

Synergetic effect of LaB₆ and ITO nanoparticles on optical properties and thermal stability of poly(vinylbutyral) nanocomposite films

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Abstract In this work, different compositions of lanthanum hexaboride (LaB₆) and tin-doped indium oxide (ITO) nanoparticles were doped into poly(vinylbutyral) (PVB) matrix to prepare PVB/LaB₆-ITO nanocomposite (PLINC) films by a solution casting method. X-ray diffraction, Fourier transform infrared spectroscopy, field emission scanning electron microscopy, thermogravimetric analysis (TGA) and ultraviolet-visible-near infrared spectroscopy (UV-vis-NIR) were employed to characterize the PLINCs. The TGA and UV-vis-NIR results reveal that the nanocomposite films possessed outstanding thermal stability. The temperature where 5 % weight loss of the PVB matrix was improved after the addition of LaB₆ and ITO particles and the property for blocking near infrared light was also enhanced as compared with the case of pure PVB film.

1 Introduction

There has been growing demand in recent years to filter out the near infrared (NIR) waves of sunlight that penetrate windows of buildings and vehicles, in accordance with worldwide energy saving and environmental preservation movement [1, 2]. The focus lies in the NIR light which carries approximately 50 % of solar energy and, at the same time, has no influence on lighting and sight, which are undoubtedly the prime functions of windows. In addition, reduction of solar heat for cars and buildings mediated by the infrared radiation component of energy from the sun

is desirable, as it would make cooling more efficient. It would reduce energy demand for both air conditioning, and also for powered lighting if daylight is not excessively blocked at the same time [3, 4].

Some materials, such as heat-reflective glasses and sputtered-coating glasses, are promising to meet these demands mentioned above in the future. However, a main hurdle facing them is mass production which is not practical at present, owing to their low yield and the time or energy-consuming process.

Polymer nanocomposite is a kind of novel and fast-growing material with nano-sized filler being uniformly dispersed in polymer matrix. Polymer nanocomposite possesses enhanced thermal properties, mechanical strength and dielectrics when compared to conventional polymer composites, as a result of strong interactions between polymer matrix and filler [5, 6]. To apply polymeric nanocomposite is a unique approach to reducing heat by blocking near infrared waves while guarantee transmission in visible region for its potential low cost and high productivity.

A well-known material to serve the purpose is represented by the nanoparticles such as tin-doped indium oxide (ITO) [7–10] and lanthanum hexaboride (LaB₆) [11–13]. They are known for providing highly transparent solar filters to absorb infrared rays. LaB₆ is a metal-like plasmonic material, and its nanoparticle with free electrons on surface absorbs NIR of 750–1,300 nm via surface plasmon resonance. ITO is a conductive oxide material, and its nanoparticle absorbs NIR of 1,300–2,500 nm due to inherent band structure [14].

Poly(vinylbutyral) (PVB) is composed of vinyl butyral, vinyl alcohol and vinyl acetate. Properties of PVB can be controlled conveniently by tuning ratios of each monomer in molecular chain where hydrophobic vinyl butyral unit

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provides elasticity, toughness and compatibility with various plasticizers, and hydrophilic vinyl alcohol unit contributes high adhesion to inorganic substrates such as glass. Hence, it is extensively applied in various fields due to the randomly amorphous structure. PVB film possesses similar properties to rubber owing to its intrinsic structure and assistance of plasticizer [15]. Moreover, PVB is a nontoxic, odorless and environment-friendly polymer and thus used as functional materials widely. All in all, PVB is an excellent organic component in organic/inorganic hybrid composite, due to its outstanding compatibility with inorganic materials [16].

LaB₆ and ITO nanoparticles have been added into PVB [17, 18], PMMA [19] polymer matrix, respectively, to prepare nanocomposite films for automotive and architectural windows. Yuan et al. [19] have employed modified LaB₆ with different sizes in polymethylmethacrylate (PMMA) matrix to investigate effect of particle size on optical properties of LaB₆/PMMA composites. Schelm and Smith [4] separately prepared LaB₆/PVB and ITO/PVB composite films, but showed opposite results for glazing applications.

In order to take full advantage of the two kinds of nanoparticles, it is desirable to create LaB₆-ITO/PVB composite, so that unique properties of each material can be integrated, and interactions among the three components may bring out some new properties. Moreover, to the best of our knowledge, there is no report that focuses on the integration of the two solid nanoparticles in PVB matrix. Herein, for the first time, our present paper elaborates the preparation and characterization of this new polymer nanocomposite film, which incorporate LaB₆ and ITO nanoparticles simultaneously into PVB matrix, and their synergetic effect on optical property and thermal stability of the nanocomposite films.

2 Experimental

2.1 Materials

PVB (average molecular weight = 35,000–45,000 g/mol), ethanol, dodecylbenzenesulfonic acid (DBS) and polyvinylpyrrolidone (PVP) were purchased from Aladdin Chemical Co., China. Solid particles of ITO and LaB₆, with an average diameter of 80–120 and 20–30 nm, respectively, were provided by Ronghua technology Co. Ltd, China.

2.2 Characterization techniques

Crystal structures of LaB₆, ITO, pure PVB and nanocomposite films were measured by X-ray diffraction (XRD,

Rigaku, D-MAX2500) using Cu K_α radiation ($\lambda = 0.1542$ nm) at a scanning rate of 8°/min in the 2θ range of 15–80°. FT-IR spectra were recorded by Nicolet-6700 (USA) spectrophotometer with 4 cm⁻¹ resolution. The KBr pellet technique was applied for monitoring changes in FT-IR spectra of samples in range of 4,000–400 wave numbers (cm⁻¹). The vibrational transition frequencies are reported in cm⁻¹. The UV-vis-NIR transmission spectra of the nanocomposite films were investigated by Shimadzu UV-3600 (Japan) UV-vis-NIR Spectrophotometer in wavelength scan range of 260–2,400 nm. Thermogravimetric analyses (TGA) were performed on STA449C Simultaneous thermal analyzer at heating rate of 10 °C/min from room temperature to 700 °C under nitrogen atmosphere. Morphology of the nanoparticles (NPs) in PVB matrix was detected by Hitachi-SEM (S-4800, Japan).

2.3 Preparation of PVB/LaB₆/ITO nanocomposite films

Suspension of the two solid powders in ethanol was conducted separately as following: (1) LaB₆ powders (1 wt%) and anionic surfactant DBS (0.05 wt%) were simultaneously dispersed in the solvent. After a few minutes of stirring, a navy blue suspension was obtained and subsequently transferred to planetary ball mill for better homogenization; (2) ITO powder (1 wt%) and surfactant PVP (0.10 wt%) were treated in the same way.

Then, PVB powder was dissolved in 50 ml of ethanol at 65 °C for at least 60 min under continuous stirring. After ITO and LaB₆ was dispersed, PVB solution (10 wt%) was mixed with the suspension by stirring for 2 h at room temperature and then ultrasonication for 1 h. Different concentrations of ITO NPs (2, 4 and 6 wt%) and LaB₆ NPs (0.3 wt%) were applied. The obtained mixtures were stirred for another 6 h at 65 °C with assistance of 30 min of sonication. The resultant suspensions were casted onto glass slides, followed by drying at 40 °C in an oven for 24 h. Subsequently, fully dried films were separated from the slide. Nanocomposite film was denoted as X %LaB₆-Y %ITO, where X (0.3) and Y (2, 4 and 6) represent mass ratios of LaB₆ and ITO in the film, respectively. Thickness of the films was controlled in 0.15–0.3 mm.

3 Results and discussion

3.1 XRD patterns

Figure 1 illustrates XRD patterns of LaB₆ powder, ITO powder, films of pure PVB and 0.3 % LaB₆-4 % ITO composite. Index processes of diffraction patterns are done

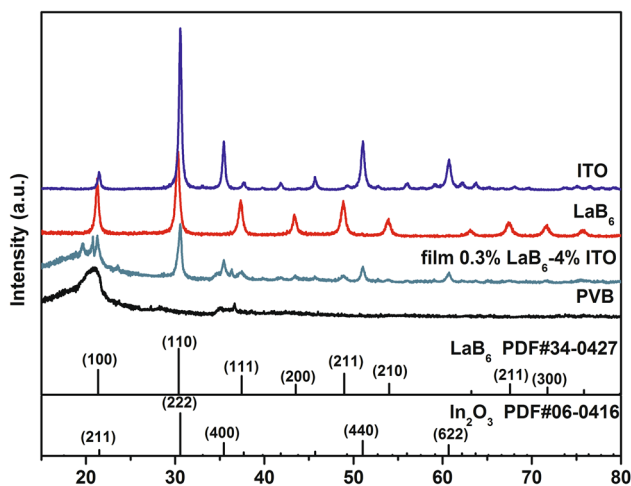


Fig. 1 XRD patterns of nanoparticles of LaB₆ and ITO, films of pure PVB and 0.3 % LaB₆-4 % ITO composite

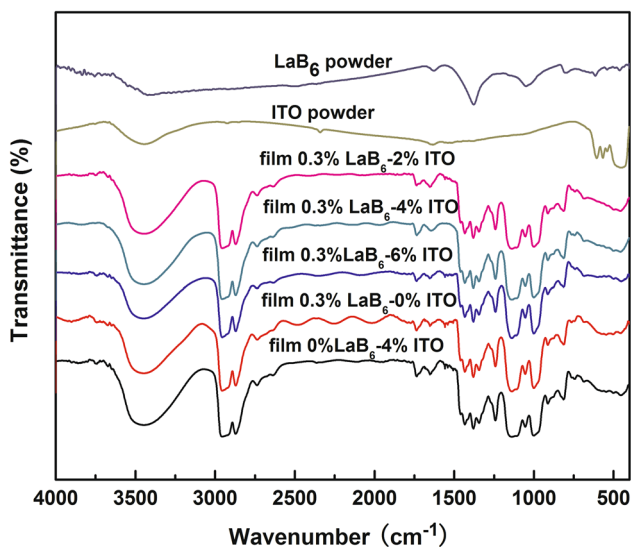


Fig. 2 FT-IR spectra of ITO powder, LaB₆ powder and various nanocomposite films

for the solid powders. Many diffraction peaks characteristic of LaB₆ and ITO (JCPDS34-0427 and 06-0416, respectively) are noticed in the patterns. The result suggests that these powders were well crystallized with no crystallographic impurity. Two typical peaks appear around $2\theta = 21.5^\circ$ and 41.0° in the pattern of pure PVB film. In the case of 0.3 % LaB₆-4 % ITO composite film, there remain many inherent diffraction peaks of ITO and LaB₆. Although in it typical peaks of PVB are not clearly observed, it still demonstrates considerable impact by two broadened peaks at $2\theta = 21.0^\circ$ and 36.0° . The observations imply that ITO and LaB₆ remain stable after dispersion in the PVB matrix.

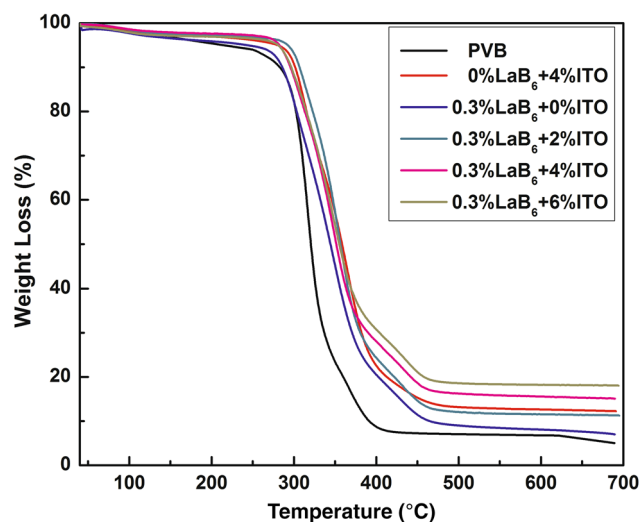


Fig. 3 TGA curves of neat PVB and various nanocomposite films

Table 1 TGA data of neat PVB and various nanocomposite films

Samples	T ₅ (°C) ^a	T ₁₀ (°C) ^a	Char yield (%) ^b
Neat PVB	206	280	5
Film 0 % LaB ₆ -4 % ITO	280	301	12
Film 0.3 % LaB ₆ -0 % ITO	234	285	7
Film 0.3 % LaB ₆ -2 % ITO	281	297	11
Film 0.3 % LaB ₆ -4 % ITO	284	297	15
Film 0.3 % LaB ₆ -6 % ITO	292	307	18

^a Temperature where the 5 and 10 % weight loss were recorded by TGA at a heating rate of 10 °C/min under a nitrogen atmosphere

^b Weight percentage of material left undecomposed after TGA analysis at a temperature of 700 °C under a nitrogen atmosphere

3.2 FT-IR spectra

Figure 2 presents FT-IR spectra of LaB₆ powder, ITO powder, 0.3 % LaB₆-2 % ITO, 0.3 % LaB₆-4 % ITO, 0.3 % LaB₆-6 % ITO, 0.3 % LaB₆-0 % ITO and 0 % LaB₆-4 % ITO. H-O-H bending region (1,600–1,700 cm⁻¹) and O-H stretching region (3,100–3,700 cm⁻¹) exist in FT-IR spectra of both the ITO and LaB₆ powders, which is attributed to absorption of H₂O on the surface of the solid particles. In the case of ITO, absorption bands in the region of 400–600 cm⁻¹ are derived from In-O vibration. In the spectra of nanocomposite films, peaks at 1,600–1,700 and 3,100–3,700 cm⁻¹ are originated from bending and stretching vibration of O-H in PVB matrix and ethanol. The absorption peaks at 2,942 cm⁻¹ are characteristic of stretching vibrations of C-H and CH₂ groups, and those at 1,420 cm⁻¹ correspond to vibration of CH₂ bond in PVB molecule. These spectra are similar in appearance, which could be attributed to low contents and weak absorptions of ITO and LaB₆ NPs.

3.3 Thermal degradation characteristics

Figure 3 displays TGA curves of neat PVB and the nanocomposite films. According to the TGA results, thermal degradation of the neat PVB and the various composite films proceeds in two steps: firstly a small and gentle weight loss, then a large and steep one when heated from room temperature to 700 °C. The first weight loss took place in the range of 100–150 °C, which might be caused by evaporation of moisture in PVB matrix and interfacial

interactions between them, as shown in Fig. 3. For the neat PVB, the second weight loss in the range of 200–350 °C is probably due to the polymer dehydration. However, that of the composite films apparently occurs in the range of 250–370 °C, which is slightly higher than that in the case of the neat PVB. In the curve of the neat PVB film, it is noticed that the second weight loss levels off at 400 °C, which corresponds to the decomposition of PVB into carbon black and hydrocarbon under nitrogen atmosphere. However, the same behavior took place even above 450 °C

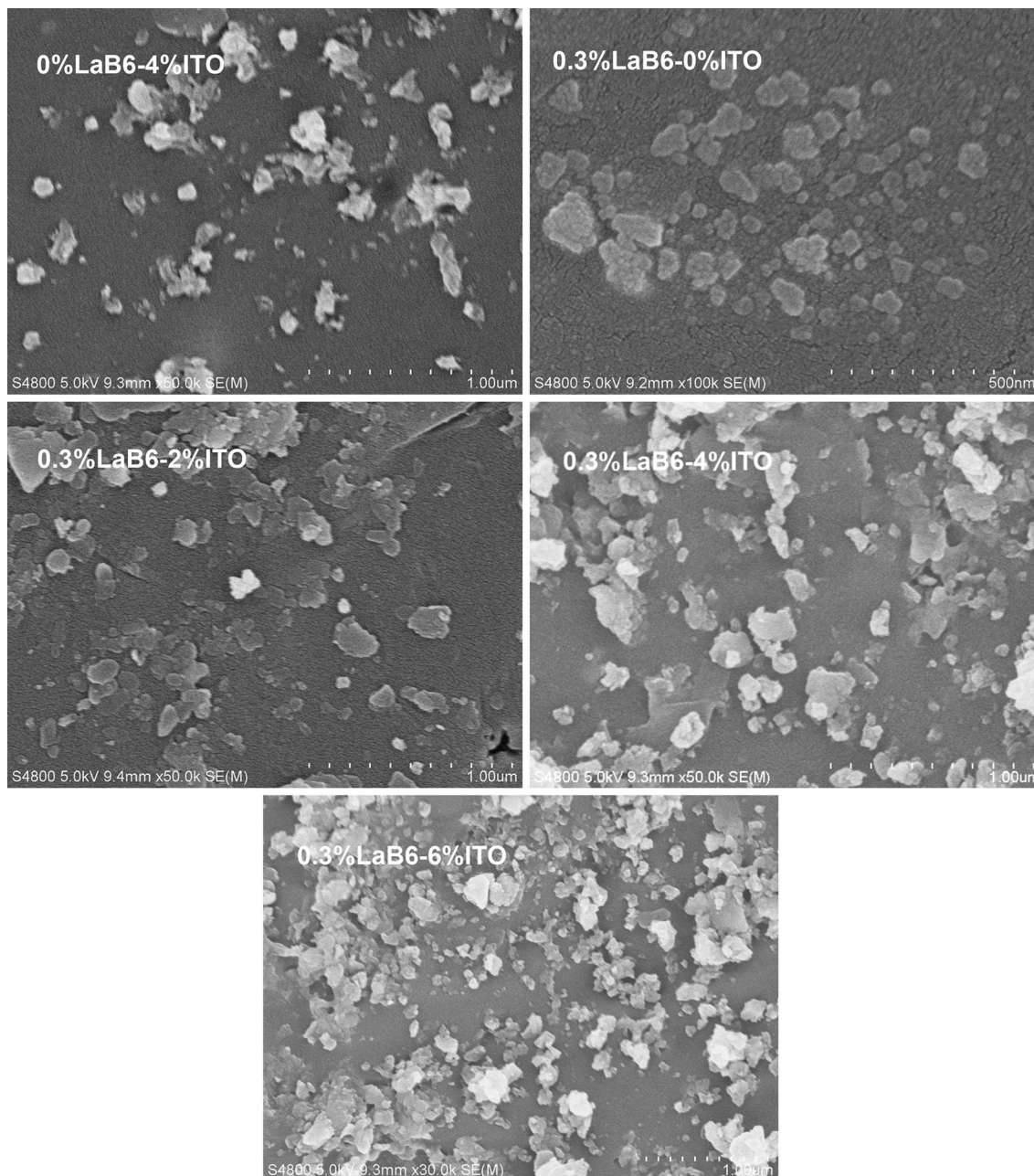


Fig. 4 FE-SEM photographs of the nanocomposite films

in the nanocomposite samples, and no obvious change is noticed in the behaviors among the five films. This phenomenon is attributed to the carbon black and solid powders in the films.

Table 1 shows the results of the thermal degradation. As expected, the char yields (7–18 wt%) of the nanocomposite films after heating are higher than that (5 wt%) of the pure PVB film. Moreover, T₅ where 5 % of weight loss occurs shifts from 206 °C in the case of the neat PVB film to 234 °C in the case of the composite films only containing 0.3 wt% of LaB₆, even to 281 °C after adding 2 wt% of the ITO particles. As the same, T₁₀ is also improved from 280 °C to at least 285 °C. It should be noted that the addition of ITO and LaB₆ into the PVB matrix plays a crucial role in improving the thermal stability of the nanocomposite films.

This improvement on thermal stability might be mainly caused by the ITO particles which act as a barrier to enhance heat insulation and restrain permeability of volatile products in degradation. Moreover, the addition of a small amount of the modified LaB₆ into the PVB matrix indeed improves the thermal stability of the composites.

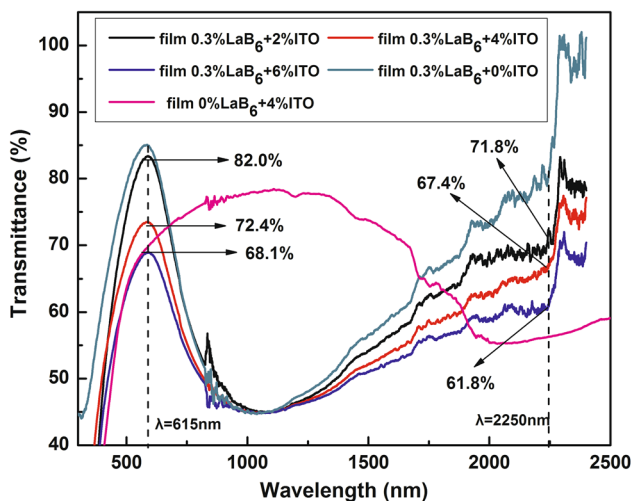


Fig. 5 UV–vis–NIR transmittance spectra of different nanocomposite films

Table 2 Transmittance of various nanocomposite films in Vis–NIR range

Sample	Transmittance at 615 nm (%)	Average transmittance (380–780 nm) (%)	Transmittance at 2,250 nm (%)	Average transmittance (1,000–2,250 nm) (%)
Film 0.3 % LaB ₆ + 0 % ITO	83.3	73.2	81.8	61.1
Film 0.3 % LaB ₆ + 2 % ITO	82.0	69.4	71.8	57.6
Film 0.3 % LaB ₆ + 4 % ITO	72.4	63.8	67.4	55.0
Film 0.3 % LaB ₆ + 6 % ITO	68.1	59.6	61.8	53.0
Film 0 % LaB ₆ + 4 % ITO	78.1	64.5	56.3	65.1

3.4 Field emission scanning electron microscopy (FE-SEM)

Figure 4 shows morphological images of the nanocomposite films investigated by FE-SEM. In these micrographs, it is observed that ITO and modified LaB₆ NPs were homogeneously dispersed in the PVB matrix, owing to their excellent adhesion and strong interfacial bonding to PVB molecules.

3.5 UV–vis–NIR spectra

Figure 5 shows transmission spectra of nanocomposite films in region of 250–2,400 nm. It is demonstrated that LaB₆ played a key role in absorbing waves between 750 and 1,300 nm, while ITO possessed a remarkable shielding effect against irradiation of 1,300–2,500 nm. Transmittance of these films at 615 and 2,250 nm, as well as average transmittance in the ranges of 380–780 and 1,000–2,250 nm, is calculated by a formula proposed by Schelm [4] and tabulated in Table 2. It is noted that when the addition of ITO increased from 0 to 6 % transmittance at 615 nm in visible region decreases from 83.3 to 68.1 %, while that at 2,250 nm in NIR region changes from 81.8 to 61.8 %. The average transmittance in the regions presents a similar trend to the transmittance at a single wavelength. In short, content of ITO added in the composites is inversely proportional to transmittance of NIR and vis between 300 and 2,500 nm. As we known, automobile glasses are required to have over 70 % transmittance but as low as possible thermal insulating property. Among five testing samples, film 0.3 % LaB₆–4 % ITO meets this demand. This result suggests that combination of LaB₆ and ITO NPs is promising for application in transparent and thermal insulating materials. However, testing thermal stability of these nanocomposite films under solar irradiation still needs further study.

4 Conclusions

The aim of this study was to investigate the synergistic effect of LaB₆ and ITO nanoparticles on thermal and

optical properties of the nanocomposite films based on PVB. TGA and UV–vis–NIR results indicated that the thermal stability and the property of blocking near infrared spectrum were obviously enhanced after addition of the two types of nanoparticles. The improvement on thermal properties of the composite materials was ascribed to strong hydrogen bonding between O–H groups in PVB and oxygen atoms of ITO nanoparticles. FE-SEM results also showed that nanoparticles of LaB₆ and ITO were dispersed homogeneously in PVB matrix. The whole process is simple and promising for potential industrialization of the nanoparticles, as well as the optical devices, especially windows of buildings and vehicles.

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