Erratum

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Bone-Lead Analysis: Development of Analytical Methodology for Milligram Samples

[Archives of Environmental Contamination and Toxicology 24, 494–497 (1993)]

An error has occurred in the sample preparation procedure (p. 495, right column, para 1) of our paper. The correct sample preparation procedure is as follows: "The bone samples were accurately weighed out (10–20 mg) into pre-cleaned 1.5-mL polyethylene centrifuge tubes; a 100- μ L aliquot of ultrapure nitric acid (Seastar Chemicals, Vancouver) was added and allowed to digest overnight at ambient temperature. The milligram samples of bone were completely dissolved by the acid. The digest was diluted to 1 mL with high-purity water. To a **100 μ L aliquot of this solution were added 40 μ L each of ultrapure nitric acid and 1% $\text{Pd}(\text{NO}_3)_2$ and made up to**

1 mL to give a final concentration of 5% HNO_3 and 400 mg/L $\text{Pd}(\text{NO}_3)_2$. The samples were diluted appropriately when the Pb concentration exceeded the linear calibration range, but maintaining the final nitric acid to 5% and palladium nitrate concentrations to 400 mg/L." The corrections are in boldface type.

We thank Dr. William Stansley, Freshwater Fisheries Laboratory, New Jersey State Department of Environmental Protection and Energy, for having drawn our attention to the error in the sample preparation procedure presented in the paper. We apologize to any users who might have encountered difficulties with the original procedure.