

## Erratum to: Melts of garnet lherzolite: experiments, models and comparison to melts of pyroxenite and carbonated lherzolite

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An unfortunate coding error led to erroneous predicted abundances of Na<sub>2</sub>O in the melting models presented in Table 5 of Grove et al. (2013). The source of this error has been discovered and corrected; the cause of the discrepancy was an error in assignment of the partition coefficient expressions used for orthopyroxene and clinopyroxene. The values presented in a corrected Table 5 are the ones obtained using the appropriate partition coefficients for Na<sub>2</sub>O. Our models of garnet lherzolite melts now have Na<sub>2</sub>O abundances that are similar to the melts predicted by Longhi (2002) and the experimentally produced melt of Walter (1998).

**Table 5** Comparison of garnet lherzolite melting model of Longhi and test of model

|                                | Primitive H&Z 10 and 1 % melts |       |      |       | Depleted H&Z 10 and 1 % melts |       |      |                    |
|--------------------------------|--------------------------------|-------|------|-------|-------------------------------|-------|------|--------------------|
|                                | 10 % <sup>a</sup>              | L '02 | 1 %  | L '02 | 10 %                          | L '02 | 1 %  | L '02              |
| SiO <sub>2</sub>               | 45.7                           | 46.2  | 46.0 | 45.6  | 45.8                          | 46.0  | 45.4 | 44.9               |
| TiO <sub>2</sub>               | 1.14                           | 0.98  | 2.09 | 1.43  | 1.04                          | 0.97  | 2.09 | 1.61               |
| Al <sub>2</sub> O <sub>3</sub> | 13.0                           | 13.0  | 14.9 | 15.1  | 12.8                          | 12.8  | 13.1 | 14.3               |
| Cr <sub>2</sub> O <sub>3</sub> | 0.18                           | 0.32  | 0.13 | 0.22  | 0.18                          | 0.34  | 0.15 | 0.29               |
| FeO                            | 10.8                           | 9.60  | 9.66 | 9.60  | 10.6                          | 9.70  | 10.6 | 10.1               |
| MgO                            | 17.4                           | 18.1  | 14.3 | 15.3  | 18.0                          | 18.5  | 16.6 | 17.2               |
| CaO                            | 9.58                           | 9.76  | 7.85 | 6.91  | 9.86                          | 10.1  | 8.48 | 8.34               |
| Na <sub>2</sub> O              | 1.96                           | 1.57  | 2.82 | 2.68  | 1.80                          | 1.32  | 2.93 | 1.96               |
| K <sub>2</sub> O               | 0.29                           | 0.28  | 2.27 | 2.13  | 0.07                          | 0.07  | 0.59 | 0.54               |
|                                |                                |       |      |       | 40.07 <sup>b</sup>            |       |      | Model <sup>c</sup> |
| SiO <sub>2</sub>               |                                |       |      |       | 45.5                          |       |      | 45.6               |
| TiO <sub>2</sub>               |                                |       |      |       | 1.27                          |       |      | 0.89               |
| Al <sub>2</sub> O <sub>3</sub> |                                |       |      |       | 10.3                          |       |      | 11.1               |
| Cr <sub>2</sub> O <sub>3</sub> |                                |       |      |       | 0.25                          |       |      | 0.16               |
| FeO                            |                                |       |      |       | 10.7                          |       |      | 10.9               |
| MgO                            |                                |       |      |       | 19.9                          |       |      | 20.0               |
| CaO                            |                                |       |      |       | 9.31                          |       |      | 9.70               |
| Na <sub>2</sub> O              |                                |       |      |       | 1.08                          |       |      | 0.94               |
| K <sub>2</sub> O               |                                |       |      |       | 0.7                           |       |      | 0.65               |
| T (°C)                         |                                |       |      |       | 1,610                         |       |      | 1,624              |

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Model calculation at 3 GPa and 10 and 1 % melting from this study and Longhi (2002)

<sup>a</sup> Melts of Hart and Zindler primitive (Hart and Zindler 1986) and depleted (Kinzler and Grove 1992) mantle compositions

<sup>b</sup> 40.07 shows the composition and temperature of a Lherzolite melt from Walter (1998) at 4 GPa and 13 wt% melting

<sup>c</sup> Forward model of melting using the Walter (1998) lherzolite composition and 13 % melting at 4 GPa. Temperature is that predicted by the model

## References

- Grove TL, Holbig ES, Barr JA, Till CB, Krawczynski MJ (2013) Mantle melting in the garnet stability field: experiments and predictive models. *Contrib Mineral Petrol* 166:887–910. doi:[10.1007/s00410-013-0899-9](https://doi.org/10.1007/s00410-013-0899-9)
- Hart SR, Zindler A (1986) In search of a bulk-Earth composition. *Chem Geol* 57(3–4):247–267
- Kinzler RJ, Grove TL (1992) Primary magmas of midocean ridge basalts 2. Applications. *J Geophys Res* 97(B5):6907–6926
- Longhi J (2002) Some phase equilibrium systematics of Iherzolite melting: I. *Geochim Geophys Geosyst* 3:art. no.-1020
- Walter MJ (1998) Melting of garnet peridotite and the origin of komatiite and depleted lithosphere. *J Petrol* 39(1):29–60