REVIEW PAPER



The Structural and Magnetic Properties of $\text{CuRh}_{1-x}\text{Mn}_x\text{O}_2$ ($0 \le x \le 0.1$) Delafossite Oxide

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Abstract The CuRh_{1-x}Mn_xO₂ (0.00 $\leq x \leq 0.10$) polycrystalline powders were synthesized by the direct solidstate reaction. The magnetic susceptibility was measured in the temperature range of 4-300 K. It was found that the magnetic susceptibility (χ) increases rapidly with the doping of Rh³⁺ by Mn³⁺ ions with existence of paramagnetic substance at low temperature. For x = 0.1, a development of new antiferromagnetic (AF) transition around 26 K and the magnetic data for $CuRh_{1-x}Mn_xO_2$ show evidence for weak ferromagnetic (FM) transition between 100 and 130 K. Clear hysteresis loops indicate that FM order exists in Mn-doped samples at 5 K. All samples behave like semiconductors. The ferromagnetism properties can be attributed to the double-exchange interaction between the Mn³⁺ and Rh³⁺ semiconductor compounds. We analyze the magnetic entropy change $\Delta S_{\rm M}$ of CuRh_{0.9}Mn_{0.1}O₂. Using Arrott plots, it was found that the phase transition for this sample at $T_N = 42$ K is of second order. The maximum entropy change of ~ 0.017 J/kg K upon a filed variation of \sim 1 T and \sim 0.076 J/kg K upon a filed variation of \sim 5

⊠ Taieb Elkhouni elkhouni@yahoo.fr T was obtained, indicating relatively small magnetic order fluctuations around T_N .

Keywords Delafossite · Powder diffraction · Magnetic susceptibility · Antiferromagnetic transition · Magnetic entropy

1 Introduction

In recent years, there has been renewed interest in geometrically frustrated magnetic system. In such systems, spin ordering is suppressed by competing exchange interactions well below the conventional ordering scale set by the Weiss temperature. Consequently, they are expected to possess unconventional magnetic states, such as spin-liquid [1] and spin-nematic states [2]. ABO₂-type materials, which have two-dimensional (2D) layered structures consisting of triangular lattices made up of magnetic B-site trivalent cations, have been studied as typical frustrated triangular-lattice spin systems [3]. In particular, $CuFeO_2$ and $CuCrO_2$ have been extensively studied in terms of magnetoelastic coupling and magnetoelectric multiferroics, which are considered to be associated with, spin frustration [4–7]. In other hands, geometrically frustrated magnetic systems have received a great deal of attention because their ordered or unordered ground states are often very exotic [1-3]. The triangular lattice antiferromagnet (TLA), in which magnetic ions reside at a triangular net, is the most obvious example of a geometrically frustrated magnetic system. A typical ordered structure in TLA is a noncollinear 3-sublattice 1200 spin configurations where the frustration of the three nearest-neighbor spins on a triangular plaquette is resolved by a 120° rotation

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of neighboring spins. The ABO₂ family with the delafossite structure (A = monovalent ion, B = trivalent ion) has been recently investigated as one of the typical materials for the TLA.

In order to search for new thermoelectric materials, $CuRh_{0.90}Mg_{0.10}O_2$ [8] has been investigated. One advantage of these materials over several others lies in their ability to be used at elevated temperatures in air. This opens the route to target systems releasing heat at temperatures as high as 1000 K. Among the studied p-type thermoelectric oxides, the layered ones such as CuRhO₂, or more recently the delafossites, all with structures containing CdI2-type layers have been particularly investigated according to the richness of their physical properties. For instance, the thermoelectric performance of several doped CuRhO₂ delafossites have been measured, leading to the following dimensionless figures of merit ZT of 0.04 at 800 K (CuCr_{0.97}Mg_{0.03}O₂) [8], 0.14 at 1100 K (CuFe_{0.99}Ni_{0.01}O₂) [9], and 0.15 at 1000 K (CuRh_{0.90}Mg_{0.10}O₂) [8]. Their crystal structure of the mother CuRhO₂ can all be described as a delafossite-type oxide belongs to the space group R-3m. It has lattice parameters a = 3.0824 Å and c = 17.0943 Å [10]. It consists of a triangular lattice of Rh^{3+} ions (S = 0, L = 0) separated by nonmagnetic ionic layers of O^{2-} , Cu^+ , and O^{2-} , stacked along the *c* axis in the hexagonal description [11]. Among other magnetic delafossite, crednerite CuMnO₂ is an ABO2-type triangular-lattice antiferromagnet that has Bsite cations of Mn³⁺ (3d⁴: t_{2g}^3 e¹_g), the crystal structure of CuMnO₂ does not consist of perfect triangular lattices [12], and rather, it consists of isosceles-triangular lattices due to Jahn-Teller distortion of Mn^{3+} (3d⁴: t_{2g}^3 e¹_g) with an orbital degree of freedom. The crystal structure of CuMnO₂ belongs to the monoclinic space group C2/m at room temperature and it has lattice constants of a = 5.596 Å, c =5.899 Å, and $\beta = 104.02^{\circ}$ [12]. Doumerc et al. [13] performed magnetic susceptibility measurements of CuMnO₂; the electronic configuration of Mn³⁺ was found to be the high spin state (S = 2) and Weiss temperature Θ is -450 K [13]. They also used Mossbauer spectroscopy to investigate Fe-doped samples and found a 3D long-range magnetic ordering which exhibits the antiferromagnetic phase transition at relatively low temperature $T_N = 64$ K. The further cooling produces the week ferromagnetic phase transition at 42 K, which is reported by the previous bulk measurements [13]. In the intermediate temperature regions of T_N $< T < \Theta$, the susceptibility shows broad peaks reflecting the low-dimensional character. So far, the microscopic magnetic correlations and the dimensionality have not been clarified.

In this paper, we discuss another intriguing aspect, namely multiferroic nature, of geometrical spin frustration in a TLA. Recently, there has been a revival of interest in research of multiferroics showing the coexistence and/or interplay of magnetism and ferroelectricity socalled magnetoelectric (ME) effects [14, 15]. Among multiferroics studied to date, some exhibit strong couplings between magnetism and ferroelectricity, since the geometrical frustration in the magnetic triangular sublattice at the M sites causes intriguing properties, such as field induced multistep magnetization change [16] and multiferroics [17]. The introduction of charge carriers, which is possible in some delafossite compounds, may further induce novel phenomena arising from the coupling between the magnetic and transport properties in the quasi-two-dimensional frustrated triangular lattice. The magnetic degeneracy was lifted and the no collinear magnetic structure was realized in the antiferromagnetic (AF) regime below the ordering temperature (T_N) to resolve the magnetic frustration [18, 19]. This magnetic frustration appeared to be removed by the introduction of some trivalent cations in M-sites. In this study, to clarify in depth such a unique Mn-substitution effect, we investigated in detail the effect of the substitution of Mn^{3+} , for Rh³⁺ on their magnetic properties of CuRhO₂

2 Experimental Section

Polycrystalline samples of CuRh_{1-x}Mn_xO₂ for different composition around 10 % were prepared using the standard solid-state reaction. Stoichiometric mixtures (0.5 g) of Cu₂O, Rh₂O₃, and Mn₂O₃ were ground and pressed in pellets. The samples were fired repeatedly in air at 1150 oC for 12 h in alumina crucibles with intermediate re-grinding and pelletizing. X-ray powder diffraction patterns of the products were collected with a PANalytical diffractometer equipped with a CuK_a source (K_{α 1} and K_{α 2}) in the 2 θ range from 10° to 90° at room temperature.

Magnetic susceptibilities measurements vs. temperature were carried out in a Quantum Design superconducting quantum interference device (SQUID) magnetometer in the range 4–300 K. The typical field used for temperature dependence measurements was 0.1 T. Magnetization measurements M vs. H used for MCE studies were realized with the BS2 magnetometer developed at Néel Institute.

3 Results and Discussion

3.1 Structural Properties

The crystal structure of $\text{CuCr}_{1-x}\text{Mn}_x\text{O}_2$ ($0 \le x \le 0.1$) samples is determined by recording X-ray diffraction patterns





shown in Fig. 1. According to the standard XRD patterns of CuRhO₂, samples are of single phase with delafossite structure (space group R-3m), without any detectable secondary phase up to 10 % Mn content. For this composition, the occurrence of structural transformation is clearly demonstrated in the inset of Fig. 1 by the splitting of the (012) Bragg peak into two components which are identified by a decrease of the parent rhombohedral phase and a development of new crednerite CuMnO₂ phase with increasing of Mn content. In fact, we conclude that the delafossite CuRhO₂ tend to be Crednerite CuMnO₂ at 10 % of Mn substitution.

Similar to previous observation by Damy et al. [20], the room temperature of crystal structure of $\operatorname{CuRh}_{1-x}\operatorname{Mn}_x\operatorname{O}_2$ $0.00 \le x \le 0.1$ compounds was found to be rhombohedral with a slight decreases of the *a* lattice constant. This small compression of the oxygen framework can be accommodated in a psedomonoclinic C2/m symmetry, which maintains a similar description of the high-content crystal lattice. The Rh/MnO₆ octahedra undergo a small compression in its equatorial plane, which produces a decrease of Rh-Mn bond with increase of x content.

Composition dependences of the unit cell volume and lattice constants are given in Table 1. For x = 0 (pure CuRhO₂), our *a* and *c* parameter values are in very good agreement with most of those previously reported for powders

as well as for single crystal (e.g. a = 3.082(3) Å, c = 17.094(3) Å from [10]).

The decreases of the unit cell volume with increasing x well agrees with the decrease of the ionic radius as rhodium (r = 0.66 Å) is substituted for manganese (r = 0.58 Å) [21]. However, the change is rather anisotropic as it is mainly due to the shrinking of the *a*parameter whereas the *c*parameter remains more or less constant.

As the Cu–O distance does not vary much in delafossite, this tendency reflects a flattening of the MO_6 octahedra. A nice geometric account of this evolution is given by J. Tate et al. [22] where the authors suggest that the strong repulsion between M^{3+} ions across the octahedron shared edges reduces the O–O distance to the contact distance.

Therefore, the decreases of the size of M cation leads to a decrease of the octahedron distortion and in turn of the

Table 1 The lattice parameters *a*, *c*, and cell volumes *V* of the $\text{CuRh}_{1-x}\text{Mn}_x\text{O}_2$ (0.00 $\le x \le 0.10$) compounds

Samples	a (Å)	<i>c</i> (Å)	V (Å ³)
CuRhO ₂	3.082(3)	17.094(3)	140.656(5)
CuRh _{0.98} Mn _{0.02} O ₂	3.079(6)	17.093(1)	140.391(2)
CuRh _{0.95} Mn _{0.05} O ₂	3.075(1)	17.092(8)	139.978(8)
CuRh _{0.90} Mn _{0.10} O ₂	3.059(3)	17.092(5)	138.541(6)

M–M distance that corresponds to the *a*parameter. We may recall that in CuMO₂, as M changes from La^{3+} to Al^{3+} [23, 24], the *a*parameter undergoes a huge decreases from 3.8 up to 2.8 Å.

3.2 Magnetic Properties

Figure 2 shows the temperature dependence of magnetic susceptibility of $\operatorname{CuRh}_{1-x}\operatorname{Mn}_x\operatorname{O}_2(0.00 \le x \le 0.10)$ samples at 0.1 T. The $\chi(T)$ curves of the parent delafossite CuRhO₂, a paramagnetic comportment in the temperature range 4 to 300 K, for the content x = 0.02 and x = 0.05 the magnetic susceptibility present a short increases of the magnetic susceptibilities with increases of x content at height temperature. The $\chi(T)$ curve of the CuRh_{0.9}Mn_{0.10}O₂ sample showed a sharp transition at $T_N = 42$ K from a paramagnetic to antiferromagnetic state followed by an increases of the magnetic susceptibility at height temperature.

This behavior of the $\chi(T)$ curves for CuRh_{1-x}Mn_xO₂ series at height temperature region agrees with results shown in [10]. In fact, in this reference, it has been reported that the increase in the magnetic susceptibility in $CuRh_{1-x}Co_xO_2$ is significant only with increasing of x content.

The replacement of Rh³⁺ by Mn³⁺ develop some ratios of Mn^{3+} – O^{2-} – Mn^{4+} bonds and the interactions between Rh-O-Mn might be an antiferromagnetic super-exchange. The competition between the Mn–O–Mn bonds are doubleexchange (ferromagnetic) and Rh-O-Mn super-exchange (antiferromagnetic) may explain the weakening increase of the magnetic susceptibility and the development of the ferromagnetism with appears of new crednerite phase CuMnO₂ in our substituted sample x = 0.1. All those results are in a good agreement with those reported by S. Hebert et al. [25] which reported that the substitution in Mnsites of the mother compound LaMnO3 induce disorder on the manganese site, and prevent the establishment of conducting paths along the Mn–O–Mn bonds. Conversely, the introduction of trivalent cations does not induce a strong ferromagnetism and the electronic transport is obviously hindered by these foreign cations on the manganese site. This supposition explain the strong ferromagnetic transition in LaMnO3 which is induced by introducing mixed-valent manganese Mn³⁺/Mn⁴⁺ and they induce disorder on the Mn-sites. These phenomena explain the variation of the amplitude transition ferro-paramagnetic with the temperature.

The inverse of the susceptibility, χ^{-1} , versus temperature curves are plotted in inset of Fig. 2. Those plots can be described in the paramagnetic region by a Curie-Weiss law, defined by the relation $\chi = \chi_0 + C/(T - \Theta)$ [10], where C is the Curie constant and Θ is the asymptotic Curie temperature. By fitting the linear paramagnetic region, the Curie-Weiss C and Θ parameters were obtained (Table 2). The negative values of Θ indicate the strong antiferromagnetic interactions resulting from the Mn substitution. Moreover, the increase of Θ value and the convergence of those values to positive prove that those interactions converge to ferromagnetic state between spin. Correlating the magnetic properties in $CuRh_{1-x}Mn_xO_2$ system with those



susceptibility

Table 2 Estimated Curie constant Curie temperature and effective magnetic moment from the hightemperature paramagnetic region for CuRh _{1-x} Mn _x O ₂ ($0.00 \le x \le 0.10$)	Samples	CuRhO ₂	2 %	5 %	10 %
	$\chi_0 \text{ (emu mol}^{-1} \text{ Oe}^{-1}\text{)}$	$1.6E^{-3}$	$4.6E^{-3}$	$9.2E^{-4}$	$11.2 \mathrm{E}^{-4}$
	C (emu K mol ⁻¹)	0.0048(4)	0.0068(8)	0.0084(3)	0.0102(7)
	$\Theta(\mathbf{K})$	-81.6(4)	-75.3(2)	-52.5(1)	-27.1(7)
	$\mu_{\rm eff}(\mu_{\rm B})$	0.1959(6)	0.2346(1)	0.2596(9)	0.2866(3)

of structural properties. The origin of increasing magnetic susceptibility is proved by the development of new phase which is deconvoluted by the development of new crednerite CuMnO₂phase. From the magnetic properties, the tendency to the ferromagnetic state proves those structural propositions.

In order to investigate the magnetic behavior of the system $\operatorname{CuRh}_{1-x}\operatorname{Mn}_x\operatorname{O}_2$, we have carried out magnetic susceptibilities versus the magnetic applied field at height temperature ($\chi(H)$ at 5 and 300 K) as shown in Fig. 3a, b, respectively. We can distinguish two different types of the χ -H curves as a function of manganese content. For all samples, we observe a typical behavior of antiferromagnetic states, when the magnetic susceptibilities at low temperatures (5 K) increase sharply and then saturates as shown in Fig. 3a. This antiferromagnetic results show that the Mn substitutions strongly weaken the antiferromagnetic and stabilize this phase.

The height temperature (300 K) χ -*H* curves for CuRh_{0.9}Mn_{0.1}O₂ ceramic is presented in Fig. 3b. For a closer observation, the curve is magnified partially. It is seen that the χ -*H* loop apparently shows hysteresis, indicating ferromagnetism of the Mn-doped CuRhO₂ sample. Also, a linear paramagnetic behavior superimposed upon hysteresis is observed, which may originate from paramagnetic regions contained in the samples due to nearly isolated Mn ions, e.g., where the local Mn concentration is the lowest, this results is in a good agreement with those reported by C. Gao et al. [26] for Fe substituted CuCrO₂.

The disorder of the Rh and Mn cations complicates the interpretation of the magnetic properties. The interaction of the M cations with each other through Rh–O–Mn linkages of approximately 180° is expected to be the dominant consideration, and this would be antiferromagnetic in nature. Recently, it was reported on evidence of room temperature ferromagnetism in Co-doped transparent CuAlO₂ semiconductor [27]. Specially, the coercively (Hc) and saturation magnetization are significantly enhanced with Co composition [27]. In Fig. 3b and at 300 K, a remarkable loop for CuRh_{0.9}Mn_{0.1}O₂ is observed, giving rise such us to a coercivity of 4000 Oe and remnant magnetization of ~ 1.3 10^{-4} emu mol⁻¹, although no magnetization saturation is obtained, probably due to the spin-glasslike essence. Therefore, the substantial ferromagnetic crednerite component CuMnO₂ in the magnetism of CuRh_{1-x}Mn_xO₂ is revealed.

The observed magnetic behavior for $\text{CuRh}_{1-x}\text{Mn}_x\text{O}_2$ is probably due to the disorder of Rh and Mn in octahedral sites resulting in short-range Rh–O–Rh and Mn–O–Mn interactions between which could give rise to antiferromagnetism coupled with short-range weak ferromagnetism. Thus, the observed ferromagnetism is essentially intrinsic to some extent. The origin of ferromagnetism in DMSs is still not very clear, although many efforts have been made to explain it. It is believed from the calculated density of states (DOS) that the double-exchange interaction is the dominant exchange mechanism in TM-doped CuAlO₂based DMSs [28]. It is also found that indirect interaction among Co centers results in ferromagnetism, whereas direct interaction among them induces antiferromagnetism [29, 30].

3.3 Magnetic Entropy

In order to investigate the change in magnetic entropy, isothermal magnetization curves are recorded around T_N for the sample CuRh_{0.9}Mn_{0.1}O₂ in a magnetic field up to 5 T. Figure 4a shows the measured isothermal magnetization for the considered sample in the form of polynomial fitting of H/M vs. M^2 Arrott plots. A more standard type of Arrott plots is observed. The plots display the nonlinear behavior, and the positive slope is clearly seen in the complete M^2 range.

According to the criterion proposed by Banerjee [31], all the properties of the Arrott plots show characteristics of second-order magnetic phase transition. The magnetocaloric effect is calculated indirectly with Maxwell's relation from the magnetization data in various applied magnetic field [32] from 0 to $\mu_0 H_{\text{max}}$ as follows:

$$\Delta S_M(T, \Delta(\mu_0 H)) = S(T, \mu_0 H_1) = \int_{\mu_0 H_1}^{\mu_0 H_2} \left(\frac{\delta M(T, \mu_0 H)}{\delta T}\right) d(\mu_0 H)$$
(1)

It could be written as follows:

$$\Delta S_M\left(\frac{T_1+T_2}{2}\right) = \frac{1}{T_2-T_1} \left[\int_0^{\mu_0 H} M(T_2,\mu_0 H) d(\mu_0 H) - \int_0^{\mu_0 H} M(T_1,\mu_0 H) d(\mu_0 H) \right]$$
(2)

where $(\partial M/\partial T)_{\mu_0 H_i}$ is the experimental value obtained from M-T curve in a magnetic field $\mu_0 H_i$.

Figure 4b presents the magnetic entropy change ΔS_M for the sample CuRh_{0.9}Mn_{0.1}O₂ as a function of temperature under different magnetic fields. A remarkable feature is the exact correspondence between the maximal ΔS_M and the recorded PM–AFM transition at T_N , while ΔS_M distributes over a relatively wide T range. The maximum entropy change of ~ 0.017 J/kg K upon a filed variation of ~ 1 T and ~ 0.076 J/kg K upon a filed variation of ~ 5 T was obtained, indicating relatively small magnetic order fluctuations around T_N . This result proves the idea of existence of weak antiferromagnetic interaction at low temperatures.

4 Discussion

The discussion above comes to a conclusion that the observed ferromagnetic behavior should be an intrinsic property of Mn-substituted CuRhO₂ solid reaction. According to refs. [33–36], the Mn³⁺ subsisting can destabilize the anti-ferromagnetic order of Rh³⁺ ions, modulate the spin configuration, and bring about ferromagnetism. Although the super-exchange interaction between Mn³⁺ is also anti-ferromagnetic [37], the coexistence of Mn³⁺ and Rh³⁺ should induce a new magnetic exchange mechanism. The single valence state of Mn³⁺ and Rh³⁺, revealed by the XRD results, excludes the possibility of double-exchange



Fig. 4 H/M vs. M^2 (a) isotherms for CuRh_{0.9}Mn_{0.1}O₂. The temperatures of the isotherms are indicated. Magnetic entropy change $(-\Delta S_M)$ (b) as a function of temperature in various magnetic fields up to 5 T



mechanism. Hence, the hole-mediated $Mn^{3+}-Rh^{3+}$ superexchange interaction can be expected to produce ferromagnetism which overcomes the anti-ferromagnetic $Rh^{3+}-Rh^{3+}$ and $Mn^{3+}-Mn^{3+}$ couplings. With the increasing Mn concentration, both the number of the $Mn^{3+}-Rh^{3+}$ pairs and the hole concentration are enhanced. Accordingly, the hole-mediated $Mn^{3+}-Rh^{3+}$ ferromagnetic coupling is strengthened, leading to the gradual increase in both M_S and Θ .

The χ -*T* curves of CuRh_{1-x}Mn_xO₂ ceramics were measured by SQUID at a magnetic field of 0.1 T. The entire Mn-doped samples exhibit greatly enhanced the magnetic effective moment ($\mu_{eff}(\mu_B)$), compared with literature values. With the change of Mn concentration, the $\mu_{eff}(\mu_B)$

values range between ~ 0.196 $\mu_{\rm B}$ and ~ 0.286 $\mu_{\rm B}$. Its $\mu_{\rm eff}(\mu_{\rm B})$ can be evaluated as much less than those of CuCr_{1-x}Mn_xO₂ ceramics [38] ($x \ge 0$). The χ -*T* results also provide strong evidence that the ferromagnetism cannot come from delafossite CuRh_{1-x}Mn_xO₂ but it is attributed to the development of new crednerite phase CuMnO₂ which is identified by XRD pattern. This ferromagnetic impurity were present; it could not account for the Mn concentration dependence of magnetization. Note that increasing the Mn concentration leads to a gradual decrease in M_S . If manganese oxides were responsible for ferromagnetic behavior, the increase of Mn concentration would presumably enhance the corresponding phase volume fraction and related magnetization signature. Instead, the opposite behavior is observed. Therefore, the ferromagnetism is considered to be an intrinsic property of Mndoped CuRhO₂ ceramics. According to ref. [38], Mn-doped CuCrO₂ can destabilize the anti-ferromagnetic order of Cr^{3+} ions, whereas the super-exchange interaction between Rh³⁺ is also known to be anti-ferromagnetic [39]. Nevertheless, the coexistence of Mn³⁺ and Rh³⁺ should induce different magnetic exchange interaction. Z. Yang et al. [40] supposed that the magnetic interactions between Mn³⁺ and Mn³⁺ ions ⁺ are ferromagnetic; the magnetic interactions between Rh³⁺ and Rh³ ions are antiferromagnetic.

On the other hand, F. Dammy et al. [20] reported that the substitution in Mn-sites of the mother compound CuMnO₂ induce disorder on the manganese site, and prevent the establishment of conducting paths along the Mn–O–Mn bonds. Conversely, the introduction of trivalent cations does not induce a strong ferromagnetism and the electronic transport is obviously hindered by these foreign cations on the manganese site. This supposition explain the strong ferromagnetic transition in CuMnO₂ which is induced by introducing mixed-valent manganese Mn³⁺/Mn⁴⁺ and they induce disorder on the Mn-sites. This mixture valence state of Mn³⁺/Mn⁴⁺ certified the possibility of double-exchange mechanism inducing three-dimensional ferromagnetic interactions in CuMnO₂ compounds and shows a weak ferromagnetism.

Hence, the hole-mediated Mn^{3+} – Rh^{3+} super-exchange interaction can be expected to produce ferromagnetism of Mn-doped CuRhO₂. The ferromagnetic Mn^{3+}/Mn^{4+} – Rh^{3+} coupling dominates over the crednerite phase CuMnO₂, giving rise to the ferromagnetic behavior and enhanced Θ .

As increasing the Mn concentration, both interactions Mn³⁺-Rh³⁺ and Mn⁴⁺- Rh³⁺ pairs are increased while the Rh³⁺-Rh³⁺ pairs are decreased. It means that ferromagnetic couplings would be strengthened whereas antiferromagnetic Rh³⁺–Rh³⁺ coupling just the reverse. Meanwhile, the magnetization has a high sensitivity to small changes in lattice constants [38]. For this $CuRh_{1-x}Mn_xO_2$ structure, a decrease of thea parameter (see Table 1) corresponds to short Rh-Mn distances, which will influence the Rh-Mn magnetic interactions. Besides, the hole concentration may change with the Mn addition, which would also have an effect on the hole-mediated exchange interactions. The competitions among these three factors, that is, the number of the Rh-Mn pairs, the Rh-Mn distances, and the hole concentration, eventually result in an almost linear reduction of M_S with the increasing Mn concentration.

5 Conclusion

In summary, we investigated the structural, magnetic properties and magnetocaloric effect of $\text{CuRh}_{1-x}\text{Mn}_x\text{O}_2$ (0 \leq $x \leq 0.1$). The structure of CuRh_{1-x}Mn_xO₂ was delafossite phase for the studied x range and a decrease of the lattice parameters a among the samples. CuRhO₂ is paramagnetic oxide; the magnetic susceptibilities increased with increasing Mn content, up x = 0.1 which develops with new antiferromagnetic transition temperature $T_N = 42$ K. The magnetization has a high sensitivity to small changes in lattice constants; for this $CuRh_{1-x}Mn_xO_2$ structure, a decrease of the *a* parameter corresponds to decreases of Rh-Mn distances, which will increase the Rh-Mn magnetic interactions to develop this new antiferromagnetic transition $T_N =$ 42 K indicating partial release of the spin frustration. The magnetocaloric effect for the sample CuRh_{0.9}Mn_{0.1}O₂has been investigated around T_N in a magnetic field up to 5 T. A maximum entropy ΔS_{max} was obtained; it changed with ~ 0.017 J/kg K upon a filed variation of ~ 1 T and ~ 0.076 J/kg K upon a filed variation of ~ 5 T, indicating relatively small magnetic order fluctuations around T_N .

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References

- 1. Balents, L.: Nature (London) 464, 199 (2010)
- 2. Chandra, P., Coleman, P.: Phys. Rev. Lett. 64, 88 (1990)
- 3. Collins, M.F., Petrenko, O.A.: Can. J. Phys. 75, 605 (1997)
- Terada, N., Narumi, Y., Sawai, Y., Katsumata, K., Staub, U., Tananka, Y., Kikkawa, A., Fukui, T., Kindo, K., Yamamoto, T., Kanmuri, R., Hagiwara, M., Toyokawa, H., Ishikawa, T., Kitamura, H.: Phys. Rev. B 75, 224411 (2007)
- 5. Kumura, T., Lashley, J.C., Tokura, Y.: Phys. Rev. Lett. 101, 220401 (2006)
- Seki, S., Onose, Y., Tokura, Y.: Phys. Rev. Lett. 101, 067204 (2008)
- Kimura, K., Otani, T., Nakamura, H., Wakabayashi, Y., Kimura, T.: J. Phys. Soc. Jpn. 78, 113710 (2009)
- 8. Kawamura, H.: J. Phys.: Condens. Matter 10, 4707 (1998)
- 9. Collins, M.F., Petrenko, O.A.: Can. J. Phys. 75, 605 (1997)
- Elkhouni, T., Amami, M., Hlil, E.K., Ben Salah, A.: J. Supercond. Nov. Magn. 8, 2182 (2013)
- Galakhov, V.R., Poteryaev, A.I., Kurmaev, E.Z., Anisimov, V.I., Bartkowski, S., Neumann, M., Lu, Z.W., Klein, B.M., Zhao, T.-R.: Phys. Rev. B 56, 4584 (1997)
- Topfer, J., Trari, M., Gravereau, P., Chaminade, J.P., Doumerc, J.P.: Z. Kristallogr. 210, 184 (1995)
- Doumerc, J.P., Trari, M., Topfer, J., Founes, L., Grenier, J.C., Pouchard, M., Hagenmuller, P.: Eur. J. Solid. Inorg. Chem. 31, 705 (1994)
- 14. Hill, N.A.: J. Phys. Chem. B 104, 6694 (2000)
- 15. Fiebig, M.: J. Phys. D: Appl. Phys. 38, R123 (2005)

- Terada, N., Nakamura, Y., Katsumata, K., Yamamoto, T., Staub, U., Kindo, K., Hagiwara, M., Tanaka, Y., Kikkawa, A., Toyama, H., Fuyuki, T., Kanmuri, R., Ishikawa, T., Kitamura, H.: J. Phys. Rev. B 74, 180315 (2006)
- 17. Kimura, T., Lashley, J., Ramirez, A.: vol. 74 (2006)
- Kadowaki, H., Kikuchi, H., Ajiro, Y.: J. Phys. Condens. Matter 2, 4485 (1990)
- Kajimoto, R., Matsuda, M., Okuda, T.: J. JAEA-Review 75, 004 (2007)
- Damay, F., Poienar, M., Maignan, A., Rodriguez-Carvajal, J., André, G., Doumerc, J.P.: J. Phys. Rev. B 80, 094410 (2009)
- 21. Shannon, R.D., Rogers, D.B., Prewitt, C.T.: J. Inorg. Chem. 10, 713 (1971)
- Tate, J., Jayaraj, M., Draeseke, A., Ulbrich, T., Sleight, A., Vanaja, K., Nagarajan, R., Wagner, J., Hoffman, R.: Thin Solid Films 411, 119 (2002)
- 23. Amami, M., Colin, C.V., Strobel, P., Ben Salah, A.: Physica B 406, 2182 (2011)
- Garlea, O., Darie, C., Bougerol, C., Isnard, O., Bordet, P.: J. Sol. St. Sci. 5, 1095–1104 (2003)
- Hebert, S., Martin, C., Maignan, A., Retoux, R., Hervieu, M., Nguyen, N., Raveau, B.: J. Phys. Rev. B 65, 104420 (2002)
- Gao, C., Lin, F., Zhou, X., Shi, W., Liu, A.: J. Alloys Comp. 565, 154–158 (2013)
- Dong, C.J., Yu, W.X., Xu, M., Cao, J.J., Zhang, Y., Chuai, Y.H., Wang, Y.D.: J. All. Comp. **512**, 195 (2012)

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- Kizaki, H., Sato, K., Yanase, A., Katayama-Yoshida, H.: Jpn. J. Appl. Phys. 44, 1187 (2005)
- Dietl, T., Ohno, H., Matsukura, F., Cibert, J., Ferrand, D.: Science 287, 1019 (2000)
- 30. Coey, J.M.D., Venkatesan, M., Fitzgerald, C.B.: Nat. Mater. 4, 173 (2005)
- 31. Banerjee, B.K.: Phys. Lett. 12, 16 (1964)
- Omri, A., Tozri, A., Bejar, M., Dhahri, E., Hlil, E.K.: J. Mag. Mag. Mat. 324, 3122–3128 (2012)
- Luo, S.J., Li, L., Wang, K.F., Li, S.Z., Dong, X.W., Yan, Z.B., Liu, J.-M.: Thin Solid Films 518, e50 (2010)
- Khlifi, M., Tozri, A., Bejar, M., Dhahri, E., Hlil, E.K.: J. Mag. Mag. Mat. 324, 2142–2146 (2012)
- Jlaiel, F., Amami, M., Boudjada, N., Strobel, P., Ben Salah, A.: J. Alloys Comp. **509**, 7784 (2011)
- Elkhouni, T., Amami, M., Colin, C.V., Strobel, P., Ben Salah, A.: J. Magn. Magn. Mater. 330, 101 (2013)
- Matsui, T., Taketani, E., Fujimura, N., Ito, T., Morii, K.: J. Appl. Phys. 93, 6993 (2003)
- Elkhouni, T., Amami, M., Colin, C.V., Strobel, P., BenSalah, A.: J. Mag. Mag. Mat. 355, 158–163 (2014)
- Okuda, T., Kishimoto, T., Uto, K., Hokazono, T., Onose, Y., Tokura, Y., Kajimoto, R., Matsuda, M.: J. Phys. Soc. Jpn. 78, 013604 (2009)
- Yang, Z., Ye, L., Xie, X.: J. Phys.: Condens. Matter 12, 2737– 2747 (2000)