REVIEW



Enzyme Recovery from Biological Wastewater Treatment

Ziyi Liu¹ · Stephen R. Smith¹

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Abstract

Enzymes are high value industrial bio-catalysts with extensive applications in a wide range of manufacturing and processing sectors, including the agricultural, food and household care industries. The catalytic efficiency of enzymes can be several orders higher compared to inorganic chemical catalysts under mild conditions. However, the nutrient medium necessary for biomass culture represents a significant cost to industrial enzyme production. Activated sludge (AS) is a waste product of biological wastewater treatment and consists of microbial biomass that degrades organic matter by producing substantial quantities of hydrolytic enzymes. Therefore, enzyme recovery from AS offers an alternative, potentially viable approach to industrial enzyme production. Enzyme extraction from disrupted AS flocs is technically feasible and has been demonstrated at experimental-scale. A critical review of disruption techniques identified sonication as potentially the most effective and suitable method for enzyme extraction, which can be scaled up and is a familiar technology to the water industry. The yields of different enzymes are influenced by wastewater treatment conditions, and particularly the composition, and can also be controlled by feeding sludge with specific target substrates. Nevertheless, hydrolytic enzymes can be effectively extracted directly from waste AS without specific modifications to standard wastewater treatment processes. Purification, concentration and stabilisation/immobilisation techniques can also greatly expand the industrial application and increase the economic value and marketability of enzyme products extracted from AS. Concentrated and purified AS enzymes could readily substitute inorganic and/or commercial bioenzyme catalysts in many industrial applications including, for example, leather processing, and in detergent and animal feed formulation. Enzyme extraction from AS therefore offers significant economic benefits to the Water Industry by recovering valuable resources from wastewater. They can also optimise important waste treatment processes, such as the anaerobic digestion (AD) of sewage sludge, increasing biogas and renewable energy production. The enzyme-extracted sludge exhibits improved treatment properties, such as increased settleability, dewaterability, and anaerobic digestibility for biogas production, assisting sludge management by wastewater treatment plants (WWTPs) and enabling the further utilisation of the residual sludge.

Stephen R. Smith s.r.smith@imperial.ac.uk

¹ Department of Civil and Environmental Engineering, Imperial College London, South Kensington Campus, London SW7 2AZ, UK

Graphic Abstract



Bio-energy production

 $\label{eq:keywords} \begin{array}{l} \mbox{Activated sludge} \cdot \mbox{Enzymes} \cdot \mbox{EPS} \cdot \mbox{Cell disruption} \cdot \mbox{Hydrolytic} \cdot \mbox{Resource recovery} \cdot \mbox{Sonication} \cdot \mbox{Wastewater treatment} \\ \mbox{treatment} \end{array}$

Statement of Novelty

This review provides a critical assessment of the stateof-the-art of scientific developments and techniques for recovering high-value enzyme products from waste activated sludge (AS) produced by biological wastewater treatment processes. A systematic and critical summary of enzyme recovery systems is presented, which demonstrates the potential for full-scale industrial application. Critical areas of understanding include: enzyme distribution in sludge, the mechanisms of enzyme extraction and product formulation, and the technical feasibility of upscaling the enzyme recovery process in practice. The potential benefits and applications of recovered enzyme products are discussed, and recommendations for future research are provided.

Introduction

Enzymes are biocatalysts produced by cells and are involved in almost all metabolic processes performed by living organisms. The majority of enzymes are globular proteins comprising a tertiary amino acid conformation, which may, in some cases, be bound to a non-protein coenzyme or metal ion cofactor. The size of enzymes can range from less than 100 to over 2500 amino acid residues [1]. Enzymes accelerate biochemical reactions by similar mechanisms to inorganic chemical catalysts (e.g., metals, metal oxides and metal ions), by enabling molecules to overcome the energy barrier necessary for a reaction to proceed and increasing the correct orientation collision of molecules [2].

Enzyme catalysis initially involves the substrate binding to the active site of the enzyme to form an enzyme-substrate complex as an intermediate state. As the reaction progresses, the enzyme detaches from the products without being consumed itself [3]. The active sites of an enzyme consist of a small proportion of the total amino acid content (usually < 10 [1]) and have a unique shape and chemical properties that determine the catalytic specificity. The active functional groups also show a degree of mobility and can orientate to their counterparts within a substrate molecule in a favourable way [4]. An enzyme concentration in solution is commonly expressed in terms of its activity, which is defined as an enzyme unit (U): one enzyme unit (1 U) represents the amount of enzyme that converts 1 µmol of substrate or generates 1 µmol of products per unit time period (e.g. min or h) under standard conditions.

Enzymes offer several key advantages over inorganic catalysts [5, 6]. For instance, they can be extremely effective under mild conditions (i.e., ambient/physiological temperatures, atmospheric pressure and neutral pH) with a catalytic efficiency several orders greater compared to inorganic catalysts. Furthermore, enzymatic reactions usually induce fewer side-reactions and generate less waste by-products. Therefore, enzymes can provide an effective, environmentally sustainable alternative to inorganic chemical catalysts in a wide range of industrial processes.

The global market for industrial enzymes was approximately \$5.5 billion in 2018 and is expected to rise to \$7.0 billion by 2023, representing an annual growth rate of 4.9% [7, 8]. Hydrolytic enzymes, such as proteases, amylases and lipases, constitute the majority of the world market [9] and are used extensively in various industries including food, leather and biodiesel production [10-13], reducing dependency on environmentally-harmful chemicals (e.g. sodium sulphide in leather tanning [10]).

Commercial enzymes are typically extracted from the biomass of bacteria, or different types of fungi, including yeast, produced in an industrial fermentation process. Industrial enzyme production uses carefully formulated culture media, to provide all the essential nutrients necessary for microbial growth, followed by further consolidation and purification [14]. Enzymes are extracted by disrupting the biomass and are formulated into marketable solid or liquid products by the addition of stabilisers (neutral salts such as ammonium sulphate or low molecular weight polyols such as glycerol or sorbitol), designed to maintain the protein conformation, and standardising agents (salts or carbohydrates such as starch, maltodextrins and sugar alcohols), to dilute the extracted enzymes to a standardised activity [15]. Synthetic culture media, in particular, represent one of the most significant costs for commercial enzyme production, equivalent to 30-40% of the total manufacturing cost [16]. Therefore, high costs of production translate into a high market retail price and, consequently, enzymes produced by standard industrial methods are generally too expensive for wide adoption in large-scale and continuous industrial processes [17]. Thus, alternative, more cost-effective approaches to producing hydrolytic microbial biomass, avoiding the need for synthetic growth media, would offer significant commercial advantages for industrial enzyme production.

Wastewater treatment is essential to protect human health and the environment and involves a series of physico-chemical and biological processes to remove pollutants from the wastewater stream, producing a residual by-product, sewage sludge. The main stages of wastewater treatment include: primary sedimentation, followed by biological treatment by the activated sludge (AS) process [18]. Total sludge production from wastewater treatment within the European Union is almost 10 M t dry solids (DS) [19] and the treatment and management of sewage sludge residues is a highly topical subject [20]. Recycling to farmland as a fertiliser and soil conditioner is the main approach to sludge management adopted in the majority of countries globally, however, other significant opportunities exist to recover potentially more valuable resources from sludge [21]. Drivers to increase value recovery in wastewater and sludge treatment systems include offsetting the significant costs associated with sludge handling and management [22], which account for up to 60% of the total operating cost of wastewater treatment plant (WWTP), as well as supporting circular economy as a whole.

Recently, biological sludge generated during wastewater treatment by the AS process has been identified as potentially an alternative source of biomass for industrial enzyme extraction [23]. Within the AS process, a diverse community of prokaryotic and eukaryotic microorganisms develop in suspended floc structures (up to 2 mm diameter) and degrade non-settable organic substrates in wastewater by producing substantial quantities of hydrolytic enzymes (e.g. protease, lipase, α -glucosidase, α -amylase, cellulase, dehydrogenase and phosphatase). These enzymes specifically cleave the sensitive linkage/bonding of large organic polymer molecules to produce simpler monomer units that can transfer across the cell membrane to be metabolised by bacteria [24, 25]. The majority of hydrolytic enzymes are either adsorbed to the cell surface or embedded in extracellular polymeric substances (EPS) surrounding bacterial cells [26]. Several studies [27-29] report the extraction of crude enzymes from biological wastewater treatment by disrupting AS flocs. This represents the first step towards developing the potential commercial scale recovery of hydrolytic enzyme products from this waste biomass source.

This paper will examine the feasibility of hydrolytic enzyme production from WWTP through a critical review of extraction techniques and conditions that influence enzyme recovery efficiency and activity from AS. Approaches to purify and stabilise enzymes extracted from AS will be examined to establish possible strategies to formulate commercially viable industrial enzyme products from this biomass source. Finally, the potential markets and applications for recovered enzymes from AS will be explored.

Enzyme Location in Activated Sludge

Extracellular Enzyme Distribution Patterns Within Activated Sludge Flocs

Microorganisms assimilate organic substrates in the AS process by producing a wide range of enzymes to catalyse the hydrolysis of large organic polymer molecules into smaller units that can be transported across the microbial cell membrane to be metabolised [26]. Various extracellular enzymes (including amylase, glucosidase, phosphatase, lipase, etc.) have been detected in AS flocs (Table 1). Therefore, AS provides an excellent matrix for recovery of complex enzyme mixtures that may be formulated into compound enzyme products for industrial application.

However, enzymes are not evenly distributed within AS flocs. Hydrolytic enzymes are not usually found in the free

Table 1	Enzymes detected
in activa	ated sludge and their
distribut	tion in sludge flocs

Enzymes	Bulk aqueous fraction (%)	Microbial cells + EPS (%)	References
Phosphatase, cyclic phosphodiesterase, glycosidase and aminopeptidase	<5	> 95	[34]
α-Amylase	Negligible	Detected	[35]
Protease and α-glucosidase	~4	~96	[36]
Leucine aminopeptidase α-Glucosidase	≤ 3	97 93	[37]
Phosphatase	1	99	[38]
Leucine aminopeptidase β-Glucosidase Lipase	15.5 11.2 18.3	84.5 88.8 81.7	[39]
Protease, α -amylase and α -glucosidase	Negligible	~100	[40]

water phase of sludge flocs but are mainly detected outside of the cell (and are therefore extracellular), either attached to the cell wall or embedded within the EPS fraction of microbial cells. Extracellular polymeric substances represent a complex mixture of bacterial metabolites (including proteins, polysaccharides, lipids, humic compounds, nucleic acids, etc.) that locate outside the bacteria cell [26]. The EPS forms a negatively-charged, three-dimensional gel-like structure (Fig. 1a) due to hydrogen bond and cation bridging effects between the polysaccharide and protein components [30]. The gel-like EPS fills the void space between adjacent microbial cells, acting as an "adhesive" material that maintains the structural integrity and mechanical stability of the floc; it also provides a protective layer for the embedded microbial biomass against extreme external environmental conditions (for instance,

sudden changes in pH or the presence of toxic substances) [31-33].

Different hydrolytic enzymes have been found within the tightly bound and loosely bound EPS (TB-EPS and LB-EPS, respectively) fractions present in AS flocs (Fig. 1b). For example, Boczar et al. [24] detected esterase, lipase and leucine aminopeptidase in the easily extracted EPS (and presumably, therefore, although not specifically assigned to, the LB-EPS fraction); other enzymes (including phosphatase, valine aminopeptidase, phosphohydrolase and some glucosyl hydrolases) were mainly tightly bound to microbial cells. Cadoret et al. [42] showed that 9–24, 4–6, 15–28 and 21–67% of the total leucine aminopeptidase, α -glucosidase, protease and α -amylase was detected in LB-EPS, respectively. Indeed, Yu et al. [40] indicated that protease (98.4%) was mainly tightly bound to the cell surface, whereas α -amylase (44–65%) and α -glucosidase (59–100%)



Fig. 1 a Chemical structure of microbial extracellular polymeric substances (EPS); b Spatial distribution of different EPS fractions and the relationship with a bacteria cell (tightly bound EPS (TB-EPS)

is the fraction that is closely associated with the cell; loosely bound EPS (LB-EPS) diffuses from TB-EPS and is located at the outer region of cellular EPS) [41]

activities were largely associated with LB-EPS, and only limited protease, α -amylase, or α -glucosidase activity was detected within the TB-EPS fraction.

The mechanism of enzyme stabilisation in the extracellular fraction of sludge flocs is not clearly understood [43]. It is possible that ionic and hydrophobic interactions contribute to the immobilisation and stabilisation of hydrolytic enzymes within the sludge matrix [44]. Wingender et al. [26] suggested that the interaction between enzymes and polysaccharides was responsible for the preservation of enzymes within EPS. Frolund et al. [45] proposed that complexation with humic compounds in EPS was also involved in enzyme stabilisation.

The majority of published literature, summarised in Table 1, indicates the glycolytic enzyme activity resides mainly in the sludge floc EPS, with relatively little or no activity in the bulk aqueous fraction. Interestingly, however Guellil et al. [46] found the opposite behaviour to be the case. Guellil et al. [46] argued that readily bio-hydrolysable substrates were depolymerised by free enzymes present in the wastewater colloidal fraction. Presumably the enzymes were released by free, non-floc forming bacteria, and/or originated and survived from hydrolysis reactions that took place during transportation of raw wastewater in-sewer. Consequently, only recalcitrant, slowly biodegradable polymers (such as cellulose and lipopolysaccharides) were identified and these were removed by physicochemical adsorption onto the sludge matrix, rather than bio-degradation, forming structural elements that contributed to sludge floc stability. Evidently, the enzyme distribution patterns in biological wastewater treatment systems are dynamic and complex. However, it would appear that the biomass in AS flocs is capable of producing glycolytic enzymes and these are present in the EPS, in response to the presence of degradable polysaccharides in settled sewage.

Influence of Substrate Incorporation

Organic substances with a particle size < 0.001 μ m (soluble) are rapidly degraded by AS microorganisms whereas materials > 1 μ m (supracolloidal and/or non-settable), which constitute approximately 50–60% of the total organic matter in settled primary effluent [47], are assimilated at a much slower rate [48, 49]. The EPS fraction of sludge flocs acts as the primary surface contacting the settled primary effluent and provides sites for physical removal followed by pre-liminary degradation of insoluble organic substrates. The extremely large surface area of EPS, and the presence of a large number of hydrophobic groups [50], means that exogenous macromolecules and particulates are effectively scavenged and adsorbed into the sludge floc matrix from the free aqueous phase of wastewater. Subsequently, the hydrolytic enzymes accumulated within the EPS matrix rapidly react

with adsorbed materials, producing smaller molecular units that can be absorbed by cells for metabolism.

The distribution of different enzymes is greatly influenced by the mobility of substrates, which is determined by the molecular weight and the extent to which they transmit through the EPS matrix [51]. Thus, substrates of higher molecular weight are less mobile, compared to smaller molecules, and accumulate within the outer EPS fraction resulting in the corresponding increase in hydrolytic enzyme activity [42]. Consequently, sludge flocs in suspended growth, biological wastewater treatment reactors, develop an efficient, diverse, stable enzyme pool, which operates synergistically and/or synchronously to degrade polymeric organic compounds without extra substrate or metabolic energy transfer for enzyme maintenance and production.

Crude Enzyme Extraction from Activated Sludge

Extraction Protocol

As most hydrolytic enzymes in sludge flocs are extracellular, and are either attached to the cell wall or embedded within the sludge EPS, crude enzyme extraction from AS can be achieved by disrupting the sludge flocs and harvesting microbial EPS. Figure 2 shows the general steps and procedures to extract enzymes from AS. Typically, AS is centrifuged to remove the bulk water fraction that contains little extracellular enzymes. The solid sediment is subjected to floc disruption (with either physical or chemical methods, alone or in combination); dilution of the sludge may be



Fig. 2 Typical stages to enzyme extraction from activated sludge

necessary prior to disruption depending on the conditions. The crude enzyme product, after solid/liquid separation, is subsequently concentrated and stabilised for application.

A wide range of techniques have been reported in the literature to extract crude enzymes from AS (Table 2), and the optimum conditions tend to be specific for different enzymes. The most frequently applied techniques, shown in Table 2, and their mechanisms of action, are critically discussed in the following sections.

Physical Disruption

Sonication

Mechanisms of Enzyme Release by Ultrasonic Treatment Sonication is a practicable technology [57] that has been applied at full-scale in the wastewater industry for waste sludge pretreatment to improve the anaerobic digestion (AD) of sewage sludge. This is achieved by disrupting the biomass to shorten the hydrolytic phase of fermentation, increase the production of biogas as a source of renewable energy and to improve the mineralization and stability of fermented sludge [57, 58]. As sonication causes cellular disruption it is also a suitable technique for extracting enzymes from AS.

The initial direct effect of sonication treatment of AS is the breakdown of sludge flocs, exposing microbial cells to the aqueous environment and increasing the vulnerability to further ultrasonic destruction. The release of extracellular and intracellular substances containing enzymes from the sludge occurs upon cellular destruction. Balasundaram and Harrison [59] suggested that enzymes are released following the stepwise breakdown of the cell structure by sonication treatment, as illustrated in Fig. 3. Cadoret et al. [42] found that the ultrasound treatment of AS significantly shifted the floc size distribution of disrupted sludge towards smaller size classes, reducing the median diameter of the flocs from 91 to 37 µm. The frequency of small flocs was, therefore, also increased; 99% of the particles had a mean diameter of $< 4 \mu m$ in the disrupted sludge, compared to 6–600 μm in untreated flocs. Zielewicz [60] indicated that disrupted microbial flocs with a particle size smaller than 100 µm were more likely to release cellular substances compared to larger flocs. Thus, reducing the floc size by sonication also aids the mechanism of enzyme extraction from AS.

The disruption of sludge flocs and cells is caused by the cavitation effect of ultrasound treatment [61, 62]. High frequency acoustic signals pass through the liquid medium and the compression and expansion cycles of the sound waves create bubbles or cavities in the liquid. Two types of cavities, stable and inertial, form depending on the pressure level and other ambient conditions [63]. Stable cavities typically oscillate non-linearly around an equilibrium size and for many

cycles of the acoustic pressure without collapse. By contrast, inertial cavities expand significantly before violently collapsing and, consequently, are much more important for cell disruption [64]. Inertial cavities can expand to many times their original size in response to the applied acoustic pressure [65]. When the negative pressure exerted exceeds the local tensile strength of the liquid, the implosive collapse of the bubbles releases a shock wave. Shock waves propagate in the surrounding medium, forming jet streams initiating hydrodynamic shear forces that disrupt cell membranes. The size of the inertial cavities at which collapse occurs is described as the linear resonance size (R_0) [64, 66] as follows:

$$R_0 = \sqrt{\frac{3\gamma P_\infty}{\rho \omega^2}}$$

where, γ is the specific heat ratio of the gas; P_{∞} is the ambient liquid pressure; ρ is the liquid density; and ω is the driving frequency of the acoustic field.

Factors Influencing the Efficiency of Enzyme Extraction by Sonication The performance of ultrasound at extracting enzymes from cellular biomass depends on several critical operational conditions, including:

Ultrasound Frequency: Yu et al. [53] sonicated AS to extract extracellular and intracellular enzymes and found that low frequency ultrasound at 20 kHz was more effective than 40 kHz, and increased the activity of α -amylase, alkaline phosphatase and acid phosphatase to approximately 20, 16 and 6 U/g volatile solids (VS) compared to approximately 17, <1 and <1 U/g VS at the higher frequency, respectively. Indeed, most studies on enzyme recovery use an ultrasound frequency in the range 20-25 kHz [29, 54, 55]. The better performance of lower frequency ultrasound may be explained because it generates larger cavitation bubbles that produce stronger shear forces when they implode in the fluid, which is more disruptive to microbial cells, compared to the smaller cavitation bubbles formed at high frequencies (e.g. > 200 kHz) [67]. Ultrasound frequencies above 30 kHz can also denature the protein structure of enzymes, mainly because of the marked increase in local pressure and temperature changes caused by the intensive collapse of small cavity bubbles [68, 69]. Furthermore, the intensity of applied ultrasound, necessary to overcome the cohesive forces of the liquid media and achieve a given degree of cavitation to create voids, is correspondingly increased at higher frequencies [70]. This is explained because the cycle of compression and expansion caused by ultrasound is shorter at higher frequencies and, therefore, molecules of the liquid cannot effectively separate to form voids. Consequently, the power requirement to initiate cavitation is much greater for highfrequency sonication [70]. Sponer [71] proposed that the

Table 2	Methods for	crude enzyme	extraction	from	activated	sludge
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Method	Optimum extraction conditions	Target enzymes	Enzyme activity (U) in crude extract	References
End-over-end rotation with PBS and FDA solution; incu- bation at room temperature	Direct detection in AS without optimisation	Esterase Ala-aminopeptidase α-Glucosidase	120 U/g VS 1060 U/g VS 40 U/g VS	[52]
		(1 U: 1 µmol product released/h)		
Stirring with CER	Dosage = 70–75 g/g VS Stirring at 600 rpm for 1 h 4 °C	Esterase Leucine aminopeptidase α -Glucosidase β -Glucosidase Lipase β -Glucuronidase Chitinase	35 U/g VS 13 U/g VS <10 U/g VS ~10 U/g VS <10 U/g VS <10 U/g VS <10 U/g VS	[45]
		(1 U: 1 µmol product released/h))	
Fractionation by Dyno Mill	Glass beads (diame- ter = 0.5 mm) Bead loading = 60% (v/v) Agitating speed of mill- ing = 3200 rpm	Protease	~ 5 U/g protein for diary sludge 40 U/g protein in municipal sludge	[27]
		(1 U: undefined)		
Stirring with CER	Dosage = 60–70 g/g VS Stirring at 900 rpm for 1 h	Lipase Protease	~210 U/g VS ~300 U/g VS	[28]
Stirring with TX100	Dosage = 0.1% for lipase, 0.5% for protease Stirring at 900 rpm for 1 h	Lipase Protease	~ 320 U/g VS ~ 4000 U/g VS	[28]
Stirring with EDTA	Dosage = 10 mM Stirring at 900 rpm for 1 h	Lipase Protease	~250 U/g VS ~1700 U/g VS	[28]
Sonication with TX100	Frequency = 15 kHz Duration = 30 min Power = ~200 W Dosage = 0.1% TX100	Lipase	335.0 U/g VS	[28]
		(1 U for lipase: 1 mmol product released/min;1 U for protease: UV absorbance increase of 0.01 at 440 nm)		
Sonication	Frequency = 20 kHz Duration = 10 min Specific power input = 552 W/g DS 4 °C	Protease α-Amylase β-Glucosidase Alkaline phosphatase Acid phosphatase	~ 3 U/g VS ~ 20 U/g VS < 1 U/g VS ~ 10 U/g VS ~ 4 U/g VS	[53]
Stirring with EDTA	Dosage = 2% Stirring at 600 rpm for 3 h $4 \degree C$	Protease α-Amylase β-Glucosidase Alkaline phosphatase Acid phosphatase	<1 U/g VS ~40 U/g VS Very low Very low ~1 U/g VS	[53]
		(1 U: 1 µmol product released/m	in)	
Sonication with TX100	Frequency = 24 kHz Power intensity = 4 W/cm ² Duration = 10 min for protease, 20 min for lipase Dosage=2%TX100	Protease Lipase	~53 U/g VS ~21 U/g VS	[29]
Stirring with CER and TX100	Dosage = 0.48 g/mL CER, 0.5% TX100 Stirring at 500 rpm for 1 h Ice-water bath	Protease Lipase	7.0 U/g VS 15.5 U/g VS	[29]
		(1 U: 1 µmol product released/m	in)	
Sonication with TX100	Frequency = 24 kHz Duration = 30 min Power intensity = 3.9 W/cm^2 Dosage = $2\% \text{ TX}100$ Ice-water bath	Protease Lipase	52 U/g VS ~ 21 U/g VS	[54]

Table 2 (continued)

Method	Optimum extraction conditions	Target enzymes	Enzyme activity (U) in crude extract	References
		(1 U: 1 µmol product released/	min)	
Sonication with TX100	Frequency = 24 kHz Duration = 30 min Power intensity = $3-7.4 \text{ W/cm}^2$ Dosage = $0.1\% \text{ TX}100$ Ice-water bath	Lipase	450 U/g VS	[43]
Stirring with CER	Dosage = 40–60 g/g VS Stirring at 1000 rpm for 1 h Ice-water bath	Lipase Protease	200 U/g VS 700 U/g VS	[43]
Stirring with TX100	Dosage = 0.1% for lipase, 1.0% for protease Stirring at 1000 rpm for 1 h Ice-water bath	Lipase Protease	400 U/g VS 3600 U/g VS	[43]
		(1 U for lipase: 1 mmol product released/min; 1U for protease: UV absorbance increase of 0.01 at 440 nm)		
Sonication	Frequency = 20 kHz Duration = 10 min Power density = 1 W/mL	Phosphatase Galactosidase Glucuronidase	~ 38 U/mg protein ~ 15 U/mg protein ~ 10 U/mg protein	[55]
		(1 U: 1 µg product released/h)		
Stirring with TX100	Dosage = 1.0% TX100 60 min (agitating speed not specified) 4 °C	Protease	5.1 U/g	[56]
		(1 U: 1 mol product released/min)		

U is the enzyme activity and is based on either the amount of product formed or substrate consumed within a specified, standard time period under defined experimental conditions; the metrics for enzyme activity units varied between the studies and are presented in brackets below each reference

VS volatile solids; DS dry solids; PBS phosphate-buffered saline; FDA fluorescein diacetate; CER cation exchange resin; TX100 Triton X100, is a non-ionic surfactant; EDTA ethylene diamine tetraacetic acid, is a chelating agent; rpm revolutions per minute, is a measure of rotational speed



Fig. 3 Representation of the three stages of cell disruption by sonication: **a** Stage 1: pores form on the outer cell wall due to the external mechanical forces exerted by cavitation, releasing some periplasmic enzymes; **b** Stage 2: chemical agents diffuse from the external solution into the cell, generating a combined disruptive effect with the

mechanical forces from sonication; and **c** Stage 3: cell membranes are destroyed by the continuous exposure to cavitation ("fatigue failure" of the cell structure) causing more severe cell disruption, releasing intracellular enzymes and substances (adapted from [59])

threshold acoustic intensity (I_t , in W/cm²) required to initiate a cavitation event is linearly related to the frequency of the ultrasound (f, in MHz), and was expressed by $I_t = 0.9965f$ - 0.80 (p-value < 0.001, for f ranging from 0.5 to 10 MHz, derived from [75]). Nguyen et al. [72] similarly showed that the threshold pressure of cavitation (P_t , in kPa) was linearly related to the ultrasound frequency (f, in kHz), with $P_t = 0.1255f + 60.25$ (p value < 0.001, for f ranging from 22 to 4800 kHz, derived from [76]).

Duration: Several studies [73–75] consistently show that effective cellular disruption and enzyme release by sonication requires a minimum treatment period > 10 min. For example, Arun and Sivashanmugam [76] found that increasing the sonication duration from 5 to 20 min significantly improved the recovery of protease, lipase and amylase from fruit waste and measured maximum enzyme activity values equivalent to approximately 43, 17 and 27 U/ ml, respectively. However, extending the sonication period to 50 min reduced the enzyme activities by 76, 41 and 81%, respectively. This can be caused by a number of factors, but, released enzymes are particularly sensitive to thermal damage from increased energy dissipation as heat from excessive energy inputs to biomass associated with prolonged sonication times (see below). Thus, uncontrolled increases in temperature by sonication treatment can be detrimental to the protein structure of most enzymes, but can be effectively avoided by: (1) applying an optimal sonication duration, (2) carrying out sonication treatment in the laboratory in an icewater bath (see Table 2) or with a coolant circulation system outside the treatment chamber for larger-scale, industrial processes, and/or (3) adopting a pulse-cycle mode of on/off operation.

Energy Level: The energy supplied for cellular disruption by sonication treatment can be expressed in several ways, shown as follows [77]:

Specific energy(kJ/kg) =
$$\frac{P_o \times t}{V \times S}$$

Energy dose(kJ/L) = $\frac{P_o \times t}{V}$
Power density(W/L) = $\frac{P_o}{V}$

Power intensity
$$(W/cm^2) = \frac{P_o}{A}$$

where, P_o is the power input (kW); *t* is the sonication duration (s); *V* is the sample volume (L); *S* represents the solids content of the sample (kg/L); and *A* is the area of the emitting surface of the sonication probe (cm²).

The energy supplied by sonication treatment may be represented by any one of these different quantitative units in published studies. However, they are not readily transferrable; therefore, it is not always possible to directly or quantitatively compare the amount of sonication energy delivered for cell disruption between different reports in the literature.

Show et al. [78] argued that the impact of cavitation bubble formation, and their subsequent implosion, on AS disruption, can be viewed as a competition between the structural strength of sludge particles and the sonication intensity. Consequently, there is a critical energy intensity level above which the mechanical forces delivered by sonication will disrupt sludge flocs. However, extremely high energy intensities can be detrimental to enzyme conformation. The sonication intensity is directly proportional to the amplitude of vibration of the ultrasonic source. Thus, high sonication amplitudes generate excessive amounts of cavitation bubbles, and the shock wave from the implosion of cavities can trigger extreme shear forces in the aqueous solution. Under these conditions, the hydrogen bonds and van der Waals' interactions of the polypeptide chains can be disrupted, which are vital to maintaining the structural integrity of enzymes [70, 79].

Indeed, Yu et al. [53] showed that ultrasound intensity was more important than the duration for controlling the efficiency of enzyme extraction. Thus, the DNA content of crude enzyme extracts, which provides an indicator of sludge floc disruption and the release of cellular substances, was small and < 7 mg/g VS at a fixed specific power input of 138 W/g DS, irrespective of the sonication duration (2-20 min). In contrast, at a fixed duration (10 min), the DNA content was significantly increased with increasing power input from 138 to 690 W/g DS, and the maximum DNA release (approximately 15 mg/g DS) was measured at 552 W/g DS. Zielewicz [60] recommended that a sonication device with a power density of 880 W/L was necessary for maximum sludge floc disintegration. Indeed, the maximum recovery of protease, equivalent to 63%, from AS disruption in the laboratory experiments of Liu and Smith [23] was obtained with an ultrasound frequency of 20 kHz, a power density of 872 W/L and 10 min duration. For continuous commercial application, Zielewicz [80] suggested that the sonication duration, which is associated with the hydraulic retention time in the disintegration chamber, should provide an energy density ranging from 5 to 25 kWh/m³ (equivalent to an energy dose of 18–90 kJ/L).

Solids Content in Sludge: Ultrasonic conditions for effective cell disruption are often specific to the characteristics of the target material [57]. The core principle of sonication treatment is the conversion of electrical energy into mechanical vibrations that generate cycles of cavitation bubble formation and collapse, producing biologically disruptive shear forces at the cellular level. Consequently, sonication devices (typically a sonication probe) have an effective functioning range over which the energy is gradually dissipated, depending on the physicochemical characteristics, and particularly the solids content, of the sample material [81].

Sonication disruption of AS for enzyme extraction is usually performed on liquid sludge and so it is necessary to consider the optimum solids content of the sludge for cell disruption. Solids concentrations above the optimum increase sludge viscosity and the absorption of ultrasound energy by particulate matter [82]. Therefore, the optimum solids content and cell disruption are related to the energy consumption efficiency of the sonication treatment [78]. Cavitation bubble formation within the sludge matrix requires sufficient liquid for vaporisation to take place and, consequently, the development and propagation of cavitation bubbles can be severely limited above the optimum solids content [78].

Optimum DS contents reported for AS disruption by sonication are in the range of 2–4%. For example, Show et al. [78] found that, at a power density of 520 W/L, significant particle size reduction of AS flocs occurred for sludge samples at < 2.9% DS, however, there was only minor disruption at higher DS contents (3.8%). Zhang et al. [83] found the optimum DS concentration for ultrasonic treatment of AS at a power density of 80 W/L was 2%. By comparison, sonication of sludge with lower solids content (0.5% DS) consumed 137% more energy, while a higher solids content (3% DS) absorbed the sound energy and reduced the disruption efficiency. Zielewicz [60] also confirmed that lower solids concentrations (ranging from 2.8 to 4.2% DS) were more susceptible to ultrasonic disruption compared to more concentrated sludge (6.3% DS).

The apparent range of optimum solids contents reported in the literature may be explained because sludge samples from different sources exhibit contrasting physicochemical properties with distinct structural strength characteristics that influence the susceptibility to cavitation. Activated sludge flocs represent a multiphase, complex matrix containing microorganisms, EPS, gas bubbles and dissolved and particulate organic and mineral matter. Abbasi et al. [84] and Pilli et al. [85] indicated that gas entrapment in sludge flocs can lower the cavitation threshold pressure. Richard [86] found that small flocs (usually with diameter $< 50 \,\mu$ m), consisting only of floc-forming bacteria without a filament backbone, that commonly occur under starvation (typically when the food to microorganism ratio (F/M) [18], is very small [87], or the sludge age is extended [88]) or chronic toxicity conditions, are particularly susceptible to disruption by external forces such as ultrasound, which is consistent with other work on the effects of floc size. Zielewicz [60] suggested that the mineral/organic matter ratio of sludge was another important factor influencing the efficiency of ultrasound disruption, due to its effect on the mechanical strength, but did not explain how they were related. The mineral/organic matter ratio of AS ranges from 0.17 to 0.67 (typical value = 0.4) [18]. As the sludge becomes more mineralized with increasing sludge age, the EPS content in AS flocs decreases, and the mineral/organic matter ratio and proportion of recalcitrant cellular, structural material increase [89]. Therefore, the efficiency of sonication treatment may decrease with increasing mineral/organic matter ratio of AS flocs due to the greater dissipation and absorbance of ultrasound energy by larger, non-disruptable inorganic and recalcitrant, structural organic fractions, compared to sludge with smaller ratios.

Ultrasound Treatment and Enzyme Activity Interestingly, ultrasound has been found to have a direct, positive effect on enzyme activity if the treatment is carried out in a controlled manner. For example, Liu et al. [90] investigated the effect of ultrasound on the activity of calcium ATPase (an enzyme attached to the plasma membrane that is responsible for regulating the intracellular Ca²⁺ concentration via expulsion of Ca²⁺ from the cytosol of cells) in callus cells of Aloe arborescens. Plasma membrane, containing Ca²⁺-ATPase, was extracted by physical chopping and homogenisation of Aloe callus cells, followed by centrifugation and chemical purification using a two-phase, aqueous polymer system. Exposing the extracted plasma membrane to a pulsed ultrasound treatment (20 kHz, 2 W, 10 s duration with a pulse cycle of 0.1 s on and 0.9 s off) increased the Ca^{2+} -ATPase activity by 26% compared to the control condition without sonication. However, increasing the ultrasound energy intensity to 10 W, delivered as a continuous wave, decreased the enzyme activity by approximately 25%.

The mechanism and reason why ultrasound treatment can increase enzyme activity is not fully understood. A possible explanation is that controlled sonication treatment may induce conformational changes to the protein structure of the enzyme, by breaking covalent bonds [91]. This may release and expose more active sites in the interior of the protein to substrate molecules in the aqueous solution, increasing the apparent enzyme activity. Capelo et al. [92] suggested that sonication may also increase the efficiency of mixing and diffusion of reaction components, resulting in a higher frequency of collision between reaction substrates and active sites of the enzyme.

Milling

Mechanisms of Cell Disruption by Bead Milling Milling treatment is an effective, alternative method of cell disruption to release cellular substances from AS [93–95]. A typical bead mill consists of a grinding chamber (vertical or horizontal) filled with small beads (e.g. glass [96] or ZrO₂ beads [97]), and a rotating shaft (fitted with several agitator disks) through its centre, imparting kinetic energy to the small beads. During the rotary movement of the cells and the beads, the energy is released by the collisions between beads and cells (Fig. 4), destroying the cell wall [98]. Suarez Garcia et al. [99] indicated that approximately 95% of the total energy released from collisions in the mill occurs in the vicinity of the agitators, where the beads are moving at the maximum speed. By contrast, the beads are almost static close to the shaft or chamber wall, where they have lower potential to collide with adjacent cells. Furthermore, lay-



Fig.4 Mechanisms of bead milling in microbial cell disruption (adapted from [100])

ers of rolling beads (Fig. 4) moving at different speeds can induce shear forces and the grinding effect also contributes to cellular disruption [100].

Factors Influencing the Efficiency of Enzyme Extraction by Milling Operational Conditions Influencing Bead Milling: The progress of cell disruption against time (*t*) by milling can be described by a first order kinetic model [101, 102], shown as follows:

Cell disruption rate = $1 - e^{-kt}$

where, the first-order rate constant, k, represents the efficiency of the milling process at cell disruption, and can be affected by bead size, loading rate and agitation speed; in particular, the first-order rate constant (k) is proportional to the square of the bead loading rate and increases with decreasing bead diameter [101, 102]. Jung et al. [27] found that the disruption rate constants (k) of diary and municipal sewage sludge, under optimum conditions, were 0.22/min and 0.20/min, respectively.

The efficacy of milling as an approach to sludge and cellular disruption largely depends on the collision frequency between agitated beads and cells. Consequently, bead size and loading rate (i.e., the ratio of the total bead volume to sample volume, v/v %) are critical metrics for optimisation. For example, Jung et al. [27] used a continuous Dyno mill and glass beads to disrupt AS to solubilise the sludge and recover cellular proteins. An optimum solubilisation ratio (i.e., the ratio of total soluble organic carbon to total carbon of the disrupted AS) of 53% was obtained with 0.5 mm diameter beads, a loading rate of 60% (v/v), and agitation speed of 3200 rpm.

The diameter of the beads used in milling has a major influence on the amount of energy required to achieve the effective disruption of AS flocs and bacterial cells. Thus, Lehne et al. [103] found the minimum, optimum specific energy value was obtained with decreasing bead diameter to 0.2 mm. However, reducing the bead size further (for example, to < 0.1 mm) limited the kinetic energy carried by the beads to a level that was insufficient to cause cellular damage at every bead-bead or bead-cell wall collision, thus, increasing the specific energy demand for effective disruption. Increasing bead size also reduced the energy and disruption efficiency, for example, beads with a diameter of 1.5 mm required a specific energy of > 10,000 kJ/kg DS to achieve 60% of the disruption obtained at 0.2 mm. This represented an energy consumption almost an order of magnitude higher compared to the smaller bead size. Rai et al. [104] similarly reported that the specific energy required for maximum cell disruption (89%) using 0.77 mm diameter beads was equivalent to 15,301 kJ/kg DS.

The speed of agitation is another important parameter influencing the performance of milling, and the first-order rate constant (k) additionally increases with agitation speed [105]. However, an optimal agitation speed is necessary from a practical point of view, since the total energy demand could become significant at higher agitation speeds compared to lower speeds. Suarez Garcia et al. [99] assessed the power consumption of a bead mill under several scenarios including running the mill empty, or with beads (0.5 mm, 65% v/v loading rate) and/or microalgae cells (concentration ranging 30–155 g/L). The power consumption over a period of 20 min showed an exponential increase, from < 50 W in all scenarios to 100-350 W as the agitation speed increased from 1500 to 5500 rpm, respectively. Similarly, the heat dissipation (mainly due to friction) also increased from 17.8 to 60.8% of the total power consumed. This was probably because the movement of the beads is more chaotic at higher agitation speeds, which forces the agitator to demand more power to keep a constant stirring speed in the chamber [99].

Cell Concentration: Postma et al. [95] investigated the effectiveness of a Dyno mill (filled with 1 mm ZrO₂ beads with 65% v/v loading rate) at disrupting microalgae cells of Chlorella vulgaris. The soluble protein yield after cell disruption was significantly influenced by the concentration of biomass fed into the unit, regardless of the agitation speed (ranging from 6 to 12 m/s). The highest yield (42%) was obtained with the smallest biomass concentration (25 g dry weight/kg) and decreased by 25% as the biomass concentration increased to 145 g dry weight/kg. This was explained by the formation of insoluble protein aggregates at the high biomass concentration due to increased local protein concentrations, shear forces and interactions with other cellular substances/components. Furthermore, microbial cell suspensions with higher concentrations often show increased viscosity, and handling such material is practically limiting due to greater biomass losses and reduced mixing properties. compared to lower biomass concentrations [95].

Microbial Cell Type: Middelberg [106] indicated that bead milling is more effective for disrupting microorganisms with larger cells (e.g. yeast and fungi) compared to smaller cell types (e.g. bacteria, which are typically 1/10 the size of yeast cells [98]), due to the enhanced collision frequency between beads and larger cells. Indeed, multiple passes of operation of bead milling are often required to provide a similar degree of disruption for bacteria cells compared with a single pass operation for yeast cells.

Furthermore, the strength of the cell wall and its resistance to shear forces/bead collisions varies between biomass materials, resulting in different disruption efficiencies and k values. For example, higher shear stress may be required to achieve the complete destruction of bacterial cell walls, which contain a rigid peptidoglycan network, compared with yeast cell walls that are mainly composed of short and branched saccharide molecules (such as glucans and mannans) [106].

Advantages and Disadvantages Bead milling has several advantages for cell disruption and recovery of cellular materials, such as a continuous mode of operation, high disruption efficiency, simple biomass loading, and applicability to various types of biomass and scales [100]. However, the vigorous friction in the chamber converts a significant portion of the available energy into heat during the milling process. This can lead to inefficient energy transfer to individual cells [100] and also may require an intensive, energy-demanding cooling system to allow the recovery of functionally fragile products (e.g. enzymes) [107]. Moreover, continuous mode operation requires extra measures, such as sieving and centrifugation to separate the beads from the disrupted sludge cells. These additional process requirements may explain why milling has not been adopted as extensively for industrial enzyme recovery compared to sonication.

Chemical Extractants

Cation Removal Agents

Cation exchange resin (CER) [108, 109], ethylene diamine tetraacetic acid (EDTA) [110, 111], and sodium tripolyphosphate (STPP) [112] have been applied as chemical agents to extract both EPS and cellular enzymes from AS. The mechanism of enzyme extraction is similar for all of these chemical agents and involves removing multivalent cations from sludge EPS to disrupt AS flocs. Cations, such as: Ca²⁺, Mg²⁺, Fe³⁺ and Al³⁺, play a significant role in maintaining the structural stability of AS flocs by forming ionic bridges and stabilising proteins (which are usually negatively charged under neutral conditions due to ionization of carboxyl groups) and also small fractions of carbohydrate and nucleic acids [113, 114]. EDTA and STPP are chelating agents and form complexes with metal ions, whereas CER is a polymer and acts as a cation exchange medium that binds

with multivalent, bridging cations, to release EPS components into solution.

Chemical extractants damage cell membranes and eventually lead to cell lysis, but are much less aggressive compared to physical methods of cell disruption. For example, Frolund et al. [45] found that 90% of the extracellular enzyme, esterase, was rapidly released within 1 h when extracting microbial EPS with CER. By contrast, <40% of the intracellular enzyme, dehydrogenase, was detected during this period; however, the majority of dehydrogenase was released by extending the treatment time up to 8 h. The differences in the patterns of detection of these extra- and intracellular enzymes suggested minimal cell lysis occurred during the initial period of CER treatment. Therefore, potential advantages of chemical extractants are that they can: (1) specifically target the recovery of extracellular enzymes, and (2) reduce cell disintegration and contamination with intracellular enzymes or other cellular components.

As may be expected, the dose rate of a chemical extractant can significantly affect the extraction efficiency. Sheng et al. [115] used EDTA to extract EPS proteins from the bacteria Rhodopseudomonas acidophila. They found that EPS extraction increased markedly (from 29.9 to 84.3 mg/g dry cells) with the EDTA dose (from 0.8 to 2.8 g/g dry cells). However, raising the dose further did not improve protein extraction, presumably because the maximum removal of divalent ions from EPS had been achieved. Similarly, Merrylin et al. [113] observed a stepwise improvement in EPS release, without cell lysis, with increasing EDTA dose up to 0.4% (equivalent to 0.71 g/g VS of sludge), which gave the maximum EPS content of 38 mg DS/L in the crude extract. Larger doses, $\geq 0.5\%$ of EDTA (equivalent to ≥ 0.89 g/g VS of sludge), had no further effect on EPS release, but increased cell lysis.

Formaldehyde

Formaldehyde (systematic name: methanal (CH₂O)) is the simplest aldehyde compound (R-CHO) and is widely used in various industries as a common precursor for producing more complex compounds and materials (e.g. polyoxymethylene and phenol formaldehyde resins) [116]. It has also been used as a chemical agent for crude enzyme and EPS extraction (in combination with, or without sodium hydroxide (NaOH)) [40]. In aqueous solution, formaldehyde binds to macromolecules (e.g. proteins in EPS) by reacting with various functional groups [117]. Formaldehyde also reacts with amino, hydroxyl, carbonyl and sulfhydryl groups of proteins and nucleic acids of the cell membrane, forming cross-linked complexes. The addition of NaOH increases the pH of the mixture, dissociating the acidic groups in EPS and increasing the solubility of EPS in water. Comte et al. [118] found that a formaldehyde-NaOH process (involving

the incubation of microbial cells from AS with 36.5% formaldehyde at 4 °C for 1 h, followed by addition of 1 M NaOH and incubation for a further for 3 h) was more effective than EDTA and CER at extracting EPS from AS cells. For example, the EPS yield (dry weight of EPS/sludge DS expressed as %) obtained for the different extractants: formaldehyde, EDTA and CER, was 47.0, 19.2 and 3.1%, respectively. Similar results were also reported by Alasonat and Slaveykova [119] investigating different methods of EPS extraction from the bacterium, *Sinorhizobium meliloti*. In this case, an equivalent formaldehyde-NaOH process applied to a suspension culture of the bacteria increased the protein content in extracted EPS by 1.6–1.8 times compared to an EDTA extraction technique (2% EDTA and incubation at 4 °C for 3 h).

Surfactants

Surfactants (e.g. Triton X100 [120] and Tween 20 [74]) are often used as additives to enhance the efficiency of EPS and enzyme extraction from AS cells. For example, Triton X100 (TX100) is a non-ionic surfactant that is widely used in cell lysis protocols. It can permeabilise microbial cell membranes and consequently improve the release of cellular proteins and enzymes [121]. Karn and Kumar [122] suggested the addition of TX100 weakens the hydrophobic interaction between enzymes and the sludge floc components they attach to, facilitating the release of extracellular enzymes from the cell by shearing forces induced by simple continuous stirring.

However, TX100 can have both positive and negative effects on extraction performance, depending on the concentration used and type of enzyme. For example, Nabarlatz et al. [29] combined TX100 addition with sonication treatment to recover protease and lipase from AS. They found that, under the equivalent sonication treatment (8 W/cm^2 power intensity, 20 kHz frequency and 10 min duration), protease extraction was significantly improved with increasing TX100 concentration, achieving the maximum enzyme activity (52.9 U/g VS) at 2% v/v TX100. However, lipase extraction followed a different pattern and, in this case, the maximum activity (11 U/g VS) was obtained with 0.5-1% v/v TX100 and increasing the TX100 concentration to > 1%v/v reduced the enzyme activity. Other researchers (e.g. [28, 43]) also report the greater sensitivity of lipase to TX100 concentration compared to protease.

Bio-surfactants offer a novel alternative to standard chemical surfactants for enzyme recovery from AS. For example, Sethupathy and Sivashanmugam [123] used rhamnolipid (a bio-surfactant produced by the bacterial strain, *Pseudomonas pachastrellae*) in combination with sonication to extract a crude compound enzyme product from AS. The optimum extraction conditions were obtained with 2% v/v of rhamnolipid and ultrasound treatment at 100 W for 15 min, providing a specific energy intensity of 30,456 kJ/ kg DS. The maximum activity of protease, α -amylase, cellulase, lipase and α -glucosidase in the crude product was 42, 52, 34, 24, and 11 U/g VS, respectively. This was comparable with the extraction efficiencies of hydrolytic enzymes obtained with the standard chemical surfactant, TX100, at 1% v/v combined with sonication treatment (75 W ultrasound energy, 10 min duration and 27,027 kJ/kg DS specific energy) [112].

Factors Influencing Enzyme Extraction from Activated Sludge

Enzyme Location in Sludge Flocs

As extracellular enzymes are located in different EPS fractions of AS flocs, the degree of floc and cell disruption required to harvest them depends on whether they are associated with LB-EPS or TB-EPS, or whether they are attached to cell walls. Thus, enzymes located in the EPS fraction (so called "exo-enzymes") can be readily extracted by harvesting the EPS. In contrast, enzymes that are closely attached to the cell surface ("ecto-enzymes") may remain after removing EPS from the cell [42] and, therefore, require more physically or chemically aggressive methods for their effective extraction. Cadoret et al. [42] suggested that, in general, the distribution of enzymes between the exo- and ecto- pools was approximately in the proportions: 5-44% and 56-95%, respectively. Different exo-enzymes are distributed between the LB-EPS or TB-EPS fractions and, consequently, also vary in extractability [124]. Thus, LB-EPS can be readily separated by shearing detachment from sludge flocs from exposure to turbulence within the extraction solution, whereas TB-EPS is less affected by physical agitation because it exhibits stronger hydrophobic properties.

Yu et al. [40] investigated the effectiveness of different methods at extracting crude hydrolytic enzymes from AS. The maximum amount of protease (an ecto-enzyme) obtained by formaldehyde extraction (AS was incubated with 36.5% formaldehyde at 4 °C for 1 h) from both LB-EPS and TB-EPS was 2.3 U/g VS. However, more than 60% of the total protease was still closely associated with the cell surface after EPS removal. In contrast, up to 84 and 79% of the exo-enzymes: α -amylase and α -glucosidase, were extracted, respectively, by a formaldehyde-NaOH process (incubation with 36.5% formaldehyde for 1 h, followed by addition of 1 M NaOH and incubation for a further 3 h) and sonication (120 W, 40 kHz, 2 min). Gessesse et al. [28] also found that extracting protease was more difficult than lipase (another example of an exo-enzyme present in LB-EPS [24]). They observed the maximum protease activity in the crude extract after extraction treatment for 1 h (60-70 g/g VS of CER addition, in the presence of 0.1% TX100). In contrast, maximum lipase activity occurred after 10 min under equivalent conditions.

Upstream Wastewater Management

Microorganisms react dynamically to changes in environmental conditions within the aquatic ecosystem of the biological wastewater treatment process, including: pH, temperature, organic substrates in the influent, the hydraulic regime and presence of toxic substances/enzyme inhibitors [125]. These conditions determine the microbial physiology and, therefore, the nature of enzyme secretion. The major factors that potentially influence the types and properties of hydrolytic enzymes in AS include:

Wastewater Composition The enzyme activity profile of AS is strongly affected by the composition of the influent wastewater. For example, Nybroe et al. [52] found that addition of readily degradable starch to wastewater increased the activity of α -glucosidase in sludge flocs (0.2–0.4 µmol/mg VS/h) compared to the control group without starch addition (<0.1 µmol/mg VS/h). Li and Chrost [126] compared the activities of four extracellular enzymes, including: leucine aminopeptidase, *β*-glucosidase, alkaline phosphatase and lipase, in communal, dairy and petroleum wastewater treatment systems. The AS from each system showed a wide range of activities for each enzyme type investigated, equivalent to: 27.8-41.5, 32.9-57.8, 31.0-47.0 and 86.0-161.2 µmol/L/h, respectively. The relative activity of each enzyme was largely related to the composition of the influent wastewater and the abundance of particular substrates. Thus, lipase activity was highest, and leucine aminopeptidase and β-glucosidase activities were lowest, for AS from petroleum wastewater, which was richer in lipids, but had reduced protein and carbohydrate contents, compared to the other wastewater types. Yu et al. [127] investigated the activity of extracellular enzymes in AS samples collected from 14 WWTPs. For all WWTPs, activities of α -amylase and α -glucosidase were significantly correlated with the polysaccharide concentration in the wastewater (Pearson's correlation coefficients were 0.79 and 0.76, respectively). Similarly, protease activity was significantly correlated with the protein concentration for WWTPs treating wastewater with a high-protein content (Pearson's correlation coefficient = 0.79).

Furthermore, anthropogenic chemical compounds (e.g. organic micropollutants) in wastewater can lead to the expression of specific enzyme types (so-called "inducible enzymes") by microbial cells in AS. For example, Liu et al. [128] found that the presence of anilines, a group of environmental pollutants associated with the manufacture of dye materials and herbicides, induced the expression of aniline

dioxygenase (1.094 U/mg protein) and catechol 2,3-dioxygenase (5.224 U/mg protein) by the bacteria, *Delftia* sp., isolated from municipal AS. Similarly, *Sphingomonas* sp. Y2, isolated by Bai et al. [129] from sewage sludge, produced inducible enzymes and degraded 99.2% of the surfactant, nonylphenol polyethoxylate, which is an environmentally active anthropogenic compound with oestrogenic and mutagenic properties.

Inducible enzymes can also be produced by the addition of secondary materials to AS. Hao and Jahng [130] found that a biodrying process was accelerated by adding spent coffee grounds as a bulking agent to dewatered AS, which increased the water removal rate to 81% in 8 days, compared to 65% removal in the control treatment without coffee grounds. The improved drying rate was not only explained by the better structural conditions, which increased microbial biodrying activity, but also because the abundance of mannan (a plant storage polysaccharide that constitutes more than 50% of total hemicellulose) in coffee grounds induced the significant production of mannanase. Indeed, the specific activity of mannanase was undetectable in the control, but increased to approximately 3100 µg/min/g DS by the 2nd day of biodrying with coffee grounds. The large amount of induced mannanase interacted synergistically with protease, amylase and cellulase, to accelerate the aerobic biodegradation of protein and lignocellulose, thus increasing heat generation and the associated biodrying rate.

These various examples illustrate the potential flexibility and capacity of AS, and biological waste treatment systems in general, to produce a wide variety of commercially important enzymes. The research suggests the addition of particular organic chemical substrates, or secondary materials, such as industrial organic wastes rich in certain substances, to AS under controlled conditions as "stimulating agents" would offer a practically feasible and viable approach to induce the production of an array of high value, specific, novel enzyme products.

Sludge Age Sludge age represents the average residence time of active microorganisms in a biological wastewater treatment reactor; it is one of the main parameters influencing the operation of the AS process and has a significant impact on the treatment performance of a WWTP [131]. A sludge age of 3-6 days is typical for WWTP where only removal of carbonaceous matter in wastewater is necessary. However, most WWTP must also remove ammonia (NH₃), which is an extremely harmful pollutant of natural water systems, and this is achieved concomitantly with organic carbon mineralisation in the AS process through biological oxidation by nitrifying bacteria. Nitrifiers are relative slow growing chemoautotrophic bacteria, therefore, a longer sludge age, typically up to 18 days, is necessary, depending on the ambient temperature and influent composition,

to main nitrification activity [18]. Sludge age can also influence the activity of hydrolytic enzymes in AS. For example, sludge hydrolase activity increased by a factor of two when the sludge age was extended from 6 to 14 days [132]. This may be explained because, at longer sludge ages, dead microbial biomass may be utilised as an energy source by the remaining viable cells. Consequently, the increased availability of organic substrates, mainly in the form of rigid cellular components such as gelatin, stimulated the production of hydrolytic enzymes in the viable microbial fraction.

Activated Sludge Reactor Type In general, biological wastewater treatment processes can be classified into suspended growth systems (such as the conventional AS process) and attached growth systems (which include trickling filters, moving bed biofilm reactors, membrane biofilm reactors, etc.). In conventional AS, sludge flocs, which are agglomerations of microorganisms that are flocculated together and embedded in EPS, are suspended by physical agitation within the treatment process. By contrast, attached growth systems develop a fixed biofilm of microorganisms that is attached to a solid support medium. Therefore, attached growth systems typically have longer sludge retention times and a higher diversity of microorganisms than conventional AS processes, which also enables them to perform more efficiently at higher organic loading rate (OLR) [133]. This behaviour also has implications for enzyme production since increasing OLR stimulates the expression of extracellular hydrolytic enzymes in microbial cells. For example, Hassard et al. [134] found the activity of amino-peptidase, α -glucosidase and phosphatase increased by 4.6, 13.5 and 6.3 times, respectively, in a rotating biofilm reactor (a semisubmerged attached growth reactor with a rotating support material of high porosity mesh plates) compared to a conventional AS system. However, excess OLR can limit enzyme expression, because diffusion of compounds of large molecular weight and electron acceptors may decline under high OLR conditions [134].

Effect of Sampling Location on Enzyme Extraction from Activated Sludge

In the AS process, settled sewage from primary wastewater sedimentation is combined with returned AS (and is referred to as mixed liquor) to treat non-settleable, soluble and colloidal pollutants. Microbial activity increases and is typically ten times greater at the front of the plug-flow AS process in response to the availability of readily degradable substrates in the influent wastewater compared to the discharge point, where only biologically recalcitrant materials are present [18]. Indeed, Liu and Smith [23] found that dehydrogenase activity (an intracellular oxidoreductase that is a general indicator of microbial activity) increased from the inlet to a maximum value (5 U/g VS) at a central position in the aeration tank and subsequently decreased by approximately 35% at the discharge end of the process. In contrast, the activities of extracellular hydrolytic enzymes in AS extracts sampled at different positions along the aeration tank, and from the thickening belt after secondary sedimentation of the AS (this fraction is usually diverted to a sludge treatment process, such as AD), were relatively consistent. The maximum activities of protease, amylase and cellulase measured in crude AS extracts were: 8.2, 52.2 and 9.9 U/g VS, respectively [23]. Thus, hydrolytic enzymes involved in bacterial carbon catabolism are active not only when organic substrates are in ample supply at the inlet to the AS process, but also when they are exhausted [135]. This behaviour may be related to an ecological survival strategy adopted by floc forming bacteria to nutrient depleted conditions and to the stability of hydrolytic enzymes in the EPS fraction. Maintaining a high density of stable hydrolytic enzymes in floc EPS allows the rapid hydrolysis and assimilation of new substrates immediately they become available without the need to divert resources to, or for a time delay in, enzyme synthesis. The practical consequence of this behaviour for commercial enzyme extraction from AS is that optimum yields of hydrolytic enzymes can be obtained from the waste AS stream collected following standard operating practice. without any special intervention or alternative management of the biological wastewater treatment process.

Enzyme Purification and Stabilisation

Purification

Crude enzyme extracts from AS inevitably contain a large amount of water, and their stability in storage and suitability for industrial application is relatively limited. The presence of inhibitory substances, co-extracted from the sludge biomass, can reduce the activity of crude enzyme mixtures, however, whilst purification can increase stability and enzymic specificity, it can also cause performance losses to a certain extent compared to the crude enzyme extract. Nevertheless, the benefits of consolidating and purifying enzymes to upgrade crude AS extracts



Fig. 5 Stepwise procedure for purification of enzymes

will further facilitate the development of marketable, high value commercial products from this secondary biomass source.

Purification of enzymes is achieved following a multistep series of processes (Fig. 5). The first step is clarification of the crude extract to remove remaining cell debris. The sedimentation rate of a bacterial cell with a diameter of 0.5 mm is less than 1 mm/h [15]. Therefore, rapid separation can be achieved by centrifugation and filtration, to separate soluble proteins (including enzymes) from organelle-sequestered proteins and retain them in the supernatant or filtrate [136].

The total volume of the clarified enzyme solution is generally very large, but the concentration of enzymes is relatively low. However, the enzyme concentration in the solution can be consolidated by chemical precipitation. Enzymes are complex protein molecules possessing both ionic and hydrophobic groups and are readily susceptible to precipitation by a number of different reagents including: (1) neutral salts (followed by a dialysis step for salt removal), e.g. ammonium sulphate, which change the electrostatic forces responsible for the solubility by acting on the water molecules surrounding the protein; (2) organic solvents, e.g. acetone, ethanol and butanol, which reduce the dielectric constant of the water solution, causing aggregation of proteins; and (3) polymers, e.g. polyethylenimines and polyethylene glycols, which precipitate proteins by inducing steric exclusion effects [137]. Alternatively, consolidation can be achieved by ultrafiltration (e.g. [138]), by separating large enzyme molecules from smaller solvent molecules through a semipermeable membrane (pore size ranging from 0.008 to $0.2 \mu m$ [18]).

Purification treatment of crude AS extracts generally leads to a diminution of enzyme activity (on an equivalent volumetric base) to some extent for several reasons [139]. Jung et al. [140] observed a reduction in protease activity, equivalent to approximately 30%, following precipitation of a crude enzyme extract from disrupted AS with 40-50% saturated ammonium sulphate, compared to the original extract, and explained that this was probably due to the loss of protein content (approximately 36.2% protein) during chemical precipitation. Nabarlatz et al. [54] purified lipase obtained from AS by precipitation with 40% saturated ammonium sulphate and dialysis (12-14 kDa), and obtained an activity recovery rate for each step equivalent to 74% and 45%, respectively. Ni et al. [56] extracted a crude enzyme product from AS using 1% TX100 and sonication treatment, followed by precipitation in 80% acetone and vacuum drying at 40 °C, and found that 47, 45, 34, and 32% of the activity in the original crude enzyme extract of collagenase, lipase, amylase, and cellulase, respectively, was preserved in the final dried enzyme product.

Higher degrees of purification can be obtained by applying chromatographic techniques after the basic clarification and consolidation stages [141], and this is usually necessary for analytical or medical applications. Enzyme molecules can be selectively separated chromatographically based on their size and shape, total charge, hydrophobic groups present on the surface, and binding capacity with the stationary phase [142]. For example, Erat et al. [143] purified glutathione reductase extracted from chicken liver and precipitated with ammonium sulphate, using Sepharose affinity chromatography, and Sephadex gel filtration chromatography. The enzyme was purified 1714-fold relative to the original crude enzyme solution, and the specific activity of the final enzyme product was 120 U/mg protein. Ion-exchange chromatography for enzyme purification is also frequently reported [144-146].

In summary, the purification/consolidation steps necessary to develop commercial products from AS may reduce the enzyme activity by 30–70% compared to the original crude extract. However, this is the case with all bioenzyme recovery/purification systems from microbial biomass [15] and, in practice, only small quantities of enzymes are used to catalyse process reactions, so it is unlikely that this is a significant barrier to the development of commercially viable, industrial enzyme products from AS produced by biological wastewater treatment. Indeed, consolidation and purification would increase the flexibility and range of industrial enduses and marketability of AS enzyme products.

Stabilisation/Immobilisation

Free enzymes often show poor storage and operational stability, they are readily inactivated and are difficult to recycle and reuse, which limits their large scale commercial application [147]. Immobilisation of free enzymes is a promising approach to overcome these drawbacks, by enhancing mechanical strength, increasing resistance to denaturation and facilitating recycling of the enzyme catalyst within a reaction system [148, 149].

Conventional enzyme immobilisation is by chemical or physical binding to an inert carrier (e.g. sephorose, zeolites, silica, agarose, alginate, polyacrylamide, hollow fibres and acrylic resins) by adsorption, entrapment or encapsulation [150]. However, the carrier introduces a large amount of non-catalytic material, typically representing 90–99% of the total mass of the enzyme product [151], which significantly dilutes the enzyme activity. Recently, carrier-free immobilisation, by intermolecular cross-linking between adjacent enzyme molecules (shown in Fig. 6), has emerged as an alternative method to conventional carrier-bound enzyme immobilisation [152–154]. This technique combines purification and stabilisation into a single step, and can be performed directly on crude enzyme extracts, such



Fig. 6 Enzyme immobilisation by cross-linking: \mathbf{a} free enzyme in solution, \mathbf{b} chemical precipitation of enzyme, \mathbf{c} cross-linking of the precipitated enzyme (represented by lines)

as those obtained from AS [155]. Furthermore, immobilising enzymes by cross-linking to the surface of a membrane or micro-reactor for use within a continuous reaction process [156] offers significant potential opportunities and is an exciting area for future research and development. For instance, Barber et al. [157] recently proposed that oxidoreductase enzymes cross-linked with flexible spacers (e.g. polyethylene glycol) could provide an effective approach to organic micropollutant degradation in municipal wastewater treatment.

Potential Applications of Recovered Enzymes from Activated Sludge

Wastewater Treatment

Jung et al. [140] extracted compound enzymes from laboratory-cultured AS by milling disruption of sludge flocs and ammonium sulphate precipitation, followed by re-suspension. The compound enzyme product (containing protease at 5.4 U/mL) was mixed with an artificial milk wastewater at a volume ratio of 1:1, and rapidly hydrolysed the protein content within 2 h, releasing the degradation product, tyrosine (a main constituent amino acid of milk protein), at a yield of 103 μ g/mg protein. The results therefore demonstrated the potential application of AS enzymes for the treatment of waste streams high in protein and lipid content, such as diary wastewater.

Damasceno et al. [158] dosed crude, solid, lipase-rich compound enzymes, obtained from the solid-state fermentation of agro-industrial wastes by *Penicillium restrictum*, into a wastewater reactor treating an artificial, high-fat dairy wastewater (with 1200 mg oil and grease per L) and found the removals of both chemical oxygen demand (COD) and turbidity were improved. Dosing the enzyme at a rate of 0.1% w/v increased the overall COD removal from 83% (in the control) to 90%, and the effluent turbidity decreased by 50% compared to the control, without enzyme dosing. Mackul'ak et al. [159] applied an enzymatic pretreatment unit (using a commercial mixture of various redox enzymes, transferases and hydrolases, with an enzyme activity equivalent to 4881 U/mL, dosed at 2 mg/L, and a treatment duration of 60 min) to an industrial paintshop wastewater with a high content of organic nitrogen (COD = 5100 mg/L, total nitrogen = 184 mg/L), followed by a 24 h lab-scale AS process. The COD and total nitrogen content of the treated wastewater decreased to 200 and 20 mg/L, respectively, compared to 840 and 45 mg/L, respectively, for the control without enzymatic pretreatment.

Mobarak-Qamsari et al. [160] obtained a crude enzyme extract from *Pseudomonas aeruginosa* KM110 (a bacterium strain isolated from wastewater collected from an oil processing plant), with lipase activity = 0.3 U/mL. The enzyme extract was applied as a hydrolysis pretreatment to synthetic dairy wastewater (dose rate = 10% v/v and incubation at 45 °C for 48 h), followed by anaerobic wastewater treatment (37 °C, retention time = 13 days). Enzymatic prehydrolysis significantly increased the COD removal and biogas output by approximately 1.5 and 2 times, to 90% and 4710 mL/L wastewater, respectively, compared to the control.

Pre-treatment of Waste for Anaerobic Digestion

Anaerobic digestion is the preferred treatment method for a wide variety of municipal, industrial and agricultural organic waste materials, and particularly those with high moisture contents, as it is effective at stabilising organic matter, reducing odour nuisance and potential human pathogens, and produces a biogas rich in methane that is a valuable renewable energy source [21, 161]. However, the initial hydrolysis step of AD is often rate-limiting to the treatment of complex organic substrates, such as sewage sludge [162, 163]. Enzymatic pretreatment can enhance anaerobic digestibility of sewage sludge by increasing lysis of microbial cells and the release of protein and carbohydrate substrates from sludge flocs, and also by accelerating the biodegradation of recalcitrant materials, such as humic and fulvic acid-like substances [164].

Arun and Sivashanmugam [165] used a crude compound enzyme extract from the fermentation of fruit peel waste (pineapple and orange) to solubilise the polymeric organic compounds in waste AS. The sludge was incubated with enzymes at 35 °C and pH 7 for 60 h and the VS reduction and COD solubilisation following enzymatic treatment were approximately 22% and 25%, respectively, compared to 5% and 6%, respectively, for the control treatment (without enzyme addition).

Yin et al. [166] applied a fungal mash (rich in hydrolytic enzymes) to pretreat AS, mixed with food waste, prior to AD. The pretreatment produced a soluble COD equivalent to 7.65 g/L within 24 h, which was approximately 1.7 times

larger compared to the control, and the biomethane yield from AD of pretreated sludge increased 2.5 times compared to AS without pretreatment.

Bonilla et al. [167] pretreated pulp and paper biosludge with protease extracted from *Bacillus licheniformis* (ratio of protein/biosludge DS, 1:100) for AD and measured the effects on the biogas yield. The pretreatment enhanced the anaerobic digestibility of the sludge, increasing the specific biogas yield (measured in a 62-day biomethane potential test) by up to 26% to approximately 160 mL/g COD compared to the control supplied with inactivated enzyme, which produced approximately 130 mL/g COD of biogas.

Bahreini et al. [168] dosed a commercial enzyme product (containing cellulase and xylanase, with 20% active protein content) into primary sludge at a rate of 1% of the DS in the feed, prior to fermentation at 35 °C for 2 days. The specific soluble COD and VFA yields during the fermentation process were 316 mg/g VS and 201 mg COD/g VS, respectively, for the control group, and increased with enzymatic treatment by 68% and 35% to 532 mg/g VS and 272 mg COD/g VS, respectively.

Organic Micropollutant Degradation

Compound enzymes extracted from AS can also significantly improve the biodegradation of organic micropollutants in municipal wastewater treatment. For example, crude, cell-free compound enzymes from AS degraded the pharmaceutical compounds: acetaminophen (an analgesic), N-acetyl-sulfamethoxazole (an antibiotic), atenolol (a beta blocker) and bezafibrate (a lipid-lowering agent), with average removal efficiencies of 85, 59, 26 and 20%, respectively, compared to the control with heat deactivated enzymes, which showed no removal [55]. The majority of enzymes involved in the degradation of organic pollutants are hydrolytic (e.g. phosphatase, galactosidase and glucuronidase), and these are abundant in AS, but, other, minor nonhydrolase enzymes, such as oxidoreductase, have a strong and critical influence on pollutant destruction and may also be extracted, or inducible, in AS [55] (also see Sect. 3.4.2).

Zhou et al. [169] investigated the effects of enzyme addition on the removal of four pharmaceutically active compounds during sewage sludge AD. The pharmaceutical compounds included: clofibric acid (a lipid regulating agent), triclosan (a broad-spectrum antibacterial agent), diclofenac (an anti-inflammatory painkiller) and carbamazepine (an epilepsy drug) and each compound was added to raw sewage sludge at a concentration of 5 μ g/L. Different enzymes were mixed with sludge, including: papain (a proteolytic enzyme extracted from the papaya plant), lysozymes or cellulase, at dose rates of 30 mg/g DS, prior to mesophilic AD (35 ± 2 °C and solids retention time = 15 days). The effects of enzyme dosing on the removal rates of the target pharmaceuticals varied depending on the type of enzyme and the specific compound type. In general, triclosan and carbamazepine were the most susceptible to enzymatic degradation, and cellulase was the least effective enzyme at organic micropollutant removal. Dosing with papain and lysozymes increased the removal of triclosan to 60 and 82%, respectively, compared to 50% for the control treatment, and carbamazepine removal increased from 51% in the control to 64 and 58% by papain and cellulase addition, respectively. However, enzymatic treatment had no positive effect on the concentrations of clofibric acid or diclofenac and the largest removals of these compounds obtained with lysozyme dosing were 53% and 58%, respectively, compared to 61% and 60%, respectively, in the control condition). Sonication treatment of the sludge (20 kHz, power density at 0.05 W/mL for 30 min) significantly improved contaminant removal, by increasing the accessibility and biodegradation of organic pollutants by AD.

Therefore, Zhou et al. [170] proposed an integrated AD system, comprising of both hydrolytic enzyme dosing and pretreatment with ultrasound. Lysozymes and papain, which gave the best overall removal performance in the previous study, were mixed at a mass ratio of 1:1 and added to sludge at a rate of 30 mg/g DS prior to sonication and mesophilic AD (solids retention time = 15 days). In this case, the micropollutants were dosed into the sludge at a higher rate equivalent to 7.5 µg/L. Enzyme dosing also showed variable effects in the integrated system, however, a comparison with sonication without enzyme addition was not possible because this treatment was not included in the experimental design. Nonetheless, comparison with the earlier results [169] suggested the degradation of clofibric acid and carbamazepine was increased to 77 and 68%, respectively, with combined enzyme dosing and sonication, compared to ultrasound alone (68 and 63%, respectively [169]). On the same basis, there was little or no change in the removal of triclosan in the integrated system (69%), compared to sonication alone (73% [169]), and diclofenac degradation declined with enzyme dosing to 47% compared to 72% with ultrasound alone [169]. Increasing the digestion temperature of the integrated system to the thermophilic range (55 \pm 2 °C), had no effect on diclofenac degradation (48%) compared to mesophilic conditions (47%), but removals of the other compounds decreased (71, 65 and 51% for clofibric acid, triclosan and carbamazepine, respectively) relative to mesophilic temperatures [170]. Thus, enzyme dosing had variable effects on the degradability of pharmaceutical compounds during AD of sewage sludge and this may be explained because: (1) enzymes demonstrate a high degree of specificity at cleaving chemical bonds of available substrates [171], (2) free enzymes may be vulnerable to and be inactivated by the AD environment, and (3) ultrasound exposure may damage the enzyme structure at thermophilic temperatures, causing the breakage of essential bonds at active sites (e.g. the disulfide bonds of lysozyme [172]).

Thus, specific enzymes may be required to degrade particular, unique contaminant compounds, and general hydrolytic enzymes, similar to those supplied by Zhou et al. [169, 170], may not be effective at removing all types of organic pollutant. Thus, the general hydrolytic enzymes extracted from AS will have variable effects on degrading a broad range of organic contaminants, but specific inducible enzymes can also be expressed to target the degradation of particular micropollutant compounds (Sect. 3.4.2). The sensitivity of free enzymes to treatment process conditions can also be reduced to enhance the longevity of activity by preparing stabilised/immobilised formulations (Sect. 4.2), Furthermore, the operational conditions (including temperature and reaction time) need to be optimised to facilitate the enzymatic hydrolysis treatment of organic micropollutants.

Biofuel Production

Selvakumar and Sivashanmugam [173] produced lipase from the anaerobic fermentation of organic waste, including: pomegranate, orange and pineapple fruit peels, followed by ammonium sulphate precipitation of the crude enzyme solution. The partially purified lipase (56.5 U/mL) was used as the catalyst for biodiesel production from yeast lipid. Under optimum conditions for lipase-catalysed transesterification (6.4 methanol/oil molar ratio, 20% enzyme concentration, 35 °C and 16 h treatment time), 97% of the original lipid source was transformed into biodiesel.

Other Applications

In addition to the cases mentioned above, recovered compound enzymes from secondary materials, such as AS, have potential applications in many other areas, for example:

Leather Industry

Abu Yazid et al. [174] produced an alkaline protease from the fermentation of hair waste, followed by partial purification of the crude enzyme solution using ultrafiltration and lyophilisation techniques. The effectiveness of the lyophilised enzymes at dehairing high pigmentation cowhides was examined by incubating sections of the cowhide with the enzyme for 24 h, supplied at dose rates of 7556–9373 U/cm². The performance of enzymatic dehairing was similar to a standard chemical treatment (22 h soaking with sodium carbonate and a non-ionic surfactant for 22 h, followed by reaction with calcium hydroxide for 1 h and sodium hydrosulphide for 30 min) with a relative effectiveness of 90–95% compared to the chemical method. This study emphasised the significant potential of using enzymatic treatments as alternatives to chemical dehairing in the leather industry.

Detergent Additives

Purified proteases, amylases, lipase and cellulase are used as biodegradable, non-toxic additives to detergents to improve the effectiveness of stain removal from fabrics, without damaging the textile quality [175, 176]. A major advantage of using enzymes in bulk detergent formulations is that they do not have harmful residues and, therefore, the environmental loading of synthetic chemicals from large scale detergent use can be reduced [176].

Zhang et al. [177] investigated the washing performance of several phosphate and non-phosphate detergents that were reformulated by adding alkali protease (280 U/mg) and/or amylase (300 U/mg). They found that the efficiency of stain removal was marginally increased with the addition of mixed enzymes compared to single enzymes, probably due to a synergistic reaction between protease and amylase. For example, the stain removal from a polyester/cotton fabric soiled with blood, milk and ink was 28% for protease addition (0.5% w/w) alone and this increased to 32% for the combined enzymatic treatment (protease:amylase = 1:1, 0.5% w/w dosage for each enzyme).

Some microorganisms in sludge are naturally able to produce enzymes that can remain active under the harsh conditions that are usually applied during detergent use (such as high temperature, alkaline pH and high salt concentration [178]). For example, El Hadj-Ali et al. [179] found that a thermostable, alkaline, serine-protease, produced by a bacterium, isolated from fishery sewage sludge (identified as Bacillus licheniformis NH1), could maintain 93% of its initial activity after incubation at 40 °C for 60 min in the presence of commercial laundry detergents (7 mg/mL). This suggested that the protease derived from secondary waste biomass may be suitable as an effective enzyme additive in detergent manufacture. Stabilised enzymes may also be suitable for use in detergent formulation. For example, Soleimani et al. [180] formulated a detergent powder containing 0.1% of α -amylase, either as free enzyme or in immobilised form on silica nanoparticles, and examined the effect on the removal of starch stains from cotton fabrics, compared to the same formulation, but without enzyme addition. The presence of immobilised or free α -amylase increased the cleaning efficiency (measured as the difference in the reflectance light intensity (%) between washed and unwashed fabric, at a wavelength of 460 nm) by approximately 2.3 and 1.5 times, respectively, compared to the control.

Animal Feed

Increasing the digestibility and nutritional value of animal feeds (e.g. for pig and poultry production) is an area of expanding interest for the utilisation and application of compound enzymes [8]. The digestive systems of pigs and poultry tend to lack certain important enzymes that are necessary to break down complex organic substrates (such as barley, rye and oats) [181]. The use of endolytic enzymes as supplementary ingredients in animal feed formulations can effectively degrade cell wall polysaccharides in the feed and reduce the viscosity of the intestinal contents, resulting in improved digestion, absorption and nutritive value of feed stuffs [182, 183].

Kalmendal and Tauson [184] investigated the effect of mixed xylanase and protease (alone and/or in combination) supplied to broiler chickens fed a wheat-soybean mealbased diet. The digestibility coefficients of starch and fat were enhanced with enzyme addition, from 93.1 to 97.3% and 89.3 to 92.5%, respectively, and the retention of crude protein was also improved from 59.7 to 64.3%. Woyengo et al. [185] fed pigs with a corn-soybean based diet, supplemented with a compound enzyme product (with 4000 U of xylanase, 150 U of β -glucanase and 3000 U of protease per kg of diet). The digestibility of "acid detergent fibre" and "neutral detergent fibre" (which are indicators of structural carbohydrate components such as the cell wall fraction of the feed) was improved by 21.1 and 42.0%, respectively, compared to the control group without enzyme supplementation. The feed conversion ratio of the pigs (measured as gained body mass to feed mass ratio, G:F) also increased by 6.64% for the enzyme treated diet. Similar results were reported by Zhang et al. [186] using a commercial multi-enzyme product (containing amylase, protease and xylanase) as a dietary supplementation for a corn-soybean based feed supplied to 35-65 day old piglets. It was found that the nutrient digestibility was improved and the G:F ratio increased from 0.59 (in the control group) to 0.65 by the addition of the multi-enzyme product to the diet at a rate of 350 mg/kg. Although the numerical increases in feed digestibility with enzyme addition reported in the literature appear relatively modest, they are considered to provide significant benefits in terms of the uptake and utilisation of dietary nutrients by animals [187].

The use of AS-recovered enzymes as animal feed additives has not yet been advocated, but is a technically feasible application for compound enzyme products derived from secondary resources. The potential adoption in feed manufacture would require the development of highly refined compound enzymes and would be subject to animal feed quality regulations to ensure livestock safety and to also protect the human food chain [188].

Secondary Benefits of Enzyme Recovery from Activated Sludge

Physical disruption of AS for enzyme recovery has several important secondary benefits for the subsequent treatment, management and utilisation of the residual sludge. The large amount of EPS in AS has a major influence on the sludge properties and LB-EPS, in particular, is responsible for the high content of bound water in sludge flocs, which reduces the compressibility and dewaterability of sludge [189]. Enzyme recovery is accompanied by sludge floc disruption, and EPS removal from microbial cells reduces the bound water content in the disrupted sludge, improving the overall settlement and dewaterability characteristics [190, 191]. Physical disruption of sludge flocs also increases organic matter solubilisation and anaerobic digestibility [192]. Martin et al. [193] showed that sonication treatment of mixed, and dehydrated, primary and secondary sludge prior to AD (carried out in an ultrasonic water bath with 150 W power input for 45 min at 25 °C) almost doubled the specific biogas yield from 88 L/kg VS for the control (without pretreatment) to 172 L/kg VS. Similarly, Chatterjee et al. [194] found the methane production of septic tank sludge increased from 299 L/kg VS destroyed (specific methane yield: 224 L/kg VS) in the control to 410 L/ kg VS destroyed (specific methane yield: 337 L/kg VS) after sonication pretreatment (ultrasound frequency of 20 kHz at a power of 100 W for 5 min).

The disrupted sludge after enzyme extraction may also open other opportunities for resource recovery. For example, the AS residue contains a large fraction of cytoskeletal macromolecules and Ni et al. [56] proposed that this was suitable as a raw material for the production of superabsorbent polymers.

Conclusions and Future Outlook

Enzymes are bio-catalytic proteins that are used in a wide variety of industrial applications. An advantage of enzyme catalysed processes is that they demonstrate high reaction and substrate specificity, and, consequently, there are few if any side-reactions and little or no waste by-products. Furthermore, the reactions proceed within ambient range conditions and do not require high temperature or pressure environments. However, the extensive application of enzymes to major continuous industrial processes is limited due to their high cost, which is mainly associated with the microbial culture media. Biological AS produced by WWTP from the microbial hydrolysis and oxidation of organic substrates in municipal wastewater is a viable, potential alternative source of industrial enzymes and also offers both economic and operational advantages to the Water Industry.

The majority of hydrolytic enzymes with commercial value in AS are either embedded in the EPS fraction or adsorbed to the cell surface. Crude enzyme extraction following AS floc disruption has been demonstrated and is technically feasible in bench-scale experiments. Enzyme recovery can be achieved following different methods of extraction, by physical disruption of sludge and chemical extractant addition, alone or in combination, and successful laboratory procedures point to how this could be developed to full-scale. Sonication treatment was identified as the most favourable method for AS disruption and enzyme release, because it (1) is readily and simply adapted to continuous operation for industrial up-scaling, compared to bead milling, which requires additional processes to separate beads from the disrupted biomass, (2) can be readily optimised to maximise disruption efficiency and energy input by adjusting the duration, ultrasound frequency and solids content of the AS biomass material, and (3) can have a positive effect on the activity of extracted enzymes if operated under controlled conditions. Sonication is also a familiar technology to the Water Industry as a sludge disruption pretreatment for AD. The optimum recovery method for a specific enzyme type can be designed depending on its location within the sludge floc and whether it is extracellular within the LB-EPS or TB-EPS, or attached to the cell surface, or intracellular, as this determines the ease of separation from microbial cells. However, in practice, at full-scale, it is likely that the objective will be to apply a relatively aggressive extraction method to AS incorporating sonication treatment in combination with surfactant addition. This procedure can extract the majority of enzymes in the EPS of AS flocs, to produce a compound hydrolytic enzyme formulation that is applicable for internal use on the WWTP or that can be marketed with further consolidation and purification for a wide range of industrial applications. For example, a direct internal market can be identified within WWTP to intensify the AD of sewage sludge by the addition of compound enzymes recovered from the waste AS sludge stream, to increase both the rate and total amount of biogas and renewable energy produced from sludge treated by the AD process. Enzyme pretreatment can also increase the extent of organic micropollutant destruction achieved during sewage sludge AD. Given the significant opportunities and advantages offered by enzyme extraction from AS, the next steps, therefore, should be to set up a pilot scale process as a precursor to industrial-scale implementation.

The balance of different extractable enzymes in AS is influenced by upstream wastewater management and is affected by the composition of the wastewater influent, the sludge age in the AS process, and the type of biological wastewater treatment reactor. Furthermore, other secondary waste materials, enriched with particular, target substrate types, can be incubated with AS biomass to induce and stimulate the expression of specific, specialised enzymes that are not typically present in the normal range of hydrolytic enzymes. This potentially offers major commercial opportunities to manipulate the enzyme yield from AS to produce specific high value enzymes that may be required for particular industrial applications. By exposing the AS biomass to anthropogenic organic chemicals, for example, this approach can also be applied to produce hydrolytic or oxidoreductase enzymes that can be engineered to increase micropollutant degradation in wastewater, sludge and other environmental media.

The crude enzyme extract from AS contains a mixture of enzymes (compound enzymes) and large amounts of microbial cellular substances and water. Consequently, the crude enzyme product is not biochemically stable and also has low economic value due to the relatively large volume. Therefore, consolidation and stabilisation are necessary to broaden the range of potential applications and the shelflife of the enzyme product. Carrier-free immobilisation (e.g. via cross-linking) combines purification and stabilisation of enzymes into a single step and is likely to be a promising approach to enhance the quality, utility and marketability of enzyme products extracted and recovered from wastewater AS. Further research is therefore necessary to optimise the purification and stabilisation of compound enzyme extracts from AS to produce industrially applicable formulations and also to determine their shelflife stability. The industrial uses of recovered and purified compound enzymes from AS are potentially extensive and can include, but are not limited to, biofuel production, animal feed supplementation, leather treatment, and detergent formulation.

Integrating enzyme recovery technology into current biological wastewater treatment systems is a technically feasible proposition to utilise waste AS for industrial enzyme production. Enzyme extraction from AS offers significant benefits to the Water Industry, including: (1) the recovery of valuable resources from wastewater, providing economic benefits to the operator, (2) facilitating enzyme addition as a practicable engineering option for sludge pretreatment to improve hydrolysis and digestibility and increase the biogas and renewable energy yield from AD, (3) the destruction of organic micropollutants in wastewater and sludge to reduce chemical emissions from WWTP to the environment, (4) increasing the digestibility, settleability and dewaterability, of enzyme-extracted AS, thus improving the treatment and management of the residual solids from WWTP, (5) increasing the overall sustainability of WWTP, and (6) contributing to circular economy.

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