Antibacterial potential of biosynthesised silver nanoparticles using *Avicennia marina* mangrove plant

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Abstract The present study was aimed to identify the antibacterial potential of biosynthesised silver nanoparticles using different plant parts (leaves, bark and root) of *Avicennia marina* mangrove plant. Of the selected three different parts, the leaf extract showed the maximum synthesis of silver nanoparticles. The in vitro antibacterial assay (100 µg disk⁻¹ concentration) showed the results of maximum zone of inhibition with the *E. coli* (18.40 ± 0.97 mm), and minimum (10.87 ± 1.33 mm) zone of inhibition with *S. aureus* but the concentrations of MIC and MBC values ranged between 6.25 and 50.0 µg ml⁻¹ between the selected bacterial strains. The FTIR results of most potent leaf extract-synthesized silver nanoparticles showed the prominent peaks (620.967; 1,061.02; 1,116.58; 1,187.94; 1,280.50; 1,353.79; 1,384.64; 1,598.50; 1,629.56; 2,854.14 and 2,927.42) in different ranges. Further, the results of XRD analysis showed the 2θ intense values (38.11 and 70.57) within the ranges of Bragg's reflection. In addition, the AFM analysis showed the results of particle sizes (71–110 nm), particle roughness (11.8 nm), maximum

height of the particle roughness (111.8 nm), and average maximum height of the particle roughness (57.5 nm). It can be concluded from the present findings that, the biosynthesis of silver nanoparticles from the leaf extract of *A. marina* can be used as potential antibacterial agents.

Keywords Biosynthesis · Antibacterial · *Avicennia marina* · Silver nanoparticles

Introduction

Nanoparticles play an important role in pharmaceutical, industrial and biotechnological applications. In particular, silver nanoparticles are proved to have potential antibacterial, antifungal and antiplasmodial and larvicidal properties (Saxena et al. 2010; Elumalai et al. 2010). But the synthesis of nanoparticles using chemical and physical methods requires high pressure, energy, temperature and toxic chemicals. In this regard, plants and plant part extracts based biosynthesis has been found to be cost effective and environmental friendly (Casida and Quistad 2005). But studies related to the synthesis of nanoparticles using mangrove sand mangrove associate plants are very limited. Marine environmental conditions are extremely different from terrestrial ones; it is surmised that mangrove plants have different characteristics from those of terrestrial plants, and therefore, might produce different types of bioactive compounds (Gnanadesigan et al. 2011a; Ravikumar et al. 2011a, b). Among the different mangrove plants, *Avicennia marina* is previously proved to have antibacterial, antiplasmodial, antiviral activities (Ravikumar et al. 2011c) and also it is proved to have high content of secondary metabolites such as polyphenols, flavonoids, alkaloids and tannins (Ravikumar et al. 2010). In this connection, the present study made

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an attempt to identify the antibacterial potential of biosynthesised silver nanoparticles using different plant parts of *Avicennia marina* mangrove plant.

Materials and methods

Collection of plant material

Different plant parts (leaves, bark and root) of *A. marina* were collected from Karangadu mangrove forest (latitude 9°38'N and longitude 78°57'E), South East coast of India, Ramanathapuram district, Tamilnadu, India. The authentication of the plant species was done by Prof. K. Kathiresan, Centre of Advanced Study in Marine Biology, Annamalai University, Porto Novo, Tamil Nadu, India. The voucher specimen (MKUCMS0024) was also maintained in the Department of Marine and Coastal Studies, Madurai Kamaraj University, Madurai, Tamilnadu, India.

Biosynthesis of silver nanoparticles

The collected samples were washed thrice with tap water and twice with distilled water to remove the adhering salts and other associated contaminants. About 10 g of various plant parts were finely cut into small pieces and boiled with 100 ml of double sterilized distilled water separately for 5 min. The boiled extracts were filtered through Whatmann no. 1 filter paper. A total of 10 ml of collected filtrate was treated with 90 ml of silver nitrate aqueous solution (21.2 g of AgNO_3 powder in 125 ml of Milli Q water) and incubated at room temperature for 10 min, resulting in the formation of brownish yellow color indicating the synthesis of silver nanoparticles (Parashar et al. 2009). After that, about 1 ml (diluted with 1:20 V/V Milli Q water) of different plant-part-synthesised silver nanoparticle solution was monitored in UV–Vis spectrophotometer (between 300 and 700 nm with 5 nm intervals) with different time intervals (15 and 30 min; 4, 6 and 8 h). After 8 h of incubation, the solution was centrifuged with 12,000 rpm for 20 min, and their pellets were re dispersed in sterile distilled water. The centrifugation and redispersion was repeated three times to ensure the complete separation of nanoparticles.

In vitro antibacterial activity

The in vitro antibacterial activity was performed with the leaf extract (the maximum silver nanoparticle producer) synthesised silver nanoparticles, and chemically synthesized silver nanoparticle (obtained from Department of Inorganic Chemistry, Madurai Kamaraj University, Madurai). In brief, 5 ml suspension of nanoparticles (5 mg ml^{-1}) was sonicated and subsequently filtered through a membrane filter

(0.2 μm , 47 mm diameter, Sartorius filter paper). The nanoparticles laden filter paper was dried in an oven for 1 h, and small disks of uniform size (6 mm dia) containing $100 \pm 15 \mu\text{g}$ nanoparticles were punched out, and stored in a desiccator at room temperature. Five different human bacterial pathogens such as *E. coli*, *Pseudomonas aeruginosa*, *Staphylococcus aureus*, *Streptococcus pyogenes* and *Bacillus subtilis* ($100 \mu\text{l}$ of 10^6 CFU ml^{-1}) were spread over the nutrient agar plate, before placing the disks on the plate. The plates were incubated at 35°C for 24 h, after which the average diameter of the inhibition zone surrounding the disk was measured. The mean and standard deviation (SD) were reported for each bacterial strain based on six replicates. Minimum inhibitory concentration (MIC) was also carried out with the leaf extract-synthesized silver nanoparticles. In brief, about 0.5 ml of various concentrations (3.125, 6.25, 12.5, 25, 50, 100, 200 and $400 \mu\text{g ml}^{-1}$) of the leaf extract-synthesized silver nanoparticle was prepared with dimethyl sulphoxide (DMSO), and mixed with 50 μl of 24 h old bacterial pathogens individually and incubated at 37°C for 48 h individually. To calculate the MIC, turbidity due to bacterial growth was observed in each concentration. To avoid the possibility of misinterpretations due to the turbidity of insoluble compounds, the minimum bactericidal concentration (MBC) was determined by sub-culturing the MIC dilutions onto the sterile agar plates. The lowest concentration of the extracts which inhibits the growth of tested bacteria is observed and tabulated.

Characterization of bio synthesised silver nanoparticles

The purified pellet was dried and subjected to the FTIR spectroscopy measurement in the diffuse reflectance mode at a resolution of 4 cm^{-1} in KBr pellets. The dried mixture of silver nanoparticles was further analyzed with X-ray diffractometer (PANalytical BV, The Netherlands) operated at a voltage of 40 kV, and a current of 30 mA, with Cu K α radiation in a θ -2 θ configuration. In addition, a thin film of sample was also prepared in the cover slip with 100 μl synthesised silver nanoparticles solution, and allowed to dry for 5 min, and the slides were analyzed with atomic force microscopy.

Results

The color intensity of the synthesised silver nanoparticles increased with the increased time duration. Of the selected three different plant parts, the leaf extract showed maximum level of OD value (1.39) after 8 h of incubation (Fig. 1). The results of disk diffusion assay showed the maximum zone of inhibition with the *E. coli* ($18.40 \pm 0.97 \text{ mm}$), followed by *Pseudomonas aeruginosa* ($17.64 \pm 0.91 \text{ mm}$) but the

Fig. 1 UV-Vis spectral analysis of silver nanoparticles using various plant parts of *A. marina* extracts

minimum zone (10.87 ± 1.33 mm) of inhibition was identified with the *Staphylococcus aureus* species. Further, the resulted concentrations of the MIC and MBC values varied between 6.25 and $50.0 \mu\text{g ml}^{-1}$ between the organisms. Further, results of antibacterial activity of disk diffusion assay, MIC and MBC with chemically synthesized silver nanoparticles are also comparable with the biosynthesized silver nanoparticle (Table 1). The results of the FTIR spectrum showed prominent peaks with different (620.967; 1,061.02; 1,116.58; 1,187.94; 1,280.50; 1,353.79; 1,384.64; 1,598.50; 1,629.56; 2,854.14 and 2,927.42) values (Fig. 2). The results of XRD 2θ intense values ranged with different degrees (38.11 and 70.57), and these results correspond to the (111) and (311) Bragg's reflection, respectively (JCPDS NO 03-0931) (Sathishkumar et al. 2009) (Fig. 3). The AFM topology results of the leaf extract-synthesized silver nanoparticles showed the average particle sizes (71–110 nm), particle roughness (11.8 nm), maximum height of the particle roughness (111.8 nm) and average maximum height of the particle roughness (57.5 nm) (Fig. 4).

Discussion

Nanotechnology is an emerging trend in the field of science and technology for the manufacturing new materials at

nanoscale levels (Tripathi et al. 2009). In the present scenario plant extract based nanoparticles biosynthesis is emerging as novel antimicrobial agents with biological, chemical and physical properties (Gong et al. 2007; Krishnaraj et al. 2010). In view of this, the present study aimed to make an attempt to identify the synthesis and antibacterial properties of silver nanoparticles using the various parts of *A. marina* mangrove plant. Of the selected different parts, the leaf extract showed maximum yield of silver nanoparticles than the bark and root aqueous extracts, and this might be due to the presence of high content of secondary metabolites such as polyphenols, flavonoids and tannins (Gnanadesigan et al. 2011b; Ravikumar et al. 2009; Ravikumar and Gnanadesigan 2011). Further, the biosynthesis of silver nanoparticles can be confirmed by the formation of yellowish brown color, and this might be due to the excitation of the surface plasmon vibration of the synthesised silver nanoparticles (Jayaseelan et al. 2010). In addition, the results of XRD pattern further corroborate the synthesis of silver nanoparticles with sharp bands of Bragg peaks, and this might be due to the stabilization of the synthesised nanoparticles by the various reducing agents of the *A. marina* leaf extract, and thus provides the crystallization nature of the silver nanoparticles (Asmathunisha et al. 2010). Further, the results of the FTIR values of synthesised silver nanoparticles showed the presence of various functional groups such as alkane groups, methylene groups, alkene groups, amine groups and carboxylic acids, and these functional groups are the major classes in many of the chemical groups and these chemical groups are previously proved to have potential reducing agents in the synthesis of silver nanoparticles (Cho et al. 2005). Moreover, the present study also proved to have potential antibacterial activities with the *A. marina* aqueous leaf extract synthesised silver nanoparticles and this might be due to denaturation of bacterial cell wall, blocking bacterial respiration, destabilization of outer membrane and depletion of intracellular ATP (Vivekanandhan et al. 2009). Further, the variation of the sensitivity between Gram positive and negative bacterial

Table 1 In vitro antibacterial potential of biosynthesized silver nanoparticles using *A. marina* leaf extract

Pathogens	<i>A. marina</i> leaf extract			Chemically synthesized silver nanoparticle		
	Disk diffusion assay (mm dia)	MIC ($\mu\text{g ml}^{-1}$)	MBC ($\mu\text{g ml}^{-1}$)	Disk diffusion assay (mm dia)	MIC ($\mu\text{g ml}^{-1}$)	MBC ($\mu\text{g ml}^{-1}$)
<i>E. coli</i>	18.40 ± 0.97	6.25 ± 0.00	12.5 ± 0.00	22.20 ± 0.75	3.125 ± 0.00	3.125 ± 0.00
<i>Pseudomonas aeruginosa</i>	17.64 ± 0.91	12.5 ± 0.00	50.0 ± 0.00	19.64 ± 1.98	12.5 ± 0.00	25.0 ± 0.00
<i>Staphylococcus aureus</i>	10.87 ± 1.33	25.0 ± 0.00	50.0 ± 0.00	14.90 ± 1.24	25.0 ± 0.00	50.0 ± 0.00
<i>Klebsiella</i> sp	14.65 ± 1.09	6.25 ± 0.00	25.0 ± 0.00	17.65 ± 2.19	3.125 ± 0.00	6.25 ± 0.00
<i>Bacillus subtilis</i>	13.93 ± 0.84	25.0 ± 0.00	50.0 ± 0.00	13.93 ± 1.38	12.5 ± 0.00	25.0 ± 0.00

Values $\pm$ SD indicates the replicates of three experiments

Fig. 2 FTIR spectrum analysis of biosynthesised silver nanoparticles using *A. marina* leaf extract. Values indicate the functional groups of various compounds

Fig. 3 XRD analysis of biosynthesised silver nanoparticles using *A. marina* leaf extract

Fig. 4 AFM analysis of biosynthesised silver nanoparticles using *A. marina* leaf extract. Numbers indicate the different sizes of nanoparticles between 71 and 110 nm ranges

isolates might be attributed by the membrane permeability (Ravikumar et al. 2010). In spite of this permeability barrier, the synthesized silver nanoparticles exhibited the

strong inhibition with gram-negative bacterial strains. The size and shape of nanoparticles play an important role in many of the pharmaceutical applications. In this regard, the

size of the synthesised nanoparticles was identified between 71 and 110 nm with various spherical shapes, which falls closer to many of the silver nanoparticles produced by other plant materials (Vivekanandhan et al. 2009; Tripathi et al. 2009; Gnanadesigan et al. 2011a). It is concluded from the present findings that, the biosynthesised silver nanoparticles using leaf aqueous extract of *A. marina* showed potential antibacterial activity with various bacterial pathogens which could be further used as a potential antibacterial agents.

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