

The effect of different layered materials on the tribological properties of PTFE composites

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Abstract: Two-dimensional (2D) lamellar materials have unique molecular structures and mechanical properties, among which molybdenum disulfide (MoS₂) and graphitic carbon nitride (g-C₃N₄) with different interaction forces served as reinforcing phase for polytetrafluoroethylene (PTFE) composites in the present study. Thermal stability, tribological and thermomechanical properties of composites were comprehensively investigated. It was demonstrated that g-C₃N₄ improved elastic deformation resistance and thermal degradation characteristics. The addition of g-C₃N₄ significantly enhanced anti-wear performance under different loads and speeds. The results indicated that PTFE composites reinforced by g-C₃N₄ were provided with better properties because the bonding strength of g-C₃N₄ derived from hydrogen bonds (H-bonds) was stronger than that of MoS₂ with van der Waals force. Consequently, g-C₃N₄ exhibited better thermomechanical and tribological properties. The result of this work is expected to provide a new kind of functional filler for enhancing the tribological properties of polymer composites.

Keywords: graphitic carbon nitride; lamellar materials; hydrogen bonds; wear resistance

1 Introduction

Polymers and their composites are extensively used as tribo-engineering materials. The excellent physical and chemical characteristics of polymer-based composites make them a promising class of important tribo-materials because of their self-lubricating performance, chemical stability, superior process ability as well as cost-effectiveness [1–3].

Polytetrafluoroethylene (PTFE) polymer exhibits extraordinary characteristics such as high thermal resistance, ease of fabrication, outstanding chemical inertness, low friction coefficient, etc. Therefore, considerable efforts have been made to extend its applications in industry and academic research where sealing and lubricating issues are of special importance. However, the poor mechanical properties and high

wear rate of PTFE have limited its application in the abovementioned fields [4–6]. To reduce the wear rate and exploit the advantages of PTFE, researchers have found that load-carrying capability and wear resistance can be improved by filling the PTFE matrix with various fillers such as organic and inorganic fibers [7, 8], and nano- and microscaled particles [9, 10]. For example, Burris and Sawyer reported improvement in anti-wear properties by up to four-orders of magnitude by incorporating nano-Al₂O₃ into PTFE [11]. The modification of tribological performance of PTFE composites was found to be closely related to the improvement in hardness, compressive strength, stiffness, and creep resistance [12–14].

Driven by the potential applications of PTFE composites in the automotive and aerospace industries, much attention has been paid to the enhancement of

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the anti-wear performance of PTFE composites. Fan et al. [15] showed that potassium titanate whiskers were effective in increasing the wear resistance of PTFE composites. Song et al. [16] revealed the synergistic effect of molybdenum disulfide (MoS_2) and glass fiber in enhancing wear resistance. When tribological stability is considered, PTFE composites were found to be superior to phenol formaldehyde resins and poly-p-hydroxybenzoic acid composites in vacuum condition [17].

Over the past several years, two-dimensional (2D) materials, e.g., WS_2 , $\alpha\text{-ZrP}$, MoS_2 , graphite, etc., have been recognized as excellent candidates for tribological modification of polymers owing to their unique molecular structures [18–20]. Among 2D materials, graphite-like carbon nitride ($\text{g-C}_3\text{N}_4$) (Fig. 1(a)) is of special interest because of the strong hydrogen bonds (H-bonds) between strands of polymeric melon units and NH/NH_2 groups (Fig. 1(b)), which contribute to its high mechanical properties [21–24].

Nevertheless, reports on the tribological properties of composites reinforced by $\text{g-C}_3\text{N}_4$ are very rare. In addition, few researchers have explored the thermal stability and tribological and thermomechanical properties of composites via the incorporation of

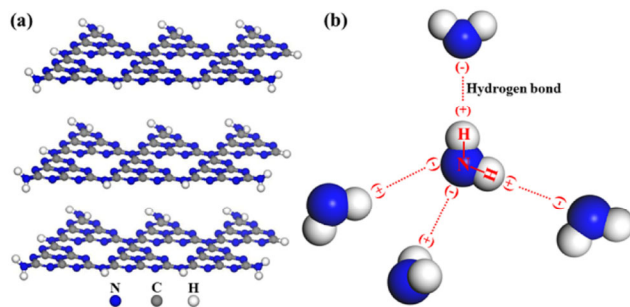


Fig. 1 Schematic illustration of (a) $\text{g-C}_3\text{N}_4$ and (b) interlayer H-bonds for $\text{g-C}_3\text{N}_4$.

Table 1 The formulations and related properties of PTFE composites.

Composites	Density (g/cm^3)	Hardness (Shore D)	Composition (wt.%)						
			PTFE	PTWs	Nano- SiO_2	Cu	Aramid pulp	MoS_2	$\text{g-C}_3\text{N}_4$
FM1 ^a	2.31	59.2	88	6.5	2.2	1.1	2.2	0	0
FM2 ^b	2.34	60.3	78	6.5	2.2	1.1	2.2	10	0
FM3 ^c	2.11	61.2	78	6.5	2.2	1.1	2.2	0	10

^a PTFE composites without addition of MoS_2 and $\text{g-C}_3\text{N}_4$ are abbreviated as FM1.

^b PTFE composites with addition of MoS_2 are abbreviated as FM2.

^c PTFE composites with addition of $\text{g-C}_3\text{N}_4$ are abbreviated as FM3.

different lamellar materials with different interaction forces between layers. In the present work, $\text{g-C}_3\text{N}_4$ is successfully synthesized, and its structure and morphology are inspected. Moreover, the addition of MoS_2 and $\text{g-C}_3\text{N}_4$ to worn surfaces and tribofilms of PTFE composites is also analyzed systematically. The experimental results show that sheet-like materials with H-bonds play a major role in enhancing thermal stability and tribological and thermomechanical performance.

2 Experiment

2.1 Materials and preparation

The formulations of PTFE composites are listed in Table 1 and obtained with our previous method [25–28].

Two different types of layered materials (MoS_2 , $\text{g-C}_3\text{N}_4$) were used as functional fillers to reinforce PTFE composites. Their morphologies are shown in Fig. 2. MoS_2 micropowders with an average diameter of 20 μm and purity of 99.9% were purchased from Beijing DK Nano technology Co. Ltd. The yellowish $\text{g-C}_3\text{N}_4$ powders were synthesized by directly heating melamine ($\text{C}_3\text{N}_6\text{H}_6$, Tianjin Kemiou Chemical Regent Co., Ltd.) at 550 $^\circ\text{C}$ for 4 h following previous works [27, 29]; the sizes are shown in Fig. 3. Detailed information on the other compositions is summarized in Table 2. The fillers were added and mixed in a shredding machine for 3 min. Next, the composites were fabricated by cold press (40 MPa, 20 min) and then sintered in an oven (375 $^\circ\text{C}$, 120 min).

2.2 Tribological tests

Tribological tests were conducted under ambient

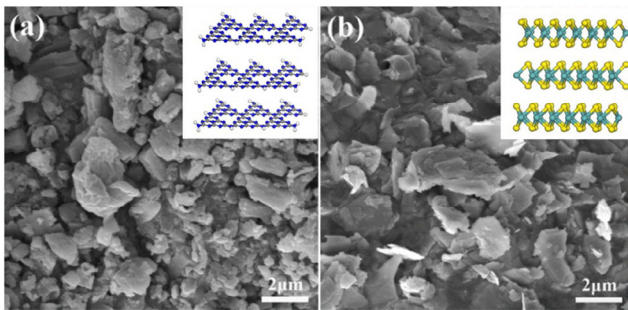


Fig. 2 SEM images of particles: (a) g-C₃N₄ and (b) MoS₂ (inset images represent corresponding schematic drawings of the crystalline structures of g-C₃N₄ and MoS₂).

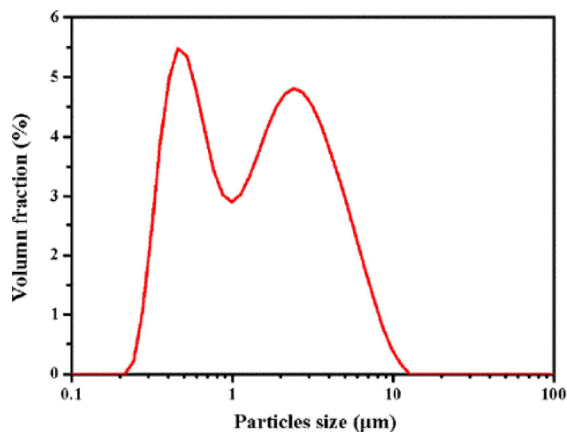


Fig. 3 The particles size range of the g-C₃N₄.

Table 2 The detailed information of raw materials.

Raw materials	Size	Supplier
PTFE	75 μm	Daikin Fluorochemicals Co., Ltd.
PTWs	0.5–2.5 μm	Shanghai Whiskers Composite Material Co., Ltd.
Nano-SiO ₂	20 nm	Beijing DK Nano technology Co., Ltd.
Aramid pulp	800–1350 μm	Teijin aramid Trade Co. Ltd.
Cu	5 μm	Beijing DK Nano technology Co., Ltd.

conditions on a tribometer (CSM, Switzerland) with ball-on-disc configuration, which is schematically illustrated in Fig. 4. During the tests, the counterbody stainless steel ball (GCr15, GB/T 18254-2002, 3 mm diameters) was static while the polymer composite disc rotated against the ball under normal load. The friction tests lasted 120 min at different sliding speeds (0.04, 0.08, and 0.12 m/s) and normal loads (1, 3, and 5 N), respectively. Prior to the tests, the surfaces of the

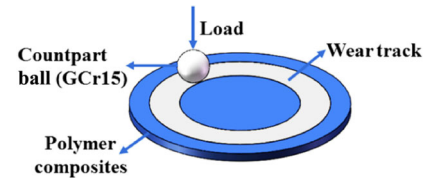


Fig. 4 Schematic illustration of ball-on-disc wear test.

specimens were finished by polishing to a roughness of approximately 0.2–0.3 μm. The steel balls were cleaned with alcohol-dipped cotton to remove surface humidity. Thereafter, each friction measurement was repeated at least three times to evaluate the friction coefficient and specific wear rate (W). The friction coefficient from the steady friction stage at last one hour was dynamically recorded, and W was calculated by the following relationship:

$$W = \Delta V / PL \text{ (mm}^3\text{/(N}\cdot\text{m))} \quad (1)$$

where P and L are the applied load and sliding distance, respectively. ΔV is the volume loss characterized according to the following equation:

$$\Delta V = SL_1 \quad (2)$$

where S and L_1 are the cross-sectional area and length of wear trace, respectively.

2.3 Characterizations and analyses

The microstructures of the specimens were measured by X-ray diffraction (XRD, Philips X' Pert Pro.) with Cu/K- α 1 radiation at 40 kV and 50 mA. The chemical structure of g-C₃N₄ was analyzed by Fourier transform infrared (FT-IR, Nexus 870) spectra. Tribochemical effects were carried out by X-ray photoelectron spectrometer (XPS, ESCALAB 250Xi, Thermo Fisher Scientific) with Mg K α radiation source. The thermal stability was explored by thermogravimetric analysis (TGA) on a Netzsch thermal analysis system (STA 449C) with temperature range of 30–800 °C in nitrogen at a rate of 10 °C/min. The storage modulus of friction materials was evaluated by dynamic mechanical analysis (DMA) on a DMA 242 C analyzer (Netzsch Instruments, Germany) with three-point bending mode heated from 25 °C to 250 °C at a rate of 5 °C·min⁻¹ and frequency of 1 Hz. The morphologies of the transfer films and worn surfaces were inspected with a field-emission scanning electron microscope (FE-SEM,

Mira 3, Xmu, Tescan) and coated with a thin gold layer to increase the resolution for observation. The 3D tomographies of wear scars were investigated using an optical interferometer (KLA-Tencer, MicroXAM-800) in at least five random areas. The size of g-C₃N₄ particles was characterized by laser particle diameter analyzer (Mastersizer 3000). The density and hardness of the samples were measured on the Micrometritics AccuPyc 1330 and HS-D TH-210 hardness tester (Beijing Time Technology Co., Ltd., China) on a Shore D scale, respectively.

3 Results and discussion

3.1 Analysis of g-C₃N₄ microstructure

FT-IR is an effective method to determine the structures and chemical bonds of the samples. The different functional groups of melamine and g-C₃N₄ in the 3,700–500 cm⁻¹ region are shown in Fig. 5(a). For the melamine, the double peaks at 3,467 cm⁻¹ and 3,419 cm⁻¹ are related to the antisymmetric stretching vibration of the -NH₂ group. The peaks at 1,551 cm⁻¹ and 1,646 cm⁻¹ correspond to the triazine aromatic stretching vibration and N-H bending vibration, respectively [30, 31]. Meanwhile, the peak at 810 cm⁻¹ corresponds to the ring deflection in melamine [31]. In the case of g-C₃N₄, the C≡N stretching peak at 1,200–1,650 cm⁻¹, triazine unit breathing peak at 810 cm⁻¹, and N-H stretching peak at broad band of approximately 3,000 cm⁻¹ were observed [22, 27, 32].

To confirm the formation of pure g-C₃N₄ crystal phases, the samples were analyzed by XRD. As seen in Fig. 5(b), there are two characteristic peaks at 13.0° and 27.5° corresponding to (100) and (002) diffraction

planes, which are associated with interlayer stacking of the conjugated aromatic system [33–35]. Therefore, synthetic g-C₃N₄ exhibits a typical graphite-like structure. From the above discussion, it is clear that g-C₃N₄ powders were successfully fabricated by directly heating melamine.

3.2 Thermal performance analysis of composites

The dynamic thermomechanical properties of PTFE composites were investigated by DMA at 1 Hz and temperature range of 25–250 °C, as shown in Fig. 6(a). The figure shows that all the composites exhibit similar temperature-dependent viscoelastic properties. Meanwhile, the storage modulus of samples can be improved by incorporating functional fillers. The g-C₃N₄-reinforced composites show higher and improved properties, which indicate that g-C₃N₄ has the ability to enhance the stiffness and elastic deformation resistance. On the other hand, the interlayer bonding strength of g-C₃N₄ originating from H-bonds is much higher than that of MoS₂ with van der Waals force. Therefore, the unique 2D layer-structure of g-C₃N₄ exhibits high mechanical properties, as mentioned in reference [36].

Excellent thermal stability has a positive effect in terms of enhanced tribological performance. Figure 6(b) and Table 3 show the influence of layered materials on thermal stability under nitrogen conditions. Data on thermal properties indicate that there is no obvious difference between thermal degradation temperatures of 5% and 10% weight loss. However, the char yield value of FM3 at 800 °C is higher than that of FM1 and FM2, which can be attributed to the higher thermal resistance of g-C₃N₄.

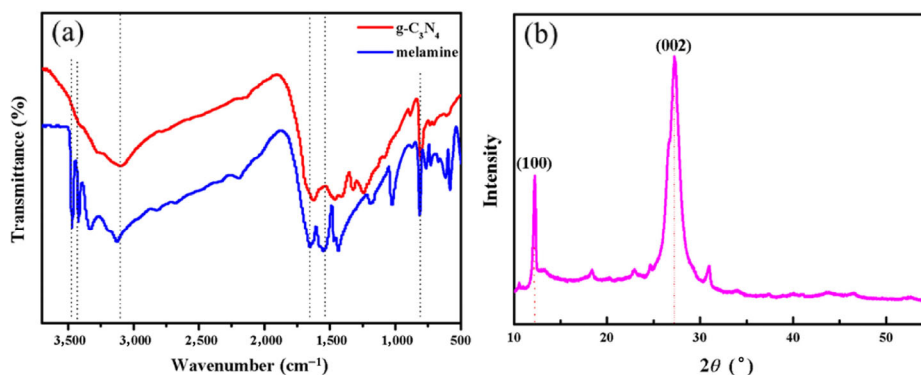


Fig. 5 Structural analysis of g-C₃N₄ powders: (a) FT-IR spectra and (b) XRD pattern.

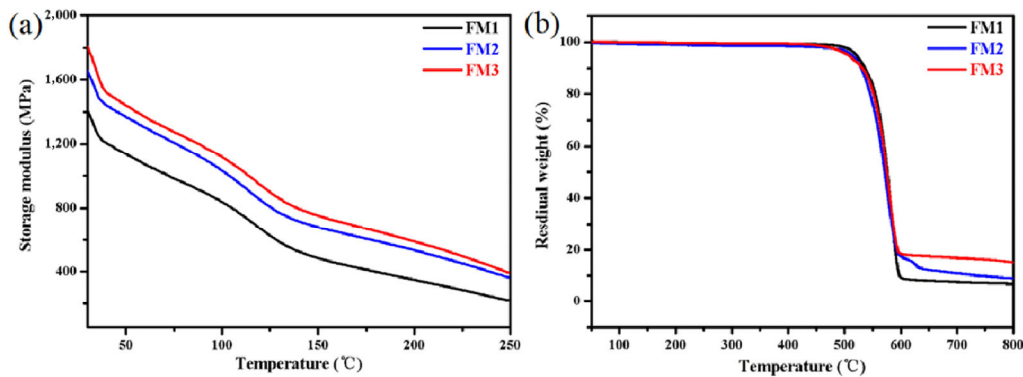


Fig. 6 Thermomechanical and thermal gravimetric analysis of PTFE composites: (a) the temperature-dependent storage modulus, and (b) TGA thermograms.

Table 3 Data of thermal properties of PTFE composites.

Composites	$T_{5\%}$ (°C)	$T_{10\%}$ (°C)	Char yield (%) at 800 °C
FM1	521.15	536.48	6.07
FM2	513.37	529.55	8.83
FM3	506.68	531.66	14.32

3.3 Friction behavior of PTFE composites

3.3.1 Effect of sliding speed and normal load on friction coefficient

Figure 7 presents the friction coefficient of composites with different compositions as a function of different sliding speeds and normal loads. Figure 7(a) shows that the friction coefficient first increases and then decreases with faster sliding speed for FM1 and FM2. However, for sample FM3, this behavior is not evident since the friction coefficients at 0.08 and 0.12 m/s show no significant difference. It is obvious that sliding speeds have little influence on the friction coefficient of FM3. The variation in friction coefficient can be explained by the higher interfacial temperature induced by frictional heat in Fig. 7(a). The elevated temperature

may cause a tribo-chemical reaction or degradation of polymers, which consumes vast amounts of energy and produces two opposite effects. On the one hand, polymer composites deform easily at high temperature owing to the reduction of elastic modulus. Consequently, the friction coefficient increases with the increase in real contact area. On the other hand, the molecules of polymer surfaces are pressed, drawn, and sheared owing to the mechanical force, and frictional heat leads to the relaxation of molecular chains. Therefore, the shear strength of composites decreases with increased temperature, which results in the decreased friction coefficient [37, 38]. To summarize, when sliding speed is low, deformation of materials is dominant. The reduction in shear strength plays a key role. Figure 7(b) shows that friction coefficient continuously decreases with increasing load for all samples; this variation trend can be explained by following equation [39, 40]:

$$\mu = kN^{n-1} \quad (3)$$

where μ and N are the friction coefficient and normal

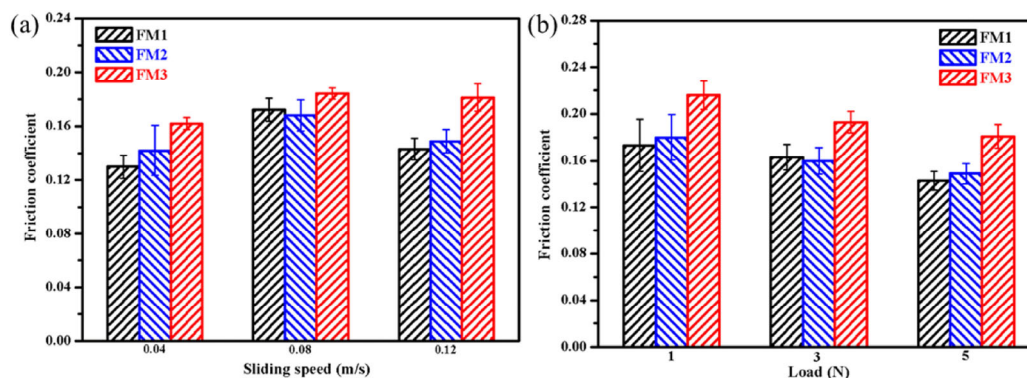


Fig. 7 Variation of friction coefficient with (a) sliding speed at 5 N, and (b) load at 0.12 m/s.

load, respectively; k and n are constants ($2/3 < n < 1$). Accordingly, the friction coefficient is inversely proportional to the load.

3.3.2 Effect of sliding speed and normal load on specific wear rate

The changes in specific wear rate of PTFE composites under different sliding speeds and normal loads are plotted in Fig. 8. It is worth noting that the incorporation of $g\text{-C}_3\text{N}_4$ enhances the wear resistance of composites under a wide range of operating conditions. However, the wear of composites increased with addition of MoS_2 . Similar experimental results were obtained by Zhang et al. [41, 42] and Wang et al. [43], which were attributed to partial oxidation of MoS_2 into MoO_3 , which contributes negatively to the anti-wear properties of composites. Detailed analysis is carried out in the next section. The decrease in wear resistance

with increasing speed for all composites (see Fig. 8(a)) can be explained by the fact that the increase in interfacial temperature at higher speed weakens the adhesion between the resin matrix and fillers. The influence of load on the specific wear rate can be seen in Fig. 8(b). The wear of composites gradually decreases with the increase in load. This may be because some large particles and debris are crushed into small particles or flakes on the frictional interface as pressure increased, which prevented direct contact between friction pairs and enhanced wear resistance effectively [44]. It was found that the anti-wear performance of $g\text{-C}_3\text{N}_4$ -reinforced composites (FM3) is better than composites with traditional fillers.

3.4 Wear behavior of PTFE composites

Figure 9 illustrates the 3D morphologies of the wear

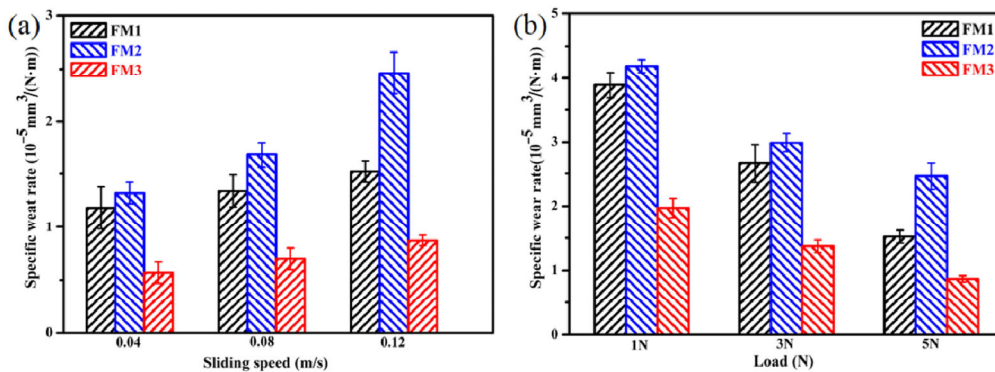


Fig. 8 Variation of specific wear rate with (a) sliding speed at 5 N, and (b) load at 0.12 m/s.

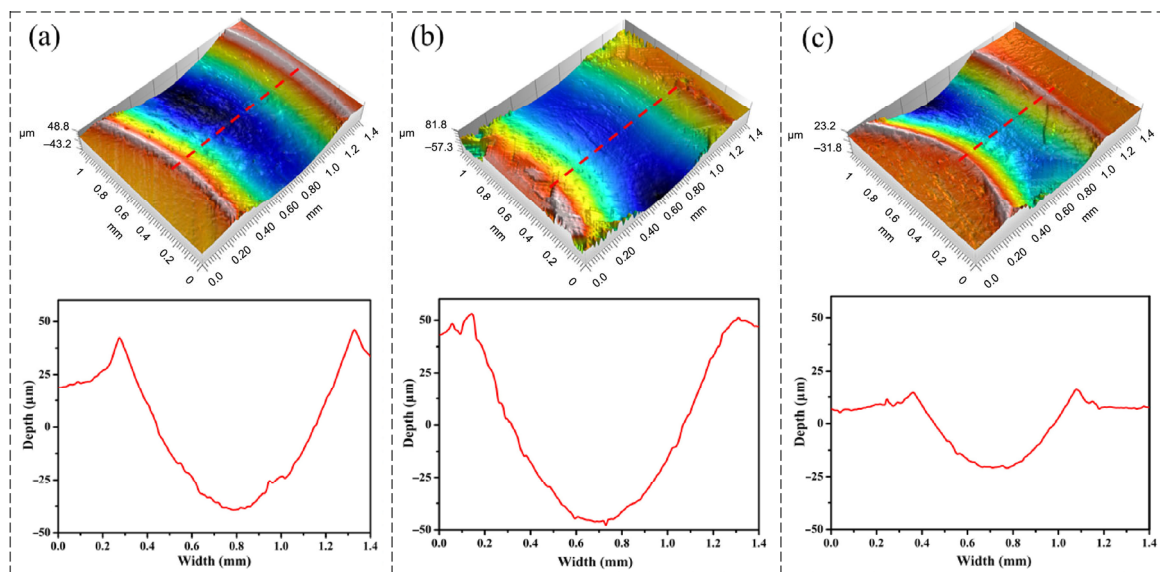


Fig. 9 3D images of grinding marks of worn surfaces: (a) FM1, (b) FM2, and (c) FM3.

tracks at 5 N and 0.12 m/s. All SEM images (Fig. 10) of the worn surface were obtained at the same condition. It is clear that the wear tracks of FM1 and FM2 are much deeper and wider compared with that of FM3, as seen in Fig. 9, which is consistent with wear resistance presented in the preceding Section 3.3. The different anti-wear properties with respect to the different fillers can also be obtained from worn surfaces (see Fig. 10). The severe wear and rough surface can be observed from delaminated worn surfaces, indicating that matrix materials were gradually peeled off (see white circle) owing to thermal softening. Therefore, the main wear mechanism is adhesive and fatigue wear, as shown in the high magnification micrograph (Figs. 10(a₁) and 10(b₁)). In the case of FM2, the specimen is worn because of the partial oxidation of composites. Under same service condition, oxidation reduces the ability to lubricate, resulting in more severe wear damage (see Figs. 10(a₂) and 10(b₂)). Adhesive wear is also the main wear mechanism as shown by the white rectangle in Fig. 10(b₂). In addition, a lot of wear debris (blue circle) can be seen from the worn surface; these have a detrimental effect on wear resistance and transfer film formation. However, the worn surface

morphology of FM3 is distinctly different. In this case, a much smoother surface is seen, in which the delamination of materials is mitigated significantly. This morphology coincides well with its superior wear resistance. The enhanced anti-wear performance by g-C₃N₄ can be attributed to the high load-bearing capacity and mechanical properties [45].

It is generally believed that transfer films are closely related to the anti-wear performance of polymer/steel tribo-systems [46]. Generally, homogeneous transfer and smooth transfer films are good for high wear resistance owing to the prevention of direct contact between frictional surfaces [47, 48]. Figure 11 shows transfer films formed on counterpart surfaces rubbing against different composites. Figures 11(a) and 11(b) show inhomogeneous and rough transfer films of FM1 and FM2, which are consistent with poor wear resistance (Figs. 9(a) and 9(b)). Contrary to the above, uniform, thin, and continuous transfer film is formed from FM3 (see Fig. 11(c)), which indicates the beneficial role of g-C₃N₄ in helping develop high-quality transfer films that strengthen the wear resistance of PTFE composites.

3.5 The evolution of tribochemistry and transfer films

MoS₂ is a well-known 2D material, a common solid lubricant with weak van der Waals force between planes. However, MoS₂ is easily oxidized into MoO₃ under oxygen-rich and elevated temperature conditions, as expressed by Eq. (4). MoO₃ has inferior lubricating performance and expands as temperature rises, leading to negative anti-wear effects/properties [49]. Therefore, tribo-oxidation of MoS₂ was explored by XPS spectra. The Mo 3d and S 2p XPS spectra on the worn surface of MoS₂-reinforced composites at 0.12 m/s and 5 N are shown in Figs. 12(a) and 12(b). In the Mo 3d spectrum, different peaks are exhibited at approximately 233.2 and 229.6 eV for Mo 3d_{3/2} and 3d_{5/2} in MoS₂ [50]. The peaks at 235.8 and 232.6 eV are assigned to the Mo 3d_{3/2} and 3d_{5/2} in MoO₃. Figure 12(b) shows a doublet at 162 and 163.2 eV corresponding to the S 2p_{3/2} and 2p_{1/2} peaks of S²⁻ in MoS₂. In addition, the peak at 168.3 eV is consistent with SO₄²⁻ [51, 52]. According to above analysis, partial oxidation of MoS₂ into MoO₃ occurred, which leads to poor wear resistance.

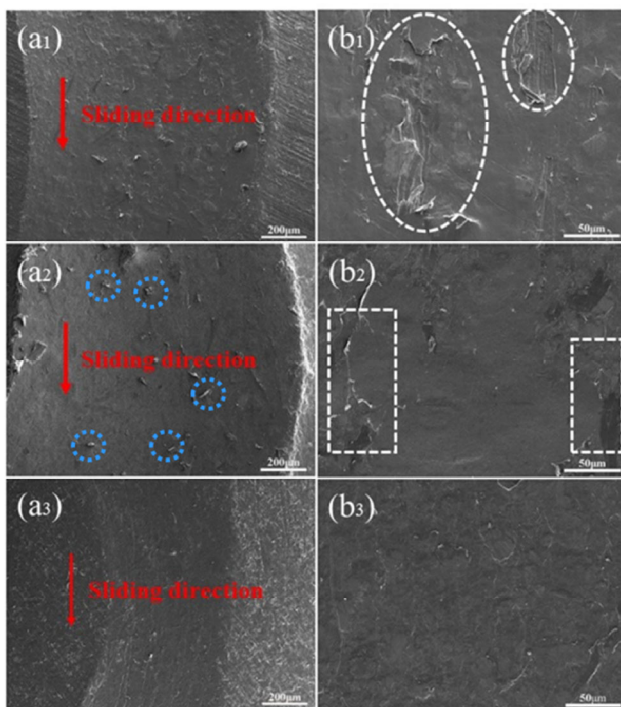


Fig. 10 SEM images of the worn surface of low and high magnification micrographs: FM1 (a₁, b₁), FM2 (a₂, b₂), and FM3 (a₃, b₃).

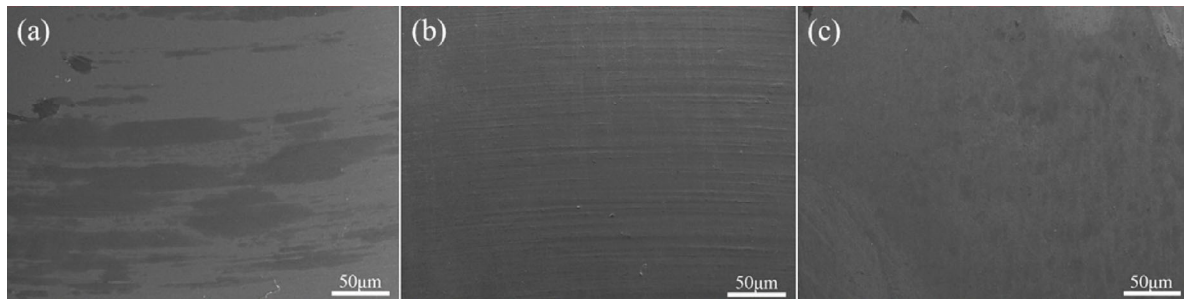


Fig. 11 Morphologies of the transfer films of counterparts: (a) FM1, (b) FM2, and (c) FM3.

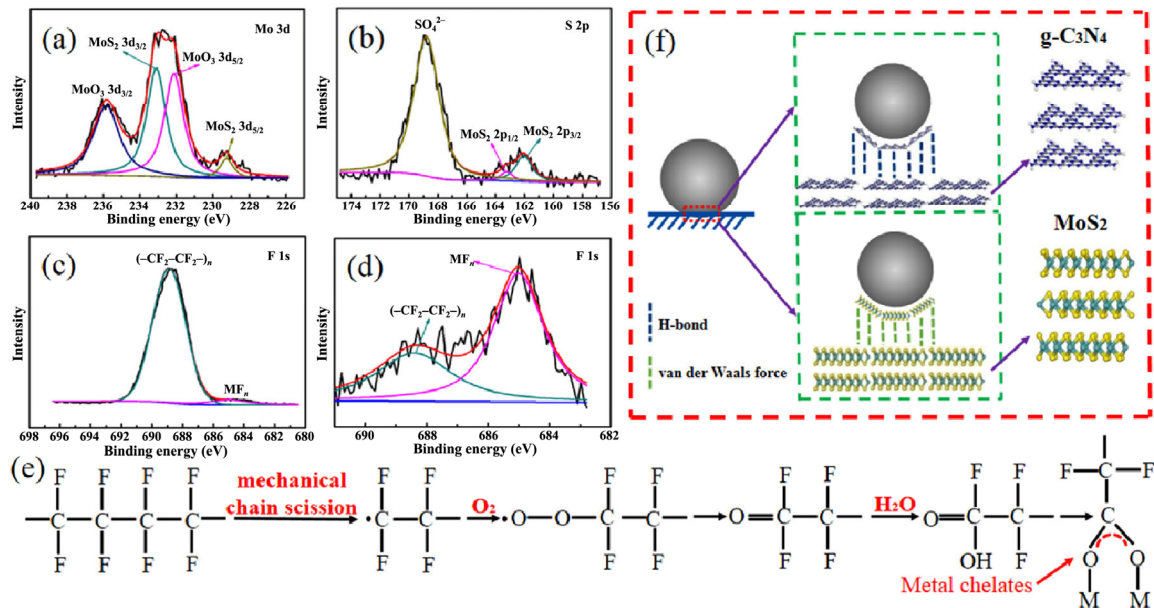
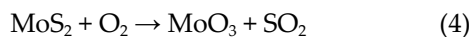


Fig. 12 XPS spectra of (a) Mo 3d, (b) S 2p, (c) F 1s on the worm surface, and (d) F 1s on metallic counterpart. The evolution of (e) tribochemistry and (f) transfer films.



Figures 12(c) and 12(d) show the F 1s spectra of composites and counterparts, respectively. It can be seen that metal fluorides were developed with the peak at approximately 685 eV [53]. When the PTFE composites rubbed against the metal counterparts, transfer films were formed owing to the easy shear of PTFE lamellae. Tribochemical reaction, mechanical compression, tension, and shear play a crucial role in transfer film formation. According to Refs. [54, 55], Figs. 12(e) and 12(f) show the evolution of tribochemistry and transfer films, respectively. During sliding, chain breakdown of the $-\text{C}-\text{C}-$ or $-\text{C}-\text{F}-$ backbone bond of PTFE can occur. Then, carboxylic acid end-groups chelate to the metal, developing robust transfer films. Extremely complex tribochemical reactions take place

on the frictional surfaces of polymer and metal, which form transfer films that reduce the wear rate of materials, often drastically.

4 Conclusions

In this paper, the effect of different layered materials (MoS_2 and $\text{g-C}_3\text{N}_4$) on thermal stability and thermomechanical properties were investigated. Meanwhile, the tribological properties of PTFE composites were also investigated at different sliding speeds and normal loads. According to the results, the following main conclusions can be drawn:

1. The two types of lamellar materials have a positive effect on thermal stability and thermomechanical properties. The char yield value and storage modulus can be significantly increased by incorporating $\text{g-C}_3\text{N}_4$.

2. Tribological properties of composites are improved by modification with $g\text{-C}_3\text{N}_4$, which has enormous potential as a functional filler. At selected load and speed, $g\text{-C}_3\text{N}_4$ -reinforced PTFE composites greatly enhance wear resistance.

3. The excellent thermal stability and tribological and thermomechanical properties of PTFE composites with the addition of $g\text{-C}_3\text{N}_4$ are highly dependent on the bonding strength of H-bonds, which is stronger than the van der Waals force of MoS_2 .

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