**ORIGINAL ARTICLE** 



# Arthrinins E–G, Three Botryane Sesquiterpenoids from the Plant Endophytic Fungus *Arthrinium* sp. HS66

Xiao-Zheng Su<sup>1,2,3</sup> · Jian-Wei Tang<sup>1,2</sup> · Kun Hu<sup>1,2</sup> · Xiao-Nian Li<sup>1,2</sup> · Han-Dong Sun<sup>1,2</sup> · Pema-Tenzin Puno<sup>1,2</sup>

Received: 20 April 2020 / Accepted: 31 May 2020 / Published online: 12 July 2020  $\circledcirc$  The Author(s) 2020

#### Abstract

Arthrinins E–G (1-3), three new sesquiterpenoids possessing non-isoprenoid botryane skeleton, were isolated from the fermentation of an endophytic fungus named *Arthrinium* sp. HS66 which colonized in the stems of *Isodon xerophilus*. Their structures were determined by extensive spectroscopic methods. Furthermore, the structure of **1** was unambiguously confirmed by X-ray diffraction, while those of **2** and **3** were verified through quantum chemical calculation of NMR data and ECD spectra.

Keywords Isodon xerophilus · Endophytic fungus · Arthrinium · Botryane sesquiterpenoid · Quantum chemical calculation

## 1 Introduction

Fungal metabolites have received considerable attention nowadays because their diverse chemical structures and bioactivities greatly facilitate the progress of drug discovery [1]. Botryanes are a class of fungus-derived sesquiterpene metabolites featuring characteristic bicyclic non-isoprenoid system [2–4], and they were found to possess broad spectrum of biological activities, such as cytotoxicity[5, 6], phytotoxicity [7, 8], and antimicrobial property [9, 10]. Furthermore, fascinating by the botryane sesquiterpenoids, researchers have conducted plenty of in-depth investigations

**Electronic supplementary material** The online version of this article (doi:https://doi.org/10.1007/s13659-020-00248-y) contains supplementary material, which is available to authorized users.

Pema-Tenzin Puno punopematenzin@mail.kib.ac.cn

- <sup>1</sup> State Key Laboratory of Phytochemistry and Plant Resources in West China, Kunming Institute of Botany, Chinese Academy of Sciences, Kunming 650201, People's Republic of China
- <sup>2</sup> Yunnan Key Laboratory of Natural Medicinal Chemistry, Kunming Institute of Botany, Chinese Academy of Sciences, Kunming 650201, People's Republic of China
- <sup>3</sup> University of Chinese Academy of Sciences, Beijing 100049, People's Republic of China

about their structure–activity relationships [11], synthesis [12, 13] and biosynthetic pathway [14–16].

Fungal endophytes have become an important source for discovering structurally novel and biologically active secondary metabolites [17]. Over the past several years, our groups have made great efforts to study the secondary metabolites from endophytic fungi inhabiting the Isodon species. As a result, isopenicin A, a potent inhibitor of Wnt signaling [18], as well as several antineoplastic compounds like phomopchalasins A and B [19] have been successfully obtained. In the present research, an endophytic fungus colonizing in the stems of Isodon xerophilus was discovered and identified as Arthrinium sp. HS66. Subsequent large fermentation and chemical investigation on this strain resulted in the isolation of three new botryane sesquiterpenoids named arthrinins E-G (1-3). Notably, compound 2 possesses uncommon 15-nor-botryane skeleton. Herein, details of the isolation, structure elucidation, and cytotoxicity of these compounds were reported (Fig. 1).

# 2 Results and Discussion

Compound 1 was isolated as colorless oil, and it was assigned a molecular formula of  $C_{15}H_{26}O_3$  according to its positive HRESIMS ion peak at m/z 277.1775 ([M+Na]<sup>+</sup>, calcd for 277.1774), which required three degrees of unsaturation. The IR spectrum showed the presence of hydroxyl (3374 cm<sup>-1</sup>) group. The <sup>1</sup>H NMR data (Table 1) exhibited

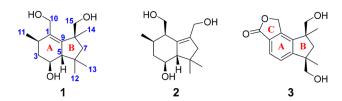


Fig. 1 Chemical structures of compounds 1–3

Table 1 <sup>1</sup>H NMR data ( $\delta$  in ppm, J in Hz) of compounds 1–3

No.	1 <sup>a</sup>	2 <sup>a</sup>	3 <sup>b</sup>
	$\delta_{\rm H}$ , mult ( <i>J</i> )	$\delta_{\rm H}$ , mult ( <i>J</i> )	$\delta_{\mathrm{H}}$ , mult ( <i>J</i> )
1		2.60 (dt, 10.2, 4.4)	
2	2.58 (m)	1.67 (overlap)	
3	β 1.97 (ddd, 12.0, 5.8, 3.6)	β 1.31 (m)	7.72 (d, 7.9)
	$\alpha$ 1.29 (overlap)	$\alpha$ 1.67 (overlap)	
4	3.63 (ddd, 11.9, 9.6, 3.6)	3.57 (td, 10.8, 4.5)	7.43 (d, 7.9)
5	2.10 (ddd, 9.6, 3.5, 2.1)	2.30 (d, 10.8)	
6			
7	β 1.74 (d, 13.0)	β 2.37 (dd, 15.6,2.3)	β 2.36 (d, 13.9)
	$\alpha$ 1.30 (overlap)	α 2.16 (dd, 15.6, 2.3)	α 1.70 (d, 13.9)
8			
9			
10	4.33 (d, 11.9)	3.69 (dd, 10.2, 4.4)	5.55 (d, 15.7)
	4.04 (dd, 11.9, 2.1)	3.45 (t, 10.2)	5.46 (d, 15.7)
11	1.10 (d, 6.9)	1.03 (d, 7.0)	
12	0.97 (s)	1.07 (s)	3.61 (d, 10.8)
			3.58 (d, 10.8)
13	1.18 (s)	1.23 (s)	1.34 (s)
14	1.38 (s)	4.05 (s)	1.37 (s)
15	3.30 (overlap)		3.64 (s)

<sup>a</sup>Recorded at 600 MHz, Recorded in CD<sub>3</sub>OD <sup>b</sup>Recorded at 500 MHz, Recorded in CD<sub>3</sub>OD

the existence of three singlet methyls ( $\delta_{\rm H}$  0.97, 1.18 and 1.38), one doublet methyl ( $\delta_{\rm H}$  1.10, d, J = 6.9 Hz), a pair of nonequivalent oxygenated methylene protons at  $\delta_{\rm H}$  4.04 (dd, J = 11.9, 2.1 Hz)/4.33 (d, J = 11.9 Hz), one hydroxylated methylene signal at  $\delta_{\rm H}$  3.30 (overlap), and one hydroxylated methine proton at  $\delta_{\rm H}$  3.63 (ddd, J = 11.9, 9.6, 3.6 Hz). The analysis of its <sup>13</sup>C NMR and DEPT spectra revealed 15 carbon resonances which were assigned as four methyls, four  $sp^3$  methylenes, three  $sp^3$  methines, two  $sp^3$  quaternary carbons and two olefinic carbons (Table 2). These data suggested that compound **1** might be a botryane sesquiterpenoid [20].

Specifically, a structural subunit of C-11/C-2/C-3/C-4/C-5 could be established as an isolated spin-system according to the <sup>1</sup>H-<sup>1</sup>H COSY correlations of H<sub>3</sub>-11/H-2/

<b>Table 2</b> <sup>13</sup> C NMR data ( $\delta$ in ppm) of compound	nds <b>1–3</b>
--	----------------

No.	1 <sup>a</sup>	2 <sup>a</sup>	3 <sup>b</sup>
	$\delta_{\rm C}$ , type	$\delta_{\rm C}$ , type	$\delta_{\rm C}$ , type
1	136.0, C	43.5, CH	145.2, C
2	34.4, CH	33.7, CH	126.0, C
3	44.1, CH <sub>2</sub>	40.8, CH <sub>2</sub>	125.4, CH
4	70.0, CH	71.9, CH	126.1, CH
5	61.4, CH	60.9, CH	157.8, C
6	40.0, C	39.7, C	50.1, C
7	56.7, CH <sub>2</sub>	51.7, CH <sub>2</sub>	47.8, CH <sub>2</sub>
8	47.5, C	136.9, C	50.2, C
9	145.9, C	139.8, C	144.7, C
10	58.9, CH <sub>2</sub>	59.9, CH <sub>2</sub>	71.1, CH <sub>2</sub>
11	20.7, CH <sub>3</sub>	19.4, CH <sub>3</sub>	173.9, C
12	23.9, CH <sub>3</sub>	25.6, CH <sub>3</sub>	71.3, CH <sub>2</sub>
13	30.9, CH <sub>3</sub>	31.6, CH <sub>3</sub>	27.2, CH <sub>3</sub>
14	26.1, CH <sub>3</sub>	59.2, CH <sub>2</sub>	24.9, CH <sub>3</sub>
15	72.6, CH <sub>2</sub>		70.8, CH <sub>2</sub>

<sup>a</sup>Recorded at 150 MHz, Recorded in CD<sub>3</sub>OD

<sup>b</sup>Recorded at 125 MHz, Recorded in CD<sub>3</sub>OD

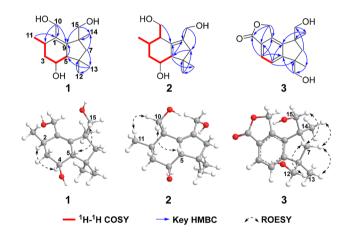


Fig. 2 Key 2D NMR correlations of compounds 1–3

H<sub>2</sub>-3/H-4/H-5. Then, the subunit can be assigned to ring A through the HMBC correlations from H-5 to C-9 ( $\delta_{\rm C}$  145.9), from H<sub>2</sub>-10 to C-1 ( $\delta_{\rm C}$  136.0), C-2 ( $\delta_{\rm C}$  34.4) and C-9, from H<sub>3</sub>-11 to C-1. Meanwhile, the HMBC correlations from H<sub>2</sub>-7 to C-13 ( $\delta_{\rm C}$  30.9) and C-14 ( $\delta_{\rm C}$  26.1), from H<sub>3</sub>-12 to C-6 ( $\delta_{\rm C}$  40.0), from H<sub>3</sub>-13 to C-5 ( $\delta_{\rm C}$  61.4) and C-12 ( $\delta_{\rm C}$  23.9), from H<sub>3</sub>-14 to C-8 ( $\delta_{\rm C}$  47.5), from H<sub>2</sub>-15 to C-9 and C-14, in combination with the remaining one degree of hydrogen deficiency, successfully established the ring B as well as its connection with ring A (Fig. 2). Hence, the planar structure of compound **1** was resolved.

The relative configuration of **1** was deduced by analyses of ROESY and <sup>1</sup>H NMR data. Randomly assigning H-2 as  $\alpha$ -oriented, the cross peaks of H-4/H-2 and H-5/H<sub>2</sub>-15, together with the coupling constant (J = 9.6 Hz) between H-4 and H-5 indicated that both H-4 and CH<sub>3</sub>-14 adopted  $\alpha$ -orientation, while both H-5 and CH<sub>2</sub>OH-15 adopted  $\beta$ -orientation (Fig. 2). Fortunately, colorless square crystals of **1** were eventually obtained through slow evaporation of methanol. Then a single-crystal X-ray diffraction experiment with Cu K $\alpha$  radiation of **1** [Flack parameter = 0.07(3)] unambiguously verified the aforementioned deduction and assigned the absolute configuration of **1** as 2*R*,4*S*,5*S*, and 8*S* (Fig. 3). Arthrinins A–D from the sponge derived fungus *Arthrinium* sp. has been reported, so we gave compound **1** the trivial name from arthrinin E [21].

Compound 2 was obtained as colorless oil. Its molecular formula  $C_{14}H_{24}O_3$  was ascertained by the positive HRESIMS ion peak (*m*/*z* 263.1618, [M+Na]<sup>+</sup>, calcd for 263.1618), indicating three degrees of unsaturation. Comparison of the <sup>1</sup>H NMR spectrum of 2 with that of 1 disclosed that 2 was structurally analogous to compound 1 (Table 1). Furthermore, the <sup>13</sup>C NMR and DEPT spectra, which demonstrated 14 carbons resonances involving three methyls, four *sp*<sup>3</sup> methylenes, four *sp*<sup>3</sup> methines, one quaternary carbon and two olefinic carbons (Table 2), suggested that the structure of **2** was highly similar to that of boledulin C [22], a 15-nor-botryane sesquiterpenoid.

More credible evidence was obtained from 2D NMR spectra (Figs. S20-23) of **2**. On one hand, the structure of ring A was established by the <sup>1</sup>H-<sup>1</sup>H COSY correlations of H<sub>2</sub>-10/H-1/H-2/H<sub>2</sub>-3/H-4/H-5 and H-2/H<sub>3</sub>-11 and HMBC correlations from H-5 to C-9 ( $\delta_C$  139.8), from H<sub>2</sub>-10 to C-9. On the other hand, the HMBC correlations from H-5 to C-8 ( $\delta_C$  136.9) and C-13 ( $\delta_C$  31.6), from H<sub>2</sub>-7 to C-5 ( $\delta_C$  60.9), from H<sub>3</sub>-12 to C-6 ( $\delta_C$  39.7) and C-7 ( $\delta_C$  51.7), from H<sub>3</sub>-13 to C-12 ( $\delta_C$  25.6), from H<sub>2</sub>-14 to C-7 and C-8 defined the ring B fragment and planar structure of **2** ultimately (Fig. 2).

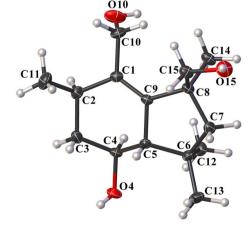
Fig. 3 X-ray crystallographic structure of 1

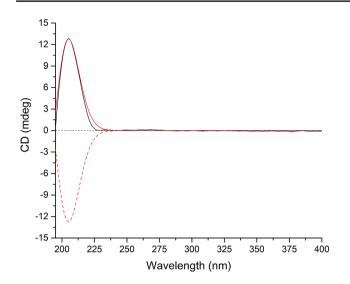
As for the configuration of 2, randomly assigning H-5 as  $\beta$ -oriented, the observed correlations of H<sub>2</sub>-10/H<sub>2</sub>-11/H-5 in the ROESY spectrum, together with the coupling constant (J = 4.4 Hz) between H-1 and H-2 implied that both CH<sub>2</sub>OH-10 and CH<sub>3</sub>-11 adopted  $\beta$ -orientation, while the analysis of coupling constant (J = 10.8 Hz) between H-4 and H-5 suggested that H-4 was  $\alpha$ -oriented (Fig. 2). Then,  $(1R^*, 2R^*, 4S^*, 5S^*)$ -2 was subjected to quantum chemical calculation of NMR chemical shifts and spin-spin coupling constants at mPW1PW91-SCRF/6-31+G(d,p)//B3LYP-D3BJ-SCRF/6-31G(d) and B972-SCRF/pcJ-1//B3LYP-D3BJ-SCRF/6-31G(d) level of theory (both with methanol as solvent and SMD solvent model), respectively, and the calculated results matched their experimental counterparts very well (Tables 3 and S3), which verified the established planar structure and relative configuration of 2. Moreover, TDDFT ECD calculation of (1R,2R,4S,5S)-2 was run at CAM-B3LYP-SCRF/def2-SVP//B3LYP-D3BJ-SCRF/6-31G(d) level of theory in MeOH with SMD solvent model, and the obtained curve, which is entirely consistent with its experimental counterpart, supported the absolute configuration of 2 to be 1R, 2R, 4S, and 5S (Fig. 4). Therefore, compound 2 was confirmed to possess 15-nor-botryane skeleton and named as arthrinin F.

Compound 3 was isolated as a colorless oil. The [M+Na]<sup>+</sup> peak at m/z 285.1100 (calcd for 285.1097) existing in the positive HRESIMS spectrum revealed the molecular formula of **3** to be  $C_{15}H_{18}O_4$ , which implied seven indices of hydrogen deficiency. Its IR spectrum indicated the presence of hydroxyl ( $3419 \text{ cm}^{-1}$ ), methyl ( $2932 \text{ and } 2869 \text{ cm}^{-1}$ ), phenyl (1607 and 1467 cm<sup>-1</sup>) and ester (1743 and 1040 cm<sup>-1</sup>) groups. The <sup>1</sup>H NMR spectrum gave two characteristic aromatic protons at  $\delta_{\rm H}$  7.43 (d, J = 7.9 Hz), 7.72 (d, J = 7.9 Hz) and two singlet methyl groups ( $\delta_{\rm H}$  1.37, 1.34) (Table 1). The <sup>13</sup>C NMR and DEPT spectra displayed 15 carbon signals, including one ester carbonyl carbon, six aromatic carbons, two  $sp^3$  quaternary carbons, four  $sp^3$  methylenes and two methyls (Table 2). The above data suggested that the structure of 3 resembled to that of dehydrobotrylactone with a botryane scaffold [23].

Table 3 Analyses of the NMR computation results of  $(1R^*, 2R^*, 4S^*, 5S^*)$ -2 and  $(6R^*, 8S^*)$ -3

Parameters	$\mathbb{R}^2$	MAE (ppm)	CMAE (ppm)
$(1R^*, 2R^*, 4S^*,$	5 <i>S</i> *)- <b>2</b>		
<sup>13</sup> C	0.9992	1.3	0.8
$^{1}H$	0.9938	0.1	0.06
(6 <i>R</i> *,8 <i>S</i> *)- <b>3</b>			
<sup>13</sup> C	0.9989	1.5	1.4
$^{1}\mathrm{H}$	0.9948	0.21	0.13

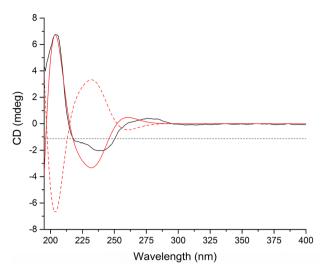




**Fig. 4** Experimental ECD spectrum of **2** (black); Calculated ECD spectra of (1R,2R,4S,5S)-**2** (shift=12.5 nm, red) and (1S,2S,4R,5R)-**2** (shift=12.5 nm, red dash)

The <sup>1</sup>H-<sup>1</sup>H COSY correlation between H-3 and H-4, along with the HMBC correlations from H-3 to C-1 ( $\delta_{\rm C}$ 145.2), C-5 ( $\delta_{\rm C}$  157.8) and C-11 ( $\delta_{\rm C}$  173.9), from H-4 to C-2 ( $\delta_{\rm C}$  126.0) and C-9 ( $\delta_{\rm C}$  144.7), from H<sub>2</sub>-10 to C-1, C-2 and C-11 established the structure of the butyrolactone motif (ring C), and its connection to a tetra-substituted benzene ring (ring A). Moreover, the HMBC cross peaks from H-4 to C-6 ( $\delta_{\rm C}$  50.1), from H<sub>2</sub>-7 to C-13 ( $\delta_{\rm C}$  27.2) and C-15 ( $\delta_{\rm C}$ 70.8), from H<sub>2</sub>-12 to C-5 and C-6, from H<sub>3</sub>-13 to C-12 ( $\delta_{\rm C}$ 71.3), from H<sub>3</sub>-14 to C-8 ( $\delta_{\rm C}$  50.2) and C-9, from H<sub>2</sub>-15 to C-14 ( $\delta_{\rm C}$  24.9) determined the existence of a cyclopentane fragment (ring B) and its linkage with ring A. As a result, the planar structure of compound **3** was elucidated as shown in Fig. 2.

As for the stereochemistry of **3**, randomly assigning H-7a as  $\alpha$ -oriented, according to the ROESY correlations observed from H-7 $\alpha$  to both H<sub>3</sub>-13 and H<sub>3</sub>-14, as well as from H-7 $\beta$  to both H<sub>2</sub>-12 and H<sub>2</sub>-15, it can be concluded that both CH<sub>3</sub>-13 and CH<sub>3</sub>-14 adopted  $\alpha$ -orientation. Then, the NMR chemical shifts of  $(6R^*, 8S^*)$ -3 were calculated at mPW1PW91-SCRF/6-31+G(d,p)//B3LYP-D3BJ-SCRF/6-31G(d) level of theory with methanol as solvent and SMD solvent model, the predicted chemical shifts matched their experimental counterparts very well (Table 3), which supported the above deduction concerning the relative configuration of 3. Furthermore, the theoretical ECD spectrum of (6R,8S)-3 was obtained at CAM-B3LYP-SCRF/def2-SVP// B3LYP-D3BJ-SCRF/6-31G(d) level of theory in MeOH with SMD solvent model, and the calculated curve matched the experimental one very well and thus supported the absolute configuration of 3 to be 6R,8S. (Fig. 5). Compound 3was given the trivial name arthrinin G.



**Fig. 5** Experimental ECD spectrum of **3** (black); Calculated ECD spectra of (6R,8S)-**3** (shift = 12 nm, red) and (6S,8R)-**3** (shift = 12 nm, red dash)

Additionally, compounds **1–3** were evaluated for their cytotoxicity against five human cancer cell lines (HL-60, A549, SMMC-7721, MCF-7, SW480), with *cis*-platin and paclitaxel as positive controls, however, no compounds showed activity against the tested cell lines (Fig. S2).

## 3 Experimental

#### 3.1 General Experimental Procedures

Optical rotations were measured with a JASCO P-1020 polarimeter. UV spectra were obtained using a Shimadzu UV-2401 PC spectrophotometer. A Tensor 27 spectrophotometer was used for scanning IR spectroscopy with KBr pellets. 1D and 2D NMR spectra were recorded on Bruker DRX-600 and 500 spectrometers with TMS as internal standard. Chemical shifts ( $\delta$ ) are expressed in parts per million (ppm) with reference to the solvent signals. HRESIMS was performed on an API QSTAR spectrometer. Semipreparative HPLC was performed on an Agilent 1200 liquid chromatograph with a Zorbax SB-C18 (9.4 mm × 250 mm) column. LC-MS/MS was performed on Agilent 6530 Accurate-Mass Q-TOF spectrometer coupled to an Agilent 1290 LC system with Zorbax SB-C18 (9.4 mm × 250 mm) column. Column chromatography was performed with silica gel (100-200 mesh, Oingdao Marine Chemical, Inc., Qingdao, People's Republic of China). Fractions were monitored by TLC, and spots were visualized by heating silica gel plates sprayed with 10% H<sub>2</sub>SO<sub>4</sub> in EtOH.

#### 3.2 Fungal Material, Identification and Fermentation

The fungal strain of Arthrinium sp. HS66 was isolated from the fresh stems of Isodon xerophilus that was collected from Kunming Botanical Garden, Kunming City, Yunnan Province, People's Republic of China, in August 2018. The isolate was identified based on sequence (GenBank Accession No. MT355097) analysis of the ITS region of the rDNA. The fungal strain was cultured on slants of potato dextrose agar at 28 °C for 7 days. Agar plugs were cut into small pieces (about  $0.5 \times 0.5 \times 0.5$  cm<sup>3</sup>) under aseptic conditions, and 15 pieces were used to inoculate three Erlenmeyer flasks (500 mL), each containing 200 mL of media (0.4% glucose, 1% malt extract, and 0.4% yeast extract); the final pH of the media was adjusted to 7.0, and the flasks were sterilized by autoclave. Three flasks of the inoculated media were incubated at 28 °C on a rotary shaker at 170 rpm for 7 days to prepare the seed culture. The detailed lager fermentation procedure was as following: Fermentation was carried out on solid rice medium in 125 Fernbach flasks (500 mL, 90 mL distilled water was added to 80 g rice and kept overnight before autoclaving). Each flask was inoculated with 5.0 mL of the spore inoculum and incubated for 30 days at 28 °C in a static incubator.

#### 3.3 Cytotoxicity Assay

Five human cancer cell lines, human myeloid leukemia HL-60, lung cancer A-549 cells, hepatocellular carcinoma SMMC-7721, breast cancer MCF-7, and colon cancer SW480, were purchased from the Shanghai Institute of Biochemistry and Cell Biology, Chinese Academy of Sciences (Shanghai, China). Cells were cultured according to the manufacturer' recommendations. All mediums were supplemented with 10% fetal bovine serum (FBS), 100 units/ml penicillin G sodium and 100 µg/ml streptomycin (HyClone). All the cells were incubated at 37 °C, 5% CO<sub>2</sub> in a humidified atmosphere. Cytotoxicity of compounds was determined by MTS method. Briefly,  $5 \times 10^3$  cells were plated in 96-well plates 12 h before treatment and continuously exposed to test compounds for 48 h. Then MTS (Promega) was added to each well. The samples were incubated at 37 °C for 1–4 h and the optical density (OD) was measured at 490 nm using a microplate reader (Bio-Rad Laboratories). The IC<sub>50</sub> values were calculated by Reed and Muench's method [24].

#### 3.4 Extraction and Isolation of Compounds 1–3

The culture medium was overlaid and extracted with MeOH by maceration. With filtration and concentration, the resultant extract was partitioned with EtOAc. Then the solvent was evaporated in vacuo to afford a crude extract (130 g).

The extraction was subjected to column chromatography on silica gel with a CHCl<sub>3</sub>/Me<sub>2</sub>CO gradient system (1:0, 9:1, 8:2, 7:3, 6:4, 1;1, 0:1) to yield seven fractions, A-G. Fraction C (CHCl<sub>3</sub>/Me<sub>2</sub>CO 8:2, 9 g) was chromatographed on a RP-18 column with a methanol/ H<sub>2</sub>O gradient system (from 30:70 to 100:0) to afford fractions C1-C8. Fraction C2 (methanol/ H<sub>2</sub>O, 40:60, 1.4 g) was subjected to chromatography over silica gel (chloroform/ Me<sub>2</sub>CO, from 80:1 to 0:1) to yield subfractions C2/1–11, subfraction C2-9 was purified by semipreparative HPLC (3 ml/min, detector UV  $\lambda_{max} = 195$  nm, MeCN/H<sub>2</sub>O 22.5:77.5) to yield **1** (3.1 mg, t<sub>R</sub> = 18.7 min) and **2** (1.8 mg, t<sub>R</sub> = 20.2 min), subfraction C2-10 was purified by semipreparative HPLC (3 ml/min, detector UV  $\lambda_{max} = 195$  nm, MeCN/H<sub>2</sub>O 40:60) to yield **3** (3.6 mg, t<sub>R</sub> = 25.8 min).

### 3.5 Physical Constants and Spectroscopic Data of Compounds 1–3

Arthrinin E (1): initially obtained as a colorless oil, by using slow evaporation of methanol in a closed tube, the colorless square crystals were obtained eventually; mp: 135–140 °C;  $[\alpha]_D^{22.2}$ : + 66.6 (MeOH, *c* 0.100); ECD (MeOH)  $\lambda_{max}$  ( $\Delta \varepsilon$ ): 220 (0.02) nm; UV (MeOH)  $\lambda_{max}$  (log  $\varepsilon$ ): 204 (3.03) nm; IR ( $\nu_{max}$ ): 3374, 2927, 2869, 1630, 1607, 1460, 1384, 1366, 1093, 1057, 1026, 1010, 986, 971 cm<sup>-1</sup>. HRESIMS at *m/z* 277.1775 ([M+Na]<sup>+</sup>, calcd for 277.1774). <sup>1</sup>H and <sup>13</sup>C NMR data, see Tables 1 and 3.

Arthrinin F (2): obtained as a colorless oil;  $[\alpha]_{D}^{22.4}$ : +107.00 (MeOH, *c* 0.060); ECD (MeOH)  $\lambda_{max}$  ( $\Delta \varepsilon$ ): 205 (0.68) nm; UV (MeOH)  $\lambda_{max}$  (log  $\varepsilon$ ): 201 (3.14) nm; IR ( $\nu_{max}$ ): 3383, 2955, 2927, 2890, 2871, 1631, 1609, 1465, 1454, 1384, 1361, 1082, 1026, 994, 978 cm<sup>-1</sup>. HRESIMS at *m*/*z* 263.1618 ([M+Na]<sup>+</sup>, calcd for 263.1618). <sup>1</sup>H and <sup>13</sup>C NMR data, see Tables 1 and 3.

Arthrinin G (3): isolated as a colorless oil;  $[\alpha]_D^{22.8}$ : + 12.55 (MeOH, *c* 0.110); ECD (MeOH)  $\lambda_{max}$  ( $\Delta \varepsilon$ ): 204 (0.21), 240 (- 0.06), 276 (0.01) nm; UV (MeOH)  $\lambda_{max}$  (log  $\varepsilon$ ): 208 (3.22), 241 (2.74), 283 (1.98) nm; IR ( $\nu_{max}$ ): 3419, 2957, 2932, 2869, 1743, 1607, 1467, 1384, 1367, 1040, 1017, 744 cm<sup>-1</sup>. HRESIMS at *m*/*z* 285.1100 ([M+Na]<sup>+</sup>, calcd for 285.1097). <sup>1</sup>H and <sup>13</sup>C NMR data, see Tables 1 and 3.

Crystallographic data for the structures of arthrinin E (1, deposition number CCDC 1997674) has been deposited in the Cambridge Crystallographic Data Centre database. Copies of the data can be obtained free of charge from the CCDC at www.ccdc.cam.ac.uk.

Crystal data for **1**:  $C_{15}H_{26}O_3$ , M = 254.36, a = 8.8861(2)Å, b = 8.8861(2) Å, c = 31.7909(6) Å,  $\alpha = 90^{\circ}$ ,  $\beta = 90^{\circ}$ ,  $\gamma = 120^{\circ}$ , V = 2173.98(11) Å<sup>3</sup>, T = 100.(2) K, space group P3221, Z = 6,  $\mu$ (Cu K $\alpha$ ) = 0.629 mm<sup>-1</sup>, 29,180 reflections measured, 2838 independent reflections ( $R_{int} = 0.0291$ ). The final  $R_1$  values were 0.0351 ( $I > 2\sigma(I)$ ). The final  $wR(F^2)$  values were 0.0969  $(I > 2\sigma(I))$ . The final  $R_I$  values were 0.0353 (all data). The final  $wR(F^2)$  values were 0.0970 (all data). The goodness of fit on  $F^2$  was 1.142. Flack parameter = 0.07(3).

#### 3.6 Computational Method

Conformational searching of  $(1R^*, 2R^*, 4S^*, 5S^*)$ -2 and  $(6R^*, 8S^*)$ -3 were undertaken with the CREST code (version 2.8) using the default iMTD-GC procedure [25]. The first 20 conformers of (1R\*,2R\*,4S\*,5S\*)-2 and (6R\*,8S\*)-3 were subjected to DFT geometry optimization at B3LYP-D3BJ-SCRF/6-31G(d) level of theory (with MeOH as solvent and SMD solvent model). Frequency analyses of all optimized conformers were undertaken at the same level of theory to ensure that no imaginary frequency exists. Then, thermal correction to Gibbs free energies obtained by frequency analyses were added to the electronic energies obtained at B3LYP-D3BJ-SCRF/6-311 + G(d,p) level of theory (with MeOH as solvent and SMD solvent model) to get the Gibbs free energies of each conformer. Subsequently, Room-temperature (298.15 K) equilibrium populations were calculated according to Boltzmann distribution law:

$$p_i = \frac{n_i}{\sum_j n_j} = \frac{e^{-\Delta G_i/RT}}{\sum_j e^{-\Delta G_j/RT}}$$

where  $P_i$  is the population of the *ith* conformer;  $n_i$  the number of molecules in *ith* conformer;  $\Delta G$  is the relative Gibbs free energy (kcal/mol); *T* is room temperature (298.15 K) here; *R* is the ideal gas constant (0.0019858995). Those conformers with a population of over 2% were subjected to subsequent NMR and ECD calculations.

NMR shielding constants were calculated with the GIAO method at mPW1PW91-SCRF/6–31 + G(d,p) level (with MeOH and SMD solvent model). The obtained shielding constants were converted into chemical shifts by referencing to TMS at 0 ppm ( $\delta_{cal} = \sigma_{TMS} - \sigma_{cal}$ ), where the  $\sigma_{TMS}$  was the shielding constant of TMS calculated at the same level of theory. The parameters *a* and *b* of the linear regression  $\delta_{cal} = a \delta_{exp} + b$ ; the correlation coefficient,  $R^2$ ; the mean absolute error (MAE) defined as  $\sum_n |\delta_{cal} - \delta_{exp}|/n$ ; the corrected mean absolute error (CMAE) defined as  $\sum_n |\delta_{corr} - \delta_{exp}|/n$ , where  $\delta_{corr} = (\delta_{cal} - b)/a$  were calculated [26, 27]. Calculation of coupling constants were run at B972/pcJ-1 level of theory (with MeOH as solvent and SMD solvent model) [28].

TDDFT ECD calculations were run at CAM-B3LYP/ def2-SVP level of theory (with MeOH as solvent and SMD solvent model) [29]. For each conformer, 30 excited states were calculated. The calculated ECD curves were generated using the Multiwfn software (version 3.7) [30]. The geometry optimization, single-point energy calculation, NMR shielding constant calculation, coupling constant calculation, and TDDFT ECD calculation were all completed in Gaussian 09 program [31].

Acknowledgements This project was supported financially by the National Natural Science Foundation of China (No. 81874298), the CAS "Light of West China" program (Pema-Tenzin Puno), and the Yunnan Science Fund for Distinguished Young Scholars (2019FJ002).

#### **Compliance with Ethical Standards**

Conflict of interest The authors declare no conflict of interest.

**Open Access** This article is licensed under a Creative Commons Attribution 4.0 International License, which permits use, sharing, adaptation, distribution and reproduction in any medium or format, as long as you give appropriate credit to the original author(s) and the source, provide a link to the Creative Commons licence, and indicate if changes were made. The images or other third party material in this article are included in the article's Creative Commons licence, unless indicated otherwise in a credit line to the material. If material is not included in the article's Creative Commons licence and your intended use is not permitted by statutory regulation or exceeds the permitted use, you will need to obtain permission directly from the copyright holder. To view a copy of this licence, visit http://creativecommons.org/licenses/by/4.0/.

#### References

- M. Barbero, E. Artuso, C. Prandi, Curr. Med. Chem. 25, 141–185 (2018)
- H.W. Fehlhaber, R. Geipel, H.J. Mercker, R. Tschesche, K. Welmar, F. Schoenbeck, Chem. Ber. 107, 1720–1730 (1974)
- T. Kimata, M. Natsume, S. Marumo, Tetrahedron Lett. 26, 2097– 2100 (1985)
- X.D. Qin, H.J. Shao, Z.J. Dong, J.K. Liu, J. Antibiot. 61, 556–562 (2008)
- R.P. Medina, A.R. Araujo, J.M. Batista, C.L. Cardoso, C. Seidl, A.F.L. Vilela, H.V. Domingos, L.V. Costa-Lotufo, R.J. Andersen, D.H.S. Silva, Sci. Rep. 9, 12318 (2019)
- Y.F. Yuan, Y. Feng, F.X. Ren, S.B. Niu, X.Z. Liu, Y.S. Che, Org. Lett. 15, 6050–6053 (2013)
- F.R. Rossi, A. Garriz, M. Marina, F.M. Romero, M.E. Gonzalez, I.G. Collado, F.L. Pieckenstain, Mol. Plant Microbe Interact. 24, 888–896 (2011)
- N. Temme, B. Oeser, M. Massaroli, J. Heller, A. Simon, I.G. Collado, M. Viaud, P. Tudzynski, Mol. Plant Pathol. 13, 704–718 (2012)
- K. Krohn, J.Q. Dai, U. Florke, H.J. Aust, S. Drager, B. Schulz, J. Nat. Prod. 68, 400–405 (2005)
- C.Q. Liang, Y.M. Shi, X.Y. Li, R.H. Luo, Y.E. Kuhnert, F. Surup, V. Wiebach, S. Bernecker, M. Stadler, Phytochemistry **117**, 116– 122 (2015)
- R. Duran-Patron, R. Hernandez-Galin, L.G. Rebordinos, J.M. Cantoral, I.G. Collado, Tetrahedron 55, 2389–2400 (1999)
- 12. C. Qiao, W. Zhang, J.C. Han, C.C. Li, Org. Lett. **18**, 4932–4935 (2016)
- C. Qiao, W. Zhang, J.C. Han, W.M. Dai, C.C. Li, Tetrahedron 75, 1739–1745 (2019)

- 14. J.R. Hanson, R. Nyfeler, J. Chem. Soc., Chem. Commun. 72–73 (1976)
- I.G. Collado, A.J.M. Sanchez, J.R. Hanson, Nat. Prod. Rep. 24, 674–686 (2007)
- J. Moraga, B. Dalmais, I. Izquierdo-Bueno, J. Aleu, J.R. Hanson, R. Hernandez-Galan, M. Viaud, I.G. Collado, ACS Chem Biol. 11, 2838–2846 (2016)
- 17. R.M.K. Toghueo, Nat. Prod. Bioprospect. 9, 311–328 (2019)
- J.W. Tang, L.M. Kong, W.Y. Zu, K. Hu, X.N. Li, B.C. Yan, W.G. Wang, H.D. Sun, Y. Li, P.T. Puno, Org. Lett. 21, 771–775 (2019)
- B.C. Yan, W.G. Wang, D.B. Hu, X. Sun, L.M. Kong, X.N. Li, X. Du, S.H. Luo, Y. Liu, Y. Li, H.D. Sun, J.X. Pu, Org. Lett. 18, 1108–1111 (2016)
- B.S. Chen, E.W. Li, L. Liu, M.F. Liao, Z.X. Zhu, W.Y. Zhuang, L. Bao, H.W. Liu, Planta Med. 84, 1055–1063 (2018)
- S.S. Ebada, B. Schulz, V. Wray, F. Totzke, M.H.G. Kubbutat, W.E.G. Mueller, A. Hamacher, M.U. Kassack, W. Lin, P. Proksch, Bioorg. Med. Chem. 19, 4644–4651 (2011)
- 22. T. Feng, Z.H. Li, Z.J. Dong, J. Su, Y. Li, J.K. Liu, Nat. Prod. Bioprospect. 1, 29–32 (2011)
- Y.R. Bao, G.D. Chen, H. Gao, R.R. He, Y.H. Wu, X.X. Li, D. Hu, C.X. Wang, X.Z. Liu, Y. Li, X.S. Yao, RSC Adv. 5, 46252–46259 (2015)
- Z.H. Gao, L.M. Kong, X.S. Zou, Y.M. Shi, S.Z. Shang, H.R. Luo, C.Q. Liang, X.N. Li, Y. Li, X. Du, W.L. Xiao, H.D. Sun, Nat. Prod. Bioprospect. 2, 249–254 (2013)

- 25. P. Pracht, F. Bohle, S. Grimme, Phys. Chem. Chem. Phys. 22, 7169–7192 (2020)
- M.W. Lodewyk, M.R. Siebert, D.J. Tantillo, Chem. Rev. 112, 1839–1862 (2012)
- 27. P.H. Willoughby, M.J. Jansma, T.R. Hoye, Nat. Protoc. 9, 643–660 (2014)
- 28. F. Jensen, Theor. Chem. Acc. 126, 371-382 (2010)
- 29. G. Pescitelli, T. Bruhn, Chirality 28, 466–474 (2016)
- 30. T. Lu, F. Chen, J. Comput. Chem. 33, 580–592 (2012)
- 31. M.J. Frisch, G.W. Trucks, H.B. Schlegel, G.E. Scuseria, M.A. Robb, J.R. Cheeseman, G. Scalmani, V. Barone, B. Mennucci, G.A. Petersson, H. Nakatsuji, M. Caricato, X. Li, H.P. Hratchian, A.F. Izmaylov, J. Bloino, G. Zheng, J.L. Sonnenberg, M. Hada, M. Ehara, K. Toyota, R. Fukuda, J. Hasegawa, M. Ishida, T. Nakajima, Y. Honda, O. Kitao, H. Nakai, T. Vreven, J.A. Montgomery, F. Ogliaro, M. Bearpark, J.J. Heyd, E. Brothers, K.N. Kudin, V.N. Staroverov, T. Keith, R. Kobayashi, J. Normand, K. Raghavachari, A. Rendell, J.C. Burant, S.S. Iyengar, J. Tomasi, M. Cossi, N. Rega, J.M. Millam, M. Klene, J.E. Knox, J.B. Cross, V. Bakken, C. Adamo, J. Jaramillo, R. Gomperts, R.E. Stratmann, O. Yazyev, A.J. Austin, R. Cammi, C. Pomelli, J.W. Ochterski, R.L. Martin, K. Morokuma, V.G. Zakrzewski, G.A. Voth, P. Salvador, J.J. Dannenberg, S. Dapprich, A.D. Daniels, O. Farkas, J.B. Foresman, J.V. Ortiz, J. Cioslowski, D.J. Fox, Gaussian 09, Revision E.01 (Gaussian Inc, Wallingford, CT, 2010)