Errata

R. Giordano, E. Sappa and S. A. R. Knox (1996). Homogeneous catalytic hydrogenation of diethyl and diphenylacetylene and isomerization of *cis*-stilbene in the presence of $[(C_5H_5)Ru_3(\mu-CO)(\mu_3-CO)(C_2Ph_2)]$. *J. Cluster Science* **6**, 179.

Page 189, Table III, line 8

The TON/h for Cluster 2 should read: 33.7^d not 11.2

Page 180, lines 10 and 11, the formula of complex (1) should read:
 $\dots[(C_5H_5)Ru_3(\mu-CO)(\mu_3-CO)(C_2Ph_2)]$ (1).

Page 183, line 6 of caption to Table 1 should read: 7.19×10^{-8} mol.

Page 183, line 11 of caption to Table 1 should read: 5.57×10^{-6} mol.

J. Mizutani, H. Imoto, and T. Saito (1995). Synthesis and structure of a seven electron triangular cluster complex $[Mo_3S_4Cl_3(dppe)_2(PEt_3)]$. *J. Cluster Sci.* **6**, 523.

Page 526, all minus signs in Table II were lost. The correct table is given below.

Table II. Positional and Equivalent Isotropic Thermal Parameters for
 $[Mo_3S_4Cl_3(dppe)_2(PEt_3)] \cdot 2THF \cdot MeOH$

Atom	x	y	z	U_{iso}
Mo1	0	0.07053(2)	0	0.0337(2)
Mo2	0.15533(7)	0	0.03905(9)	0.0375(4)
S1	0.0949(2)	0	0.1630(3)	0.0395(14)
S2	-0.1000(2)	0	-0.1522(3)	0.0446(14)
S3	0.0971(2)	0.09260(13)	-0.0730(2)	0.0440(12)
P1	0.0762(2)	0.17420(12)	0.1374(2)	0.0421(11)
P2	-0.09194(15)	0.16609(12)	-0.1445(2)	0.0442(11)
P3	0.2566(3)	0	-0.0489(4)	0.077(2)

Table II. Continued

Atom	x	y	z	U_{iso}
C11	−0.1061(2)	0.09188(13)	0.0674(2)	0.0551(13)
C12	0.3020(2)	0	0.2187(3)	0.0677(15)
C11	0.0919(6)	0.1817(5)	0.2866(7)	0.052(4)
C12	0.1169(6)	0.2459(7)	0.3367(9)	0.091(7)
C13	0.1277(8)	0.2497(7)	0.445(2)	0.112(9)
C14	0.1111(8)	0.2030(8)	0.5006(10)	0.095(8)
C15	0.0741(8)	0.1347(7)	0.4386(10)	0.097(7)
C16	0.0706(6)	0.1275(5)	0.3365(8)	0.069(5)
C21	0.1896(6)	0.2031(4)	0.1696(7)	0.054(4)
C22	0.2100(7)	0.2569(5)	0.1252(9)	0.087(7)
C23	0.2979(8)	0.2707(5)	0.1546(9)	0.080(6)
C24	0.3659(9)	0.2331(8)	0.2341(11)	0.105(10)
C25	0.3529(6)	0.1822(6)	0.2847(11)	0.100(7)
C26	0.2629(7)	0.1621(5)	0.2543(9)	0.075(5)
C3	0.0021(6)	0.2475(4)	0.0619(8)	0.062(4)
C4	−0.0309(6)	0.2446(4)	−0.0727(9)	0.061(5)
C51	−0.1013(5)	0.1716(5)	−0.2842(7)	0.052(4)
C52	−0.0853(6)	0.1137(5)	−0.3381(7)	0.069(5)
C53	−0.0985(7)	0.1194(7)	−0.4508(9)	0.101(7)
C54	−0.1229(7)	0.1819(6)	−0.5102(8)	0.083(6)
C55	−0.1378(6)	0.2381(7)	−0.4620(8)	0.088(7)
C56	−0.1325(6)	0.2331(5)	−0.3539(8)	0.063(5)
C61	−0.2093(5)	0.1816(5)	−0.1834(7)	0.049(4)
C62	−0.2346(9)	0.2337(5)	−0.1478(11)	0.096(7)
C63	−0.3333(7)	0.2441(7)	−0.1859(11)	0.098(8)
C64	−0.3890(7)	0.1985(8)	−0.2466(12)	0.118(9)
C65	−0.3653(7)	0.1416(7)	−0.2870(10)	0.110(8)
C66	−0.2711(6)	0.1298(6)	−0.2551(10)	0.093(7)
C71	0.322(2)	−0.073(2)	−0.038(3)	0.10(2)
C72	0.357(2)	−0.072(2)	0.068(2)	0.099(14)
C81	0.204(2)	0.0329(11)	−0.210(2)	0.13(2)
C82	0.109(2)	0	−0.297(2)	0.19(2)
C91	0.350(2)	0.078(2)	0.021(3)	0.13(2)
C92	0.4081(13)	0.0825(8)	−0.014(2)	0.093(12)
OA	−0.634(2)	0.0660(14)	0.581(3)	0.2
CA1	−0.562(4)	0	0.584(5)	0.2
CA2	−0.641(3)	0.036(2)	0.496(3)	0.2
CA3	−0.706(3)	0.025(2)	0.555(3)	0.2
CA4	−0.666(3)	0.053(2)	0.637(3)	0.2
OB	−0.313(3)	0	0.474(3)	0.2
CB1	−0.408(3)	0.056(2)	0.391(4)	0.2
CB2	−0.376(2)	0.027(2)	0.329(3)	0.2
CB3	−0.421(5)	0	0.387(7)	0.2
CB4	−0.398(3)	0.031(2)	0.472(3)	0.2
OC	−0.108(3)	0	0.307(3)	0.2
CC1	−0.049(4)	0	0.299(5)	0.2