



# Trends and challenges in analytical chemistry for multi-analysis of illicit drugs employing wastewater-based epidemiology

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## Abstract

Wastewater-based epidemiology (WBE) for quantification of illicit drug biomarkers (IDBs) in wastewater samples is an effective tool that can provide information about drug consumption. The most commonly quantified IDBs belong to different chemical classes, including cocaine, amphetamine-type stimulants, opioids, and cannabinoids, so the different chemical properties of these molecules pose a challenge in the development of analytical methods for multi-analyte analysis. Recent workflows include the steps of sampling and storage, sample preparation using solid-phase extraction (SPE) or without extraction, and quantification of analytes employing gas or liquid chromatography coupled with mass spectrometry. The greatest difficulty is due to the fact that wastewater samples are complex chemical mixtures containing analytes with different chemical properties, often present at low concentrations. Therefore, in the development of analytical methods, there is the need to simplify and optimize the analytical workflows, reducing associated uncertainties, analysis times, and costs. The present work provides a critical bibliographic survey of studies published from the year 2020 until now, highlighting the challenges and trends of published analytical workflows for the multi-analysis of IDBs in wastewater samples, considering sampling and sample preparation, method validation, and analytical techniques.

**Keywords** Illicit drugs · Extraction · Liquid chromatography · Gas chromatography · Wastewater-based epidemiology

## Introduction

The consumption of drugs of abuse has increased in recent years, becoming a problem on a global scale, with implications for public health, security, and the economy [1–3]. According to the 2022 World Drug Report, 284 million people worldwide, aged from 15 to 64 years, used a drug in 2020, representing a 26% increase over 2010 data [4]. Drug consumption leads to environmental and health concerns, because wastewater treatment plants (WWTPs) are not designed to completely remove these compounds, which are considered emerging contaminants [5–8].

Traditionally, drug consumption data are obtained using population surveys combined with crime statistics and medical reports [7, 9–13]. However, this approach cannot monitor rapid changes in drug use [1, 7], in addition to being expensive and time-consuming and potentially underestimating actual consumption levels due to social taboos [7, 11, 12, 14, 15]. A complementary tool that has become consolidated globally is wastewater-based epidemiology (WBE), involving the quantification of chemical or biological markers, such as illicit drugs and/or their stable metabolites, from the analysis of untreated wastewater containing these substances excreted from the human body in urine and feces [3, 9, 10, 16]. The application of WBE for drug detection allows the estimation of short- and long-term consumption levels, with the advantages of being a faster, direct (almost real-time), noninvasive, accurate, and less laborious tool, compared to the use of population surveys [1, 3, 8, 11]. It is possible to investigate temporal trends of drug use by the community [3, 16] and in educational institutions [17], the use of new psychoactive substances (NPS) [18, 19], the influence of weekends and national holidays [20], festive events [6, 20], seasons [1, 9], and music events [12, 21], and the impact of

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social isolation measures imposed due to crises such as the COVID-19 pandemic [5, 10, 22–25].

The use of WBE employing multi-analysis of illicit drug biomarkers (IDBs) is highly attractive, because it can reduce costs and analysis times [26]. However, a wastewater sample is a chemically complex matrix, so it is a major challenge to develop analytical methodologies for the identification and quantification of compounds with different properties, especially considering the wide range of pKa and hydrophilic/lipophilic characteristics [20]. The classic illicit drugs most investigated in WBE (see Figure S1) are represented by four chemical classes: cocaine, amphetamine-type stimulants (ATS), opioids, and cannabinoids [1, 9, 16, 22]. Cannabinoids are not always included in the same analysis with other IDBs (medium to high polarity) due to the low polarity, lipophilicity, and acidic character ( $pK_a \cong 4.7$ ) of the main biomarker, 11-nor-9-carboxy- $\Delta^9$ -tetrahydrocannabinol (THC-COOH) [7, 14, 15].

Several published studies have reported different analytical workflows for the detection and quantification of IDBs in wastewater. The most common is the extraction of analytes by solid-phase extraction (SPE) followed by analysis using liquid chromatography–tandem mass spectrometry (LC–MS/MS). However, other less common analytical protocols for the quantification of IDBs have also been reported [27–29]. The aim of the present work is to discuss the challenges and trends of analytical workflows for multi-analysis of IDBs in wastewater samples, providing a critical appraisal of sample collection and preparation methods, validation procedures, and analytical techniques. In an attempt to capture recent trends, most of the studies cited were published from the year 2020 onwards. Table S1 summarizes the information on analytical workflows of the most recent studies concerning the identification of IDBs in wastewater.

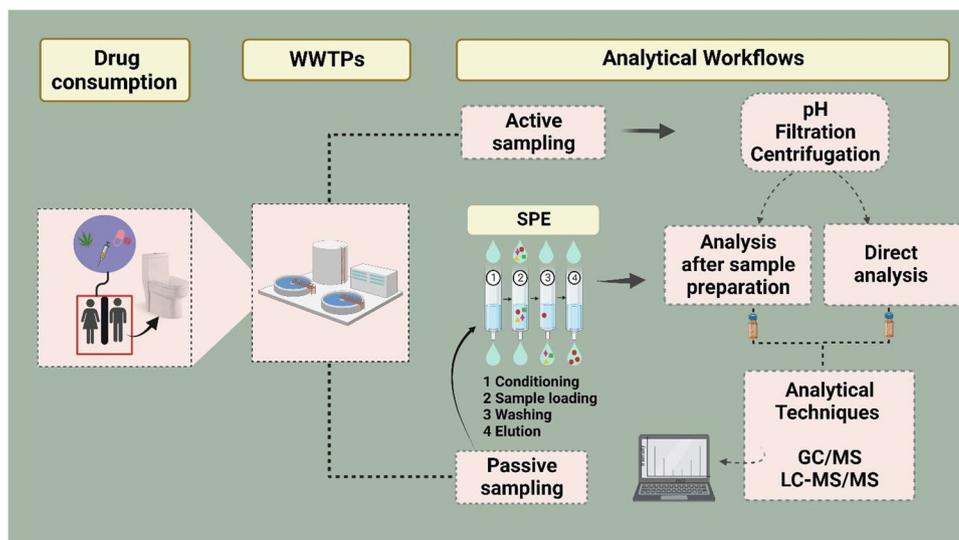
## Sampling and conditioning after collection

The analytical procedures in WBE for IDBs involve the steps of sampling, sample preparation, and instrumental analysis [11, 26], shown graphically in Fig. 1.

The sampling methodology directly affects the representativeness of the samples [30], so for this reason, 24-h composite sampling of wastewater using active sampling with autosamplers has been the predominant approach [1–3, 5–7, 11, 12, 31, 32]. Given that autosamplers are sophisticated equipment, other sampling strategies have also been used, such as short sampling times [17] and grab sampling [20, 33, 34].

Passive sampling has attracted increasing attention, especially due to its low cost [27]. The most reported passive sampling method for the determination of IDBs in wastewater is the use of polar organic chemical integrative samplers (POCIS), which consist of a medium to high hydrophilicity solid phase compressed between two porous membranes [16, 27]. These samplers can be immersed in water for more than a week, accumulating compounds by passive diffusion [16, 27, 30], enabling the preconcentration of analytes and attainment of low quantification limits [27]. While 24-h composite sampling allows monitoring of the rise and fall of drug consumption during specific days, such as at weekends, POCIS makes it possible to monitor long-term peak events [16, 30]. The study carried out by Bishop et al. [16] compared the estimates of drug use obtained by 24-h composite sampling and passive sampling using POCIS for 30 days. Although the POCIS values were mostly underestimates (possibly due to occlusion of the POCIS membranes), the results were considered acceptable. It should be noted that POCIS has limited application for cannabinoids, possibly due to the

**Fig. 1** Schematic summary of the steps involved in the analysis of IDBs in wastewater



composition of the sorbent, and, to the best of our knowledge, has only been adopted in a few studies [27].

Despite presenting adequate performance, other limitations of POCIS are related to the presence of suspended solids and microorganisms, as well as the possibility of accumulation of analytes in the membrane, instead of in the adsorptive phase, which can compromise sampling performance [16]. As an alternative, McKay et al. [30] suggested two configurations of passive samplers with affinity for acidic, neutral, and basic analytes, using (1) microporous polyethylene tubes (MPTs) loaded with a hydrophilic and lipophilic sorbent, and (2) MPTs loaded with a gel formed by a sorbent with the aforementioned properties, together with agarose. The MPT sampler was considered sensitive and presented linear accumulation over time for cocaine, ATS, opioids, and 20 other compounds. Another type of passive sampling that can be highlighted is the use of diffusive gradients in thin films (o-DGT) to sample organic compounds. Liu et al. [35] tested three different resins (Oasis HLB from Waters Corporation, USA; XAD18 from Sigma-Aldrich, USA; and XDA-1 from Sunresin Co. Ltd., China) as binders to the o-DGT membrane for passive sampling of 15 illicit drugs and other organic compounds. These resins provided similar results in terms of the amounts and concentrations of the analytes quantified.

Several studies have suggested that wastewater sampling should be performed by collection proportional to the flow, which takes into account possible fluctuations in concentrations and flows over time [6, 7, 24]. However, this requires the use of a flowmeter to activate an automatic sampler, which limits its application [36]. Consequently, sampling proportional to time is most widely employed, where the recommended collection interval does not exceed 5–10 min, to follow the flux of toilet flushes [15]. In addition, an ideal monitoring campaign requires long-term sampling at a wide range of WWTPs [32], enabling the tracking of possible consumption peaks and the entry of new psychoactive drugs over time. However, due to technical limitations, the 7-day sampling campaign is the most frequent [2, 11, 21, 32, 37–39].

After sample collection, one of the greatest challenges in WBE is to maintain the chemical stability of IDBs, since they can be significantly biotransformed within hours. This is especially important in situations where it is not possible to analyze the samples immediately, which requires that they be stored (sometimes for weeks) under appropriate conditions [15]. Storage of raw wastewater samples at low temperature is often used to decrease microbial degradation, with temperatures of 4 °C [6, 7, 12, 16, 23, 39, 40] and especially –20 °C [8, 9, 32, 33, 37, 38, 41, 42] being most reported. This strategy can be effective in maintaining short-term stability. However, for long-term storage, freezing the SPE cartridge before elution has been shown to

be the best way to ensure the stability of IDBs [11, 43]. In addition, adjustment of samples to pH 2 is often performed [1, 7, 16, 20, 21, 33, 41, 44, 45], which inhibits microbial activity [15, 16]. However, an acidic medium can lead to decreases of the concentrations of opioids [41] and favors the biotransformation of THC-COOH into 11-hydroxy- $\Delta^9$ -tetrahydrocannabinol (THC-OH) [15], making it difficult to recover in the samples. Hence, some studies have avoided acidification [43], maintaining the pH at the value measured at the time of collection, which is usually between 7 and 8 [3, 6, 9, 10, 12, 24, 37].

## Sample preparation

Sample preparation represents the most costly and time-consuming step, and possible errors can lead to sample contamination and loss of analytes, affecting the accuracy of the results. Wastewater samples collected are composed of suspended solids, colloids, and microorganisms that can adsorb the analytes or can cause clogging of the SPE cartridges, affecting the accuracy of the results [26, 41]. Centrifugation and filtration with fiber filters are the two most widely used approaches to remove these compounds [11, 39, 41, 44]. An alternative to the aforementioned filtration is the use of syringe filters before SPE [27, 33, 41] and/or before injection in the analytical instrument [6, 38, 40, 43, 45]. However, the possible loss of some IDBs during filtration processes must be considered (especially for less polar compounds, such as cannabinoids), since the compounds may be adsorbed on the material [16]. Pandopulos et al. [46] investigated the loss of cannabinoids onto a glass microfiber filter, in comparison to analysis by LC–MS/MS without filtration, observing that the filtration step significantly reduced the recoveries.

Wastewater samples are chemically complex mixtures with analytes present at very low concentrations (on the order of nanograms per liter), so the analysis requires the application of effective sample preparation methods to remove interferents and preconcentrate the target analytes, in order to provide greater sensitivity in the analytical determination of IDBs [29, 47]. SPE is the predominant sample preparation technique used in WBE research [5, 48]. Most studies have employed offline SPE [1, 6, 9, 11, 21, 23, 24, 37], where sample preparation and chromatographic analysis are performed separately [26]. However, disadvantages of this approach are that it is time-consuming, uses higher volumes of sample and solvents, and can suffer from analyte losses that decrease the extraction efficiency [41]. An alternative is to use online SPE [41] coupled with gas or liquid chromatographic analysis, which provides an automated and sequential process [15, 26], decreasing the number of steps and, consequently, reducing both the possibility of contamination and the use of organic solvents [41]. In comparison of

online and offline SPE protocols, Wang et al. [41] reported variations lower than 21% in the concentrations of 12 IDBs, indicating the suitability of online SPE as a way to simplify the procedure and reduce the time required for extraction of IDBs from wastewater samples. However, limitations of online SPE are related to optimization of the elution steps and especially the preconcentration [49], where a high sample volume is required to reach detectable levels, which may not be supported by the technique.

The selection of an appropriate sorbent for SPE is crucial, since it must ensure effective extraction and recovery of the analytes, without any significant losses [1]. The type of sorbent employed must consider the properties and the chemical structures of the target IDBs [5]. Reversed-phase sorbents with hydrophilic and lipophilic properties, such as Oasis HLB and Strata-X (Phenomenex, CA, USA), have been extensively used in the multi-analysis of molecules in a wide polarity range, providing good recovery percentages in the extraction of IDBs [9, 11, 12, 31, 37, 41].

Another more selective sorbent for SPE is the mixed-mode type, with reversed-phase and strong cation exchange, such as Oasis MCX (Waters Corporation, MA, USA) and Strata-X-C (Phenomenex, CA, USA) [1, 7, 16, 17, 21, 23, 45]. Although the addition of sulfonic acid to the divinylbenzene group makes it highly selective for basic compounds, these sorbents have been found to provide good recovery percentages (around 80%) for cannabinoids, as reported in studies carried out by Sulej-Suchomska et al. [1] and Cruz-Cruz et al. [2]. Christophoridis et al. [39] compared the use of Oasis HLB and Oasis MCX cartridges for extraction of IDBs from wastewater samples, where the HLB cartridges provided slightly lower recoveries for all the compounds evaluated (except for cannabinoids). More recently, a novel mixed-mode weak cation exchange sorbent for SPE was proposed, involving the modification of polystyrene-divinylbenzene (PS-DVB) microspheres with mercaptosuccinic acid. The authors reported satisfactory recoveries (84–106%) for the extraction of cocaine, ATS, and opioids, when the sorbent was compared with other SPE cartridges (Oasis HLB, Oasis MCX, and Oasis WCX, from Waters Corporation, MA, USA) [47].

In general, the reported studies show that there is no single solution when selecting the SPE sorbent, especially when cannabinoids (low polarity) are included with other polar analytes from the classes of cocaine, ATS, and opioids. It is likely that one of the main factors leading to diverse analytical results, comparing different studies that used a similar SPE sorbent, is the strong influence of the wastewater matrix, whose chemical composition varies considerably according to region [30].

The solvents employed in SPE of IDBs in wastewater depend on the composition of the chosen cartridge, although methanol is by far the most widely used. In the cartridge

conditioning step, the commonest procedure is to pass methanol followed by ultrapure water [3, 6, 9, 11, 12, 37, 39, 43], or acidified ultrapure water when a mixed-mode cation exchange sorbent (MCX) is used [1, 2, 5, 7, 16, 17, 21, 26]. The reported volumes of wastewater samples range from 50 mL to 1 L, at low flow rates, while for MCX sorbents, sample acidification may be necessary to promote ionization of the molecules [5, 26]. The cleanup step uses a washing solvent that must be sufficiently strong to remove the interferences, with the most common being water or methanol [2, 3, 9, 12, 24, 37, 41, 43]. However, a cleanup step is not always performed, since it can cause losses of analytes [1, 5, 6, 11, 17, 21]. For the elution step, methanol (sometimes with acid or base additives) is typically used [3, 6, 9, 11, 12, 32, 37, 39, 41, 43]. The use of methanol basified with ammonia is less commonly reported, but is suitable for elution from MCX sorbents [1, 7, 16, 17, 21], with the objective of neutralizing the charges and disrupting the electrostatic interactions between the analytes and the sorbent [26].

Despite being the dominant technique, drawbacks of SPE that need to be overcome include the cost and the time required. Liquid–liquid extraction (LLE), which is faster and consists of simpler steps, has been applied for the quantification of cannabinoids in wastewater samples [46, 50], considering that the other IDBs have high water solubility, which hinders their mass transfer to the organic phase. Liquid-phase microextraction (LPME) has also been applied due to its simplicity and use of lower volumes of organic solvents ( $\mu\text{L}$ ) and sample [25]. Wu et al. [8] developed an enrichment bag-based liquid-phase microextraction (EB-LPME) technique, where a flat polypropylene membrane bag was used for the extraction of six IDBs. In order to automate LPME, Nascimento et al. [25] developed a semi-automated LPME method for the simultaneous extraction of nine illicit drugs.

Chen et al. [48] investigated the use of thin-film microextraction (TFME) with polydimethylsiloxane (PDMS) loaded with divinylbenzene (DVB) particles, which provided rapid extraction of IDBs using low volumes of sample and solvent, although the TFME efficiency was strongly influenced by pH, with low performance in an acidic medium. In another study, Zhang et al. [29] synthesized magnetic polystyrene-divinylbenzene-glycidylmethacrylate microspheres modified with nano-petal-shaped covalent organic frameworks (NP-COF@Mag-PS/DVB/GMA), applied as sorbent for magnetic dispersive SPE of 12 IDBs in wastewater samples. The authors reported that the microspheres could be reused 20 times and that the extraction efficiency was dependent on the pH of the sample ( $\sim 6$ – $8$ ), since acidic and basic conditions could suppress the hydrogen bonds between the  $-\text{NH}-$  groups in the IDBs and the microspheres due to the ionic state of the polar drugs.

The available information suggests that although the use of SPE for multiple classes of illicit drugs can be challenging

and costly, especially in the case of large monitoring campaigns, this technique is by far the most effective extraction option due to the efficiency of preconcentration and isolation of analytes. For this reason, the commercial SPE products currently employed are reliable and have become the industry standards.

Nonetheless, other studies have employed the direct injection (DI) of wastewater samples after applying sample filtration and/or centrifugation [5, 32–34, 42, 44]. Ng et al. [34] developed a method employing DI into an LC–MS/MS instrument for determination of compounds belonging to the cocaine, ATS, and opioid classes, achieving fast separation and adequate sensitivity. Comparison was made of the results obtained by DI and SPE, which showed that DI was more reliable, since the areas of the illicit drug peaks were considerably larger and detectability of the compounds was higher than when using SPE. It appears that DI of wastewater into chromatography instruments may be an attractive option when the analyte concentration meets the minimum validation requirements, especially the limit of quantification (LOQ). It is important to point out that technological advances currently allow LOQ values to be obtained using DI that would previously have required sample preconcentration steps. However, it is necessary to evaluate possible strong matrix effects, and there may be a need for more regular maintenance of the analytical instruments.

## Analytical Techniques

LC–MS/MS has been extensively described for the identification/quantification of IDBs in wastewater samples [3, 6, 8, 9, 12, 24, 30, 43] due to its robustness, excellent sensitivity and selectivity, and advantages such as good separation with short retention times [8, 14, 15, 27, 34]. The LC systems reported include the use of high-performance liquid chromatography (HPLC) [1, 31, 38], ultra-high-performance liquid chromatography (UHPLC) [7, 27, 37, 41, 42, 47], and ultra-performance liquid chromatography (UPLC) [11, 17, 39, 44, 45]. UPLC and UHPLC have similar configurations and are based on the separation principles of HPLC, but using a stationary phase with smaller particle diameter, which generates greater pressure in the chromatographic column, resulting in increased sensitivity and faster analysis [51]. Micro-liquid chromatography–tandem mass spectrometry ( $\mu$ LC–MS/MS) has also been reported for the analysis of cocaine, ATS, and NPS, with the advantage of using a low flow rate, resulting in greater ionization efficiency (less suppression of ions) and increased sensitivity of the analysis, compared to conventional UHPLC [52].

In the LC methods, the main separation mechanism reported is reversed-phase [12], with octadecyl silica (C18) as the commonest stationary phase [1, 6, 7, 27, 29, 32, 37,

39, 42]. Biphenyl [3, 21, 24, 34, 38], pentafluorophenyl [8, 9, 23], and hydrophilic interaction liquid chromatography (HILIC) [5, 53] columns have also been used due to the polar character of most IDBs. The solvents employed are frequently water as mobile phase A and methanol or acetonitrile as mobile phase B, usually with the addition of acid to favor the protonation of molecules [1]. The favoring of protonation is extremely important, considering that electrospray ionization (ESI) is the main ionization method [1–3, 7, 8, 10–12, 27–29], with the basic character of cocaine, ATS, and opioids favoring determination in positive acquisition mode [7, 8, 16, 21, 26, 32, 33, 41]. However, when acidic molecules (such as cannabinoids) are included among the analytes, two frequently used procedures are either to maintain positive-mode ionization for all classes of drugs [2, 3, 5, 6, 9, 24, 39] or to add negative-mode acquisition for cannabinoids [1, 3, 12, 17, 21, 23, 27, 37], since the latter provides more abundant ionization for cannabis molecules [1].

In the case of LC–MS/MS, the triple quadrupole (QqQ) is the most reported  $m/z$  analyzer [6, 7, 10, 12, 16, 31, 34, 38, 39, 41, 42] due to its sensitivity, selectivity, and robustness [14, 15, 34]. A hybrid triple quadrupole-linear ion trap (Qtrap) has also been used [17, 21, 24, 30, 33, 45, 47]. However, these analyzers may have limitations when the monitoring of many analytes is desired, due to the number of consecutive scans required for the quantification/confirmation of compounds, which makes the window time very wide. Therefore, high-resolution mass spectrometry (HRMS) has been explored as an alternative due to its potential to identify analytes by means of the exact mass [15]. For this, the Orbitrap analyzer has been most frequently cited in studies of illicit drugs in wastewater [2, 5, 24].

Despite its high sensitivity, reproducibility, and avoidance of the use of liquid solvents, gas chromatography/mass spectrometry (GC/MS) has been infrequently reported as a common analytical technique for the analysis of IDBs in wastewater, possibly due to the polar character of most of the compounds [7, 8, 14] as well as the need for derivatization with *N,O*-bis(trimethylsilyl)trifluoroacetamide (BSTFA) [54] or *N*-methyl-*N*-(trimethylsilyl)trifluoroacetamide (MSTFA) [55] prior to analysis, which makes the procedure more laborious [15, 56]. The chromatographic run times are also longer for GC/MS, compared to LC–MS/MS. In terms of GC columns, satisfactory separations of target analytes have been obtained using 5%-phenyl methyl siloxane columns, as in the work of Cong et al. [54], who quantified cocaine, ATS, and opioid compounds in wastewater samples using offline SPE and GC/MS analysis. The gas chromatograph was coupled to a mass spectrometer operating with an electron ionization (EI) source and a single quadrupole analyzer, while other work has used an ion-trap analyzer [55].

Considering that not every drug present in wastewater originates from illicit consumption, and that most IDBs are

chiral (as racemates or single enantiomers) [55], enantioselective separation of chiral drugs has been explored as a complementary tool [13, 43, 53, 55]. This approach allows a better understanding of the different pathways of drug synthesis, as well as the differences between prescription drug use, illicit use, and direct dumping [13, 22]. For example, ATS have an asymmetric carbon with two enantiomers [*R*-(-) and *S*-(+)] [17], so these are the IDBs that have been most investigated using LC–MS/MS [13]. Recent studies have reported the analysis of ATS by GC/MS [13, 55], with extension to cocaine and opioids [43], but it was essential to use chiral columns (such as cellobiohydrolase phase), which are more expensive than non-chiral columns [22]. To overcome this, some studies have added chiral derivatization reagents in the organic extract obtained after SPE, such as (*R*)-(-)- $\alpha$ -methoxy- $\alpha$ -(trifluoromethyl)-phenylacetyl chloride [(*R*)-MTPA-Cl], which proved to be more economical due to the use of conventional phenyl methyl siloxane columns for GC/MS analysis [13, 40, 55].

The sample preparation and instrumental analysis techniques discussed above are costly and tend to be time-consuming. As an alternative, Mao et al. [28] synthesized Au@Ag biosensors immobilized on a nanoporous glass paper functionalized with poly-L-lysine (pLL), with detection by surface-enhanced Raman spectroscopy (SERS), where the noble metal nanostructures promoted electromagnetic enhancement. Although the study was initially limited to the analysis of methamphetamine in wastewater, the results were in agreement with analysis by LC–MS/MS, showing that the method was a viable and innovative alternative to mass spectrometry. The work was restricted to one analyte, so the viability of the method for use in multi-analysis of illicit drugs remains unproven. However, it is expected that studies employing biosensors will be extended to other IDBs as an alternative to SPE followed by LC–MS/MS or GC/MS analysis.

## Analytical validation and consumption estimates

The main objective of analytical workflows is to develop methods that can provide reliable results and quantification of all the target analytes, with low limits of detection (LOD) and quantification (LOQ) [15]. To the best of our knowledge, there are no specific guidelines available concerning analytical validation of WBE methods for quantification of IDBs, although recent studies have used protocols provided by the International Conference on Harmonization (ICH) [3, 34, 40, 55], the European Medicines Agency (EMA) [7, 53], the International Union of Pure and Applied Chemistry (IUPAC) [25, 37], the Analytical Procedures and Methods Validation for Drugs and

Biologics Guidance for Industry [29, 47, 57], and the Scientific Working Group for Forensic Toxicology (SWG-TOX) [37]. It is also worth mentioning the best-practice protocol developed by the European Monitoring Center for Drugs and Drug Addiction (EMCDDA) and the Sewage analysis CORE group—Europe (SCORE) [58], as well as the established analytical methodology developed by Castiglioni et al. [59]. In general, the validation protocols employ figures of merit including selectivity, linearity, LOD, LOQ, accuracy, precision, matrix effect, and recovery [1, 3, 16, 21, 24, 37, 41].

Due to the unpredictable chemical composition of wastewater samples, it is a challenge to control the influence of matrix effects on the ionization of analytes in LC–MS/MS or GC/MS [8, 25]. Among the main strategies used to correct for these effects, the most successful has been the addition of deuterated internal standards (IS) to the samples [1, 11, 16, 25], as reported in all 30 of the studies considered in Tables S1 and S2. The use of IS can also be effective in correcting for possible losses of analytes during extraction and instrumental analysis [11, 16, 31]. Method validation guidelines indicate that acceptable recovery values for IDBs in wastewater are in the range from 80 to 120% [40]. However, it is evident that in multi-analysis approaches, the diversity of chemical compounds, with most IDBs being polar compounds, makes it difficult to meet the recovery requirements for the quantification of THC-COOH [46, 50] due to its low polarity and lipophilic character.

The LOD and LOQ values are defined as the lowest detected and quantified concentrations of analytes present in a sample, respectively. The most traditional method for LOD and LOQ calculation employs the values obtained for replicates of analytical blanks, although some authors have used administrative values [16, 21, 24, 37, 41]. However, obtaining analyte-free samples represents one of the major problems faced in the validation of analytical methods for the determination of IDBs in wastewater. Hence, water or the mobile phase have often been used [20, 38].

The SWGTOX guideline indicates that a laboratory can define analyte LOQ values according to an administrative decision, even when lower LOQ values might be obtained using replicates of analytical blanks [60]. Table S2 summarizes the LOD and LOQ values of the 30 most recent studies concerning the identification of IDBs in wastewater. It can be seen that lower LOD and LOQ values were achieved using HPLC, UPLC, or UHPLC coupled to tandem mass spectrometry with ESI ionization. These instrumental systems enabled low LOD and LOQ values to be obtained in the studies of Zhao et al. [38] and Sulej Suchomska et al. [1], mainly using UPLC and UHPLC, where a small particle diameter of the stationary phase provided better peak resolution [41, 45]. It can also be seen in Table S2 that the use of GC/MS instrumentation resulted in low LOD and LOQ

values [54], which were very similar to those obtained using LC instrumentation.

LOD and LOQ values  $\leq 5 \text{ ng L}^{-1}$  were obtained using offline SPE [1, 38, 45, 54, 61] or online SPE [41], showing the excellent ability of SPE protocols to preconcentrate analytes and provide sample cleanup. Alternative methods using EB-LPME for the extraction of ATS and opioids also achieved low LOD and LOQ (around  $10 \text{ ng L}^{-1}$ ) [8]. Higher LOD and LOQ values were found by Nascimento et al. [25], who employed LPME and GC/MS for quantification of cocaine and ATS, which could be explained by the absence of the derivatization step required for analysis of polar IDB compounds. It is noteworthy that the highest LOD and LOQ values shown in Table S2 were obtained in a study using MPT passive sampling and LC-MS/MS analysis [30], where sample preparation was performed without a cleanup step, so the matrix effect was likely to affect the performance of the analysis, because ion suppression in the ESI could affect the IDB signal responses.

Comparison of methods that used an extraction step [1, 38, 45] with techniques based on DI of the samples [33, 34, 42] (Table S2) showed that the LOD and LOQ values were higher without an extraction step, which could be explained by matrix effects and the absence of the preconcentration factor inherent in extraction techniques such as SPE. In the work of Cruz-Cruz et al. [2], employing LC-HRMS analysis, higher LOD and LOQ values for cocaine, ATS, opioids, and cannabinoids were obtained, when compared to the LC methods using tandem mass spectrometry with low-resolution  $m/z$  analyzers. This could be explained by the lower sensitivity of the HRMS technique, where the exact masses of ions were obtained but without selected ion monitoring.

Finally, a goal of WBE studies is to perform back-calculations from the quantified IDB values in order to estimate per capita consumption of drugs, as shown in the equation below. For this, it is necessary to have knowledge of other parameters including the concentration ( $\text{ng L}^{-1}$ ) of the biomarker ( $C$ ), the daily wastewater flow ( $Q_v$ ), a correction factor ( $f$ ) based on the molar mass ratio between the parent compound/metabolite and its percentage excretion in urine, and the population served by the WWTP ( $inhab$ ) [59, 62, 63].

$$\text{Daily consumption} = \frac{CxQ_vxf}{inhab} \text{ (mg/day/1000 people)}$$

For example, Sodr e et al. [6] used this equation to obtain average daily consumption values for cocaine and cannabis in Brazil, during a typical week (no festive events), of 1739 and 11,471 mg/1000inhab/day, respectively. These values compared with 2754 and 14,342 mg/1000inhab/day for cocaine and cannabis, respectively, during a festive carnival period. Hence, the consumption estimates indicated an increase in the per capita consumption of drugs

during festive events, which could provide useful information for public safety actions.

The population served by the WWTP can be estimated using census data or chemical parameters that make it possible to monitor population fluctuations due to holidays and festive events [3]. In recent years, the most widely used parameters have been biological oxygen demand (BOD) [5, 16], chemical oxygen demand (COD) [10, 12], ammonia (product of urea hydrolysis) [11, 21, 54], and creatine concentration [20]. Although reported in recent studies, the use of COD is not recommended as an indicator of population size [64], with  $\text{NH}_4\text{-N}$  [3] and total nitrogen [24] being considered the most reliable and realistic parameters. The decision concerning which to use will also strongly depend on equipment availability and the analytical costs [24].

## Outlook

It is evident that WBE studies represent an important way to estimate the consumption of drugs of abuse by a given community or to monitor their use during festive events and holidays, providing essential information for the formulation of public security, health, and social policies. However, WBE for quantification of IDBs is a dynamic field with different analytical workflows, which, despite being effective, require further improvements in terms of efficiency, speed, and cost reduction. Advances in knowledge about the stability of analytes in wastewater samples are still required, especially for multiclass analysis, due to the difficulties inherent in storing large amounts of samples for long periods, occupying little space and with retardation of microbial degradation, which emphasizes the need for faster analytical workflows or more effective analyte preservation procedures. The main analytical challenges are associated with the chemical diversity of the IDBs, requiring the development of new sorbents for cartridges used in SPE (the most common extraction method) or other sample preparation methods that comply with the need for preconcentration and reduction of matrix effects. The increasing use of different drugs and the emergence of new psychoactive compounds require the development of novel methodologies for estimation of long-term drug consumption that are simpler, faster, and inexpensive, and can increase the reliability of results.

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## Declarations

**Conflict of interest** The authors declare no conflict of interest.

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