench method.<sup>18</sup>

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