

The Nano Research Young Innovators (NR45) Awards in nanoenergy

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Following the success of the inaugural 2018 *Nano Research* Young Innovators (NR45) Awards in nanobiotechnology [1], it is our great pleasure to announce the recipients of the 2019 NR45 Awards in nanoenergy. Congratulations to all the 44 outstanding young investigators under the age of 45. They were selected through a competitive process by an award committee from the journal's editorial board. *Nano Research* launched the NR45 Awards program to feature young researchers in various fields of nanoscience and nanotechnology, in recognition of their distinguished accomplishments and/or potential to make substantial contributions to their fields. NR45 awardees are recognized during the *Nano Research* Symposium integrated in the annual US-SINO Nano Forum, which provides a platform for stimulating communication and fostering collaborations among scientists across the globe. This year, we highlight 44 innovators in the field of energy, covering research foci including batteries, catalysis, fuel cells, photovoltaics, supercapacitors, solar fuels, bioelectricity, and fundamental studies of materials synthesis, structures and properties. Their contributions to this special issue contain 16 review articles and 28 research articles.

This special issue starts with review articles on energy storage materials. Guihua Yu from the University of Texas at Austin reviews the development of stretchable and conductive polymer materials, including molecular design, materials processing and electrode/device engineering, for supercapacitors [2], which are a type of devices that can store and release electrical energy in seconds. Hye Ryung Byon from Korea Advanced Institute of Science and Technology highlights the design principles of organic molecular materials, partially enabled by nanotechnology, for aqueous redox flow batteries [3], a type of devices suitable for grid-scale energy storage. Yuan Yang from Columbia University surveys the importance, preparation and performance of vertically-aligned nanostructures as battery electrodes [4]. Yong-Sheng Hu from Institute of Physics, Chinese Academy of Sciences summarizes the strategies for developing Ni-based layered metal oxide cathode materials for Na ion batteries [5].

Catalytic materials are another area of focus. Yujie Xiong from University of Science and Technology of China reviews the recent progress in controlling the phases of photocatalyst materials, with new insights into rational phase design for improved performance in energy and environment related reactions [6]. Shuangyin Wang from Hunan University summarizes the design principles and synthetic strategies, including active site exposure, mass transport enhancement, and electron transfer acceleration, for the triple-phase interfaces of electrocatalytic systems involving H₂, O₂ and H₂O [7]. Dingsheng Wang from Tsinghua University reviews the design, synthesis, characterization, and electrocatalytic properties of a type of emerging catalyst materials, i.e. single atom catalysts, derived from metal

organic frameworks (MOFs) [8]. Eranda Nikolla from Wayne State University discusses nanoengineering strategies employed to advance electrocatalyst materials for solid oxide fuel cells and electrolyzers [9]. Karthish Manthiram from Massachusetts Institute of Technology provides an in-depth overview of the state-of-the-art heterogenized molecular catalysts for electrochemical CO₂ reduction to fuels and useful chemicals [10].

Judy Cha from Yale University summarizes the recent progress on *in-situ* characterizations of electrochemical intercalation processes for transition metal chalcogenide materials [11]. Yanfeng Zhang from Peking University reviews on-site scanning tunneling microscopy/spectroscopy investigations of the atomic structures and electronic properties of monolayer MoS₂ and its heterostructures, and discusses the implications for electrocatalytic H₂ evolution [12]. Hui Wu from Tsinghua University summarizes the synthetic routes for oxygen-deficient metal oxide materials and their applications in catalysis, energy storage, solar cells and pollutant removal [13]. Jing Zhao from University of Connecticut reviews colloiddally fabricated metal-quantum emitter nanostructures and discusses the plasmon-exciton interactions therein [14]. Chinedum O. Osuji from University of Pennsylvania reviews self-assembly processes of soft matter to form nanostructures and their applications as optical materials and metamaterials [15]. Ardemis A. Boghossian from Ecole Polytechnique Fédérale de Lausanne (EPFL) discusses interfacing biological parts with nanoparticles as an approach to enhance bioelectricity generation in microbial fuel cells and biophotovoltaics [16]. Elad Gross from the Hebrew University of Jerusalem overviews the opportunities and challenges of employing infrared nanospectroscopy to study the chemical information of energy materials [17].

These review articles are followed by original research papers. On battery materials, Yan Yu from University of Science and Technology of China reports Cu-modified multichannel porous TiO₂ nanofibers as a high-rate and long-cycle anode material for Na ion batteries [18]. Shaojun Guo from Peking University presents Co-doped 1T-phase MoS₂ nanosheets embedded in N, S co-doped carbon nanobowls showing high rate performance and cycling stability for Na ion batteries [19]. Qiang Zhang from Tsinghua University reports a sandwich-structured Li metal electrode with improved Li plating and stripping behavior [20], suitable for high-energy rechargeable batteries. Guanglei Cui from Qingdao Institute of Bioenergy and Bioprocess Technology, Chinese Academy of Sciences has developed a flame-retardant quasi-solid polymer electrolyte for rechargeable Na metal batteries [21]. Ying Shirley Meng from University of California, San Diego reports a polyol method to synthesize layer-structured oxide, spinel-structured oxide and olivine-structured phosphate nanoparticles as cathode materials for Li ion batteries [22].

In the area of electrocatalysis, Yanguang Li from Soochow

University introduces a solvent-free nanocasting method to prepare various ordered mesoporous metal sulfide/carbon composite materials with improved electrocatalytic performance for the H₂ evolution reaction [23]. Liangbing Hu from University of Maryland reports a millisecond-scale fast synthesis method to prepare CoS/graphene composite materials for water electrolysis [24]. Jinlong Gong from Tianjin University demonstrates the synthesis of Pt-Pd bilayer nanocages improving the catalytic activity for H₂ evolution [25]. Karl J. J. Mayrhofer from Max-Planck-Institut für Eisenforschung GmbH describes a liquid atomic layer deposition method toward maximized utilization of iridium for catalyzing electrochemical water oxidation to O₂ in acidic solutions [26]. Zhangquan Peng from Changchun Institute of Applied Chemistry, Chinese Academy of Sciences reports a mesoporous ternary Co-Fe-Ni oxide material with a hierarchical bimodal channel structure as both the anode catalyst and the cathode catalyst for water electrolysis [27]. Shannon W. Boettcher from University of Oregon reports composition-dependent electronic structures and catalytic activities of ternary Ni-Co-Fe oxyhydroxide materials as electrocatalysts for the O₂ evolution reaction [28]. Changzheng Wu from University of Science and Technology of China reports the development of an active O₂ evolution electrocatalyst based on an oxygen-deficient perovskite-structured Sr-Co oxide [29]. Xiaoming Sun from Beijing University of Chemical Technology has invented a new approach of using cationic surfactants to construct superaerophobic nanoarray electrodes with enhanced catalytic performance for water oxidation [30]. Jin Suntivich from Cornell University presents the substitutional effects of 3d transition metals in molybdenum/tungsten nitride for electrocatalytic O₂ reduction in acid [31].

Dehui Deng from Dalian Institute of Chemical Physics, Chinese Academy of Sciences describes single Fe atoms confined in a multiscale carbon foam exhibiting high activity for electrocatalytic reduction of CO₂ to CO [32]. Xinbo Zhang from Changchun Institute of Applied Chemistry, Chinese Academy of Sciences reports an Fe-N-C electrocatalyst material that can convert CO₂ and H₂O to CO/H₂ mixtures of different ratios [33]. Gengfeng Zheng from Fudan University reports that increased N doping in MOF-derived mesoporous carbon leads to enhanced electrocatalytic activity for CO₂ reduction to CO [34]. Yongye Liang from Southern University of Science and Technology emphasizes the importance of dispersing metal phthalocyanine molecules on the surface of highly-conductive carbon nanotubes to exposing the intrinsic electrocatalytic properties of these molecular materials for CO₂ reduction reactions [35].

Thermal catalysis and relevant materials are also covered. Xiaoqing Huang from Soochow University reports the tuning of surface electronic structure of Pt-Pb bimetallic nanocrystals to enhance their catalytic performance for the ethanol reforming reaction to produce clean H₂ fuel [36]. Ding Ma from Peking University has developed a MOF-derived Fe-based catalyst for efficient conversion of N₂ and H₂ to ammonia [37]. Qiang Fu from Dalian Institute of Chemical Physics, Chinese Academy of Sciences reports controlled growth of uniform monolayer and bilayer ZnO nanostructures on Au(111) surface and their hydroxylation under catalysis-relevant conditions [38]. Jovana Zečević from Utrecht University has studied the structural stability of a series of oxide nanoparticles using liquid phase transmission electron microscopy and demonstrates a correlation between stability and Gibbs free energy of hydration [39]. Aleksandra Vojvodic from University of Pennsylvania presents a theory and experiment combined study of the structure and distribution of Fe dopants in two-dimensional (2D) CoO nanoislands on Au(111) as a function of the Fe/Co ratio [40]. Kristie J. Koski from University of California, Davis reports pressure-induced semiconductor-to-metal phase transition behavior for 2D layered silicon telluride nanoplates [41].

On solar energy conversion, Dunwei Wang from Boston College

reports the influences of Pt catalyst loading on the charge transfer and recombination processes at the interface of a GaN-protected Si photoanode [42]. Tierui Zhang from Technical Institute of Physics and Chemistry, Chinese Academy of Sciences has developed a polymer-templated method to synthesize holey graphitic carbon nitride nanotubes with enhanced photocatalytic activity for H₂ production under visible-light irradiation [43]. Robert Baker from the Ohio State University reports on the influences of Cu vacancies, interstitial O atoms, and phase impurities on the electron dynamics and photochemistry of Cu-Fe oxide photocathodes [44]. Wei You from University of North Carolina at Chapel Hill reports their discovery that more fluorination on both the donor polymer and the non-fullerene acceptor leads to higher overall efficiency for the solar cell [45].

Overall, this issue highlights a diverse range of topics on broadly-defined nanomaterials for energy conversion and storage. We sincerely thank all the awardees for their excellent contributions to this 2019 NR45 Special Issue. We hope that you will enjoy this special issue, and encourage you and your colleagues to nominate outstanding young innovators in your field for the following years' NR45 awards.

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